



# A sensitive biosensor for determination of pathogenic bacteria using aldehyde dehydrogenase signaling system

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## Abstract

Aldehyde dehydrogenase (ALDH) was first developed as an enzymatic signaling system of a biosensor for sensitive point-of-care detection of pathogenic bacteria. ALDH and specific aptamers to *Salmonella typhimurium* (*S. typhimurium*), as organic components, were embedded in organic-inorganic nanocomposites as a biosensor signal label, integrating the functions of signal amplification and target recognition. The biosensing mechanism is based on the fact that ALDH can catalyze rapid oxidation of acetaldehyde into acetic acid, resulting in pH change with portable pH meter readout. The altered pH exhibited a linear relationship with the logarithm of *S. typhimurium* from  $10^2$  to  $10^8$  CFU/mL and detection limit of 46 CFU/mL. Thus, the proposed biosensor has potential application in the diagnosis of pathogenic bacteria.

**Keywords** *Salmonella typhimurium* · Aldehyde dehydrogenase · Nanocomposite · pH meter

## Introduction

Point-of-care (POC) diagnostic technology with high performance is becoming increasingly popular, playing a vital role in maintaining a healthy society and food safety [1, 2]. Qualitative or quantitative detection of clinically relevant biological targets, such as small molecules, proteins, nucleic acids, and pathogens, is of crucial significance for improving the level of diagnosis, timely and effective treatment of diseases, and prevention of pathogen infection [3]. Among these approaches, biosensors with the functions of target-specific

recognition and signal transduction play an important prominent role [4]. Biosensors convert biorecognition element of binding to its target into physicochemical detection signals [5]. To facilitate the use of biosensors at POC locations, several physicochemical detectable signals, including glucose [6], pressure [7], temperature [8], hydrogen [9], and electrochemical signal [10], are used as readouts. These signal transduction strategies revamp the commercially available hand-held meters, such as personal glucose meter (PGM), pressure meter, thermometer, hydrogen detector, and screen-printed carbon electrodes [11]. Smartphone has also been used as a low-cost hand-held reader for POCT [12, 13]. When combined with biochemical analysis, these commercial hand-held devices can be employed for POC testing of a wide variety of targets. It must be noted that these biosensors require a signaling system to generate signals for hand-held meters used in POC testing. For instance, the PGM converts a substrate to glucose mainly based on enzymatic signaling system. There are several classes of enzymes, such as invertase [14], glucoamylase [15], hexokinase [16], galactosidase [17], and alkaline phosphatase [18], which have been shown to produce glucose from a variety of substrates. Numerous catalysts, such as catalase [19], platinum nanoparticles [20], and CuO/Co<sub>3</sub>O<sub>4</sub> heterojunction nanofibers [21, 22], have been explored as enzymatic signaling system to catalytically decompose substrate for the generation of gas in closed system by using a cheap and

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portable gas pressure meter as readout. Based on portable thermometer to read signal, the photo-heat conversion materials, such as graphene oxides [23], are usually used as the signal amplification platform [24, 25]. Platinum nanoparticles can efficiently catalyze the dissolution of ammonia borane complex to generate hydrogen gas with hand-held hydrogen detector meter readout [9].

In recent years, hand-held pH meters, with the advantages of low cost and simple operation, have been developed for measuring the pH of solutions [26]. Based on the pH meter as a signal reading tool, fabricated biosensors had been applied to a variety of targets, such as mycotoxin [27], proteins [28, 29], DNA sequences [30], and bacterial pathogens [31]. Enzymes, such as acetylcholine esterase [26], glucose oxidase [31], alkaline phosphatase [18], and urease [32], which could effectively catalyze their corresponding substrates to cause changes in the solution pH, have been developed as powerful tags of these fabricated biosensors. Therefore, the enzyme activity as the signal label and sensitivity of the pH meter determine the analytical performance of a biosensor.

Acetaldehyde dehydrogenase (ALDH) is an important class of enzyme which oxidizes aldehydes to their corresponding carboxylic acids [33]. As a crucial regulator of aldehyde metabolism in the human body, ALDH can effectively protect the body from active aldehydes [34]. Several acetaldehyde biosensors have been constructed for the detection of acetaldehyde in mixture solution or gas using ALDH. For instance, Avramescu et al. developed amperometric biosensor using ALDH for the determination of acetaldehyde to control wine quality [35], while Iitani et al. constructed fiber optic biosensors for the measurement of acetaldehyde in liquid phase using ALDH [36]. Furthermore, bioelectronic gas sensors incorporating ALDH have been developed for convenient analysis of acetaldehyde in expired gas [37], and a stick-type biochemical gas sensor with ALDH has been developed for convenient analysis of breath acetaldehyde [38]. However, to the best of our knowledge, ALDH-catalyzed rapid oxidation of acetaldehyde to form acetic acid has not yet been applied as a signal output in immunosensors.

A novel nanocomposite with phosphate as inorganic component and protein as organic component has attracted widespread attention in the field of biosensors owing to its large surface area, good stability, and simple preparation [39]. The nanocomposite can improve the catalytic activity and durability of the organic component (protein). Multiple proteins are embedded in nanocomposites with several integrated functions [40]. Ye et al. developed a POC biosensor based on concanavalin A-glucose oxidase- $\text{CaHPO}_4$  organic-inorganic nanocomposites as signal tags and a hand-held pH meter signal readout. Glucose oxidase-embedded nanocomposites can efficiently convert glucose to gluconic acid, which causes a decrease in the pH [31]. Ye et al. prepared concanavalin A-invertase- $\text{CaHPO}_4$  integrating both essential functions of

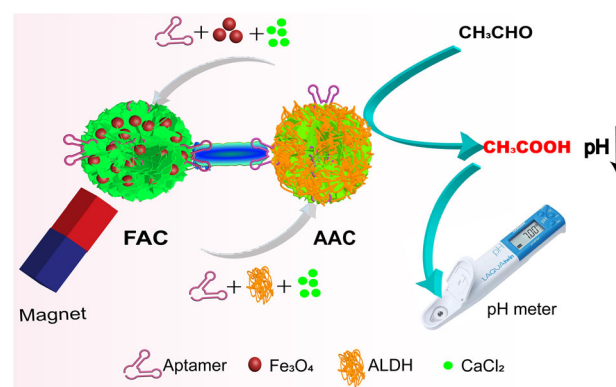
biological recognition and signal amplification for POC testing of *Escherichia coli* O157:H7 [41]. Bai et al. synthesized antimicrobial peptides-invertase- $\text{Ca}_3(\text{PO}_4)_2$  organic-inorganic nanocomposites with multifunctional target recognition and signal transduction for POC determination of pathogenic bacteria [42]. As a signal output, the invertase in the nanocomposites hydrolyzes sucrose to glucose, which can be reliably quantified on a PGM.

We first established a portable biosensor based on ALDH-aptamer- $\text{Ca}_3(\text{PO}_4)_2$  organic-inorganic hybrid nanocomposite (AAC), as a signal probe and pH meter as a readout (Scheme 1). The pathogenic bacterium, *S. typhimurium*, was used as the detection model. AAC was synthesized by a simple one-step process using ALDH and Salmonella-specific aptamer as organic materials and calcium phosphate as inorganic material. Ferroferric oxide ( $\text{Fe}_3\text{O}_4$ )-embedded aptamer- $\text{Ca}_3(\text{PO}_4)_2$  hybrid nanocomposite (FAC) was exploited as the magnetic capture probe for the isolation of *S. typhimurium*. In the presence of *S. typhimurium*, a sandwich complex was formed between capture probe FAC and signal probe AAC. The ALDH in the sandwich immunocomplex catalyzed conversion of acetaldehyde into acetic acid. The generated acetic acid caused a change in  $\text{H}^+$  concentration, thus altering the pH value. Therefore, the change in the pH reflected the concentration of *S. typhimurium*, and could be strictly quantified with a hand-held pH meter.

## Experimental section

### Chemicals and reagents

ALDH and  $\text{Fe}_3\text{O}_4$  were obtained from Sigma-Aldrich, Inc. (Saint Louis, MO, USA). The aptamer of *S. typhimurium* (CTGATGTGTGGGTAGGTGTCGTTGATTCTTC TGGTGGG) [43] was synthesized by Sangon Biotech Co., Ltd. (Shanghai, China). Acetaldehyde was obtained from Sangon Biotech Co., Ltd. (Shanghai, China). 2-Hydroxy-1-



**Scheme 1** Schematic illustration of the synthesis of FAC and AAC and formation of the sandwich structure for *S. typhimurium* detection.

ethanethiol ( $\beta$ -CH<sub>2</sub>OHCH<sub>2</sub>OSH) was purchased from Beijing Dingguo Changsheng Biotech Co., Ltd. (Beijing, China), and CaCl<sub>2</sub>•2H<sub>2</sub>O was bought from Shantou Xilong Chemical Factory, Guangdong (Shantou, China). 10 × PBS consisted of 10 mM NaH<sub>2</sub>PO<sub>4</sub> and 10 mM Na<sub>2</sub>HPO<sub>4</sub>, and 1 × PBS consisted of 1 mM NaH<sub>2</sub>PO<sub>4</sub>, 1 mM Na<sub>2</sub>HPO<sub>4</sub>, and 100 mM KCl. Acetaldehyde-based substrate reaction solution comprised 2 mM CH<sub>3</sub>CHO, 10 mM  $\beta$ -CH<sub>2</sub>OHCH<sub>2</sub>OSH, and 500 mM KCl (pH 8.5). Strains of *Staphylococcus aureus* (*S. aureus*, CICC 21600), *Listeria monocytogenes* (*L. monocytogenes*, CICC 21529), *S. typhimurium* (CICC 24119), and *Escherichia coli* O157:H7 (*E. coli* O157:H7, CICC 21530) were kindly provided by China Center of Industrial Culture Collection (Beijing, China). The pH meter (B-713) used in this work was purchased from HORIBA Co., Ltd. (Kyoto, Japan) and was employed for pH measurements during quantitative detection. All the analytical pure chemicals were not further purified, and all aqueous solutions were prepared from ultrapure water (18.3 M $\Omega$ ·cm, Milli-Q; Millipore, Billerica, MA, USA).

### Synthesis of FAC and AAC

FAC and AAC nanocomposites were prepared in a 1.5-mL Eppendorf tube, according to previous studies, with some modifications [31, 42]. For the synthesis of FAC, 20  $\mu$ L of CaCl<sub>2</sub> aqueous solution (200 mM) was added to 800  $\mu$ L of PBS (pH 7.4) containing 8  $\mu$ L of aptamer (10  $\mu$ M) and 40  $\mu$ L of Fe<sub>3</sub>O<sub>4</sub> (50 mg/mL), immediately mixed by shaking, and then incubated at 25 °C for 18 h. The FAC nanocomposites were harvested by magnetic separation, washed thrice, and resuspended in 200  $\mu$ L of 1 × PBS (pH 7.4). For the synthesis of AAC, 800  $\mu$ L of 10 × PBS (pH 8.5) were added to 50  $\mu$ L of KCl (100 mM) and 10  $\mu$ L of  $\beta$ -CH<sub>2</sub>OHCH<sub>2</sub>OSH solution (1 M) to prepare antioxidant phosphate buffer. Then, 80  $\mu$ L of ALDH (5 mg/mL) and 20  $\mu$ L of aptamer (10  $\mu$ M) were added to the phosphate buffer and mixed well. Subsequently, 20  $\mu$ L of CaCl<sub>2</sub> aqueous solution (200 mM) was added to the mixing buffer, mixed gently, and incubated at room temperature for 12 h. The AAC nanocomposites were centrifuged (10,000 rpm, 6 min), washed with 1 × PBS (pH 7.4), resuspended in 200  $\mu$ L of 1 × PBS (pH 7.4), and stored at 4 °C for subsequent use.

### Characterization of FAC and AAC

The synthesized FAC and AAC nanocomposites were characterized by scanning electron microscopy (SEM, Hitachi su8010, Tokyo, Japan), X-ray diffraction (XRD, XD6, Beijing, China), and energy-dispersive X-ray spectrometer (EDS, USA). For SEM and EDS analyses, the suspensions of FAC and AAC were dried on the conductive film and sprayed with gold coating. For XRD analysis, the FAC and AAC suspensions were dried at

37 °C, and the respective precipitates were collected and separately ground into fine powders.

### Detection of *S. typhimurium*

A total of 5  $\mu$ L of FAC and 20  $\mu$ L of AAC were simultaneously added into 50  $\mu$ L of 1 × PBS (pH 7.4) containing different concentrations of *S. typhimurium* and incubated at 37 °C for 20 min to form FAC-target bacteria-AAC sandwich immunocomplex. The sandwich immunocomplex was separated by magnet and washed twice with 50  $\mu$ L of 1 × PBS (pH 8.5), and then mixed with 100  $\mu$ L of acetaldehyde-based substrate reaction solution and incubated at 45 °C for 30 min. Subsequently, the supernatant pH was detected using a portable pH meter. The registered change in pH value ( $\Delta$ pH) as a signal was relative to the amount of *S. typhimurium*. The  $\Delta$ pH was defined as the difference between the pH signals recorded by the pH meter in the presence and absence of the target bacteria. All the data were obtained from measurements performed in triplicate.

### Determination of milk samples

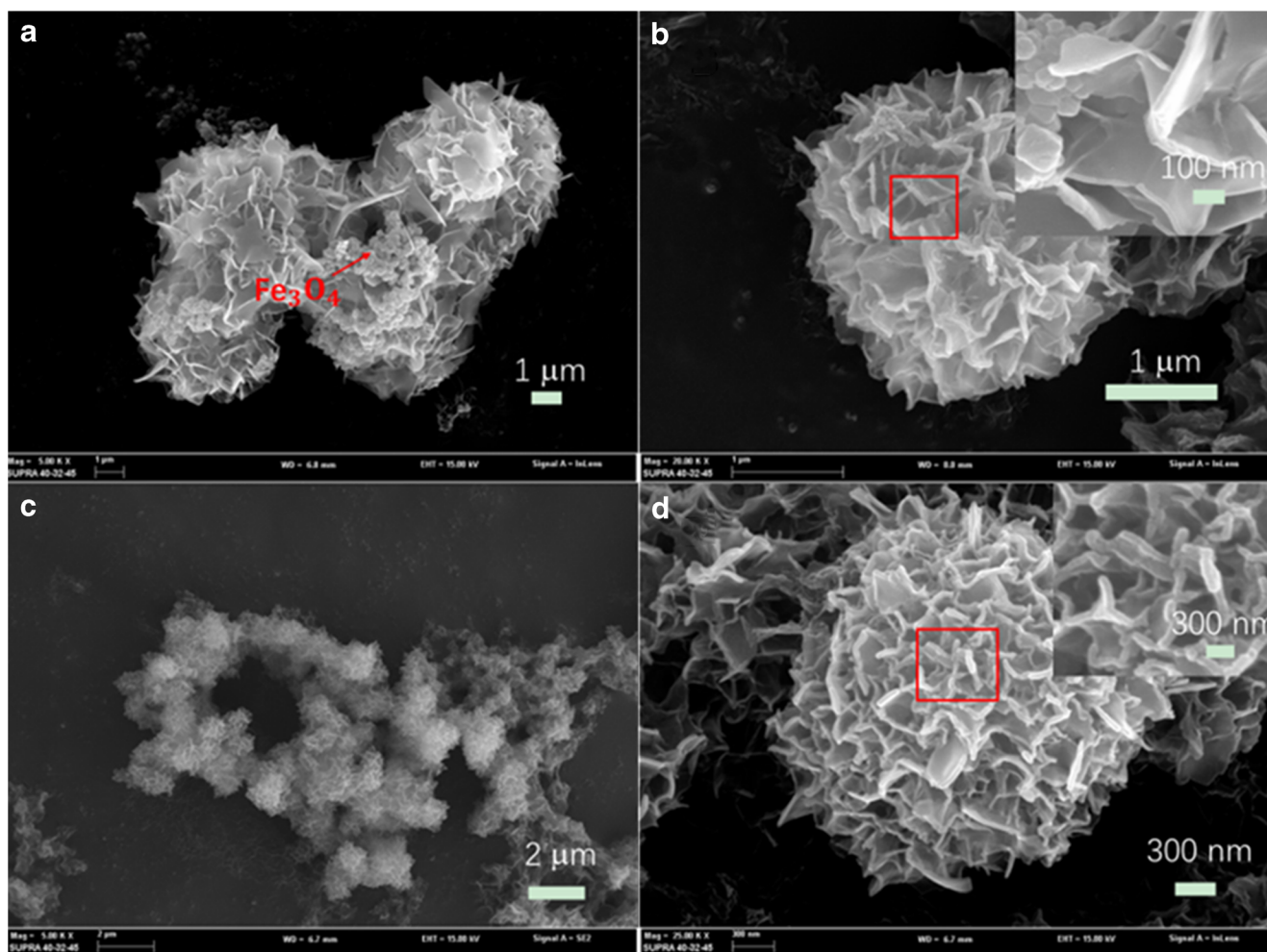
For recovery analysis, the *S. typhimurium* suspensions were added to sterilized milk samples to achieve bacterial densities of 10<sup>5</sup>, 10<sup>6</sup>, and 10<sup>7</sup> CFU/mL, respectively. All the milk samples were processed according to the abovementioned testing procedure for *S. typhimurium*. The recovery (%) of *S. typhimurium* in milk samples was analyzed based on calibration plot.

## Results and discussion

### Characterization of FAC and AAC

Low-resolution SEM images clearly revealed that the FAC (Fig. 1a) and AAC (Fig. 1c) nanocomposites were uniform porous spheres with a diameter of about 2  $\mu$ m, whereas high-resolution SEM images demonstrated interlaced peony-like flower morphology, which can effectively increase specific surface area. As shown in Fig. 1b, a large number of Fe<sub>3</sub>O<sub>4</sub> nanoparticles could be observed from the partial zoom image of FAC. Figure 1d exhibits the dense, thick, and rough petals of AAC, and it can be presumed that the amide group of ALDH and Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> formed complex, promoting cross-linking between the petals. During the process of petal growth, ALDH may have got attached to the petals through co-precipitation, making the petals thick and rough.

To investigate the elemental composition of FAC and AAC, EDS mapping was conducted (Fig. S1, see Electronic Supplementary Material (ESM)). As shown in Fig. S1 (see ESM), FAC was mainly composed of Fe, Ca, P, C, N, and



**Fig. 1** SEM images of **a, b** FAC and **c, d** AAC

O elements, with Fe and N derived from  $\text{Fe}_3\text{O}_4$  and aptamer, respectively. The EDX image showed (ESM Fig. S1b) six typical elements, including Ca, S, P, C, N, and O, in AAC, which can be attributed to  $\text{Ca}_3(\text{PO}_4)_2$ , ALDH, and aptamer components. Preliminary results showed that FAC and AAC nanocomposites were successfully synthesized from the corresponding components.

The XRD patterns (ESM Fig. S2a) of FAC fitted well with  $\text{Ca}_3(\text{PO}_4)_2$  (JCPDS 00-018-0303) and  $\text{Fe}_3\text{O}_4$  (JCPDS 01-077-1545) particles, confirming the inorganic component in the hybrid nanoflowers. These findings further confirmed that  $\text{Fe}_3\text{O}_4$  particles dispersed into FAC. The XRD patterns (ESM Fig. S2b) of AAC revealed that the relative intensities and positions of all the diffraction peaks corresponded to  $\text{Ca}_3(\text{PO}_4)_2$ , which further proved the successful synthesis of FAC and AAC with the corresponding functional groups.

### Optimization of experimental conditions

To achieve sensitive performance of the biosensor, some important experimental parameters were optimized. First, the

signal probe AAC was ensured to effectively catalyze the conversion of acetaldehyde into acetic acid, generating maximum signal value of  $\Delta\text{pH}$ . The amount of ALDH (0.1–0.5 mg) and aptamer (4–24  $\mu\text{L}$ ) embedded in AAC directly affected the signal amplification effect of the biosensor. As presented in Fig. S3a (see ESM), with the increase in the amount of ALDH, the signal value of  $\Delta\text{pH}$  gradually increased. When the amount of ALDH was 0.4 mg, the signal value of  $\Delta\text{pH}$  reached the maximum and then decreased. Excessive ALDH may hinder the formation of flower-like nanocomplexes, and produce tiny petals, leading to nonspecific adsorption and weak response signals. Therefore, 0.4 mg of ALDH was selected as the optimal dosage for synthesizing AAC. As shown in Fig. S3b (see ESM), the  $\Delta\text{pH}$  presented a function related to the volume of the aptamer used to synthesize AAC. When the volume of the aptamer was 20  $\mu\text{L}$ , maximum  $\Delta\text{pH}$  value was obtained. Therefore, 20  $\mu\text{L}$  was selected as the optimum aptamer volume for the synthesis of AAC. Nucleic acid aptamers are inherently acidic to a certain extent, and AAC synthesis requires weak alkaline conditions. When excess aptamers are added, the pH of the reaction system is

reduced, thus inhibiting the growth of AAC. Furthermore, as the amount of aptamer embedded in FAC determines the ability of FAC to capture the target bacteria, the volume of aptamers in FAC was also optimized. As shown in Fig. S3c (see ESM), with the increasing volume of the aptamer from 2 to 12  $\mu\text{L}$ , the  $\Delta\text{pH}$  value also increased until a plateau appeared at 8  $\mu\text{L}$ . Therefore, 8  $\mu\text{L}$  was selected as the optimal aptamer volume for the synthesis of FAC.

Subsequently, the incubation time of the sandwich immunocomplex formed by AAC, target bacteria, and FAC was optimized. The incubation time ranged from 20 to 60 min, and the signal value of  $\Delta\text{pH}$  was almost a plateau and began to decline after 60 min. Hence, 20 min was selected as the best incubation time for the formation of sandwich complex (ESM Fig. S3d). In addition, we optimized the catalyzing time for the conversion of acetaldehyde into acetic acid by the signal probe AAC. As shown in Fig. S3e (see ESM), the signal value of  $\Delta\text{pH}$  gradually increased from 10 to 30 min, reached a plateau from 30 to 50 min, and then decreased after 50 min. Therefore, we chose 30 min as the optimal catalyzing time. As presented in Fig. S3f (see ESM), the optimal reaction temperature was 45  $^{\circ}\text{C}$  depending on the maximal signal value of  $\Delta\text{pH}$ .

When compared with some previous similar approaches, the detection time performance of the developed biosensor is superior, with the detection method requiring only 50 min for efficient completion, which is comparatively faster than the other methods (Table 1). In the developed biosensor, FAC, *S. typhimurium*, and AAC are incubated together for 20 min, and ALDH of AAC could catalyze rapid oxidation of acetaldehyde into acetic acid in 30 min. The developed immunosensor is a simple and quick method for detecting pathogenic bacteria. In a previous study, Ye et al. [31] developed organic-inorganic nanocomposites as signal tags for

detecting *E. coli* O157:H7, combined with a pH meter or PGM to read the signals, which required 175 and 115 min for efficient detection, respectively. The 96-well plates fixed with *E. coli* O157:H7 antibodies were successively incubated with the target bacterium *E. coli* O157:H7 and signal label nanocomposites, which catalyzed the reaction of the substrate to produce corresponding signal. Thus, this method is time-consuming and relatively complex. Similarly, the immunosensor established by Bai et al. [42] is mainly based on immunomagnetic nanocomposites, signal label nanocomposites, and PGM readout. Although the entire procedure is relatively simple, the formation of sandwich structure of immunomagnetic nanocomposites, target bacteria, signal label nanocomposites, and enzymatic substrate of signal tags nanocomposites requires up to 115 min.

## Sensitivity

The  $\Delta\text{pH}$  of the proposed biosensor was assayed under optimal conditions with different concentrations of *S. typhimurium*. As shown in Fig. 2, the signal value of  $\Delta\text{pH}$  showed a linear relationship with the concentration (logarithm) of *S. typhimurium* in the range of  $10^2$ – $10^8$  CFU/mL. The linear regression equation is  $Y = 0.252X - 0.287$  ( $R^2 = 0.9856$ ), where  $Y$ ,  $X$ , and  $R^2$  represent  $\Delta\text{pH}$ , *S. typhimurium* concentration (logarithm), and correlation coefficient, respectively. The limit of detection (LOD) was calculated as 46 CFU/mL, which generated a signal of  $\geq 3$  SD above the mean of the negative control values. These results demonstrated that the established biosensor has satisfactory sensitivity, large linear range, and low LOD. The sensitivity of the developed biosensor is almost the same or better than that of some previously reported methods (Table 1). When compared with the other detection methods, the developed

**Table 1** Performance comparisons of different methods for foodborne pathogen detection

| Target Bacterium       | Methods                                                                                            | Linear range (CFU/mL)    | LOD (CFU/mL)    | Total detection time (min) | Reference |
|------------------------|----------------------------------------------------------------------------------------------------|--------------------------|-----------------|----------------------------|-----------|
| <i>E. coli</i> O157:H7 | ELISA based on nanocomposite signal tags and pH meter readout                                      | $10^1$ – $10^6$          | 10              | 175                        | [31]      |
| <i>E. coli</i> O157:H7 | ELISA based on nanocomposite signal tags and PGM readout                                           | $10^1$ – $10^5$          | 10              | 115                        | [41]      |
| <i>S. typhimurium</i>  | Sandwich immunoassay based on magnetic nanocomposites, signal tags nanocomposites, and PGM readout | $10^1$ – $10^7$          | 10              | 115                        | [42]      |
| <i>E. coli</i> O157:H7 | Sandwich immunoassay based on magnetic beads, gold nanoparticles, and electrochemical readout      | $10^2$ – $10^5$          | 148             | 75                         | [11]      |
| <i>E. coli</i> O157:H7 | Ferrocene antimicrobial peptide-modified biosensor                                                 | $10^3$ – $10^8$          | $10^3$          | 90                         | [10]      |
| <i>S. typhimurium</i>  | Immunomagnetic nanomaterials and thermal sensor readout                                            | $3 \times 10^2$ – $10^3$ | $3 \times 10^2$ | 90                         | [8]       |
| <i>S. typhimurium</i>  | Sandwich immunoassay based on magnetic nanocomposites, signal tag nanocomposites, and pH readout   | $10^2$ – $10^8$          | 46              | 50                         | This work |

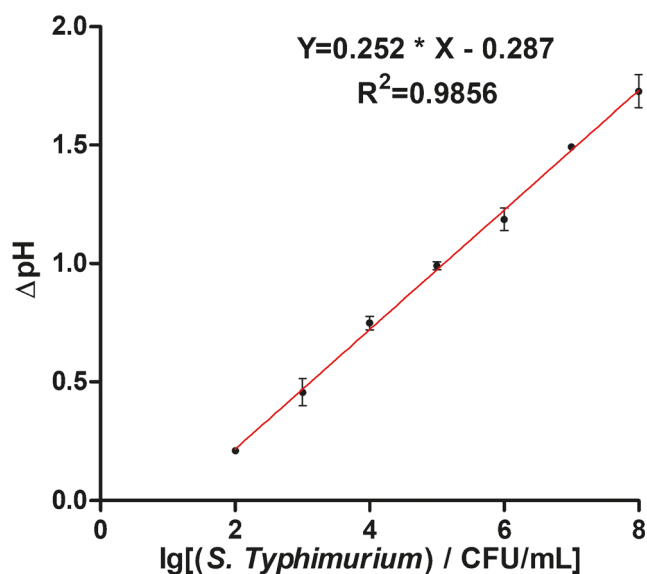


Fig. 2 Linear regression equation for *S. typhimurium* detection

method and those of Ye et al. [31, 41] and Bai et al. [42] have low LOD. The sensitivity of our developed method is 46 CFU/mL, which is at least 1–2 orders of magnitude higher than that of conventional detection methods. It must be noted that the enzyme activity and stability in the nanocomposites enhance with increasing number of enzymes embedded in the nanocomposites, which promote signal amplification efficiency of the signal tag nanocomposites. For signal reading using a glucose meter, a glucose meter strip is essential, which increases the cost of the detection method.

The deteriorated stability at the elevated temperatures and inactivity in the organic medium limit wide application of natural enzyme. The immobilization of enzyme in organic-inorganic nanocomposites provides it with a much higher stability compared with the free form of enzyme due to the unique hierarchical nanostructure. The developed biosensor has good stability. Therefore, the developed biosensor based on enzyme embedded in organic-inorganic nanocomposite has better stability than conventional technologies (e.g., enzyme-linked immunosorbent assay) using free enzyme [44]. AAC and FAC organic-inorganic nanocomposites are both much easier to prepare (one-step co-precipitation method) and do not need any complex covalent modification. The hand-held pH meter read signal at no cost. All of these make the developed biosensor more cost-effective than traditional technologies (e.g., enzyme-linked immunosorbent assay and lateral flow).

### Specificity

To assay the specificity of the developed biosensor for detecting *S. typhimurium*, several control pathogenic bacteria, such as *L. monocytogenes*, *S. aureus*, and *E. coli* O157:H7, were used to challenge the detection approach. As shown in Fig. 3,

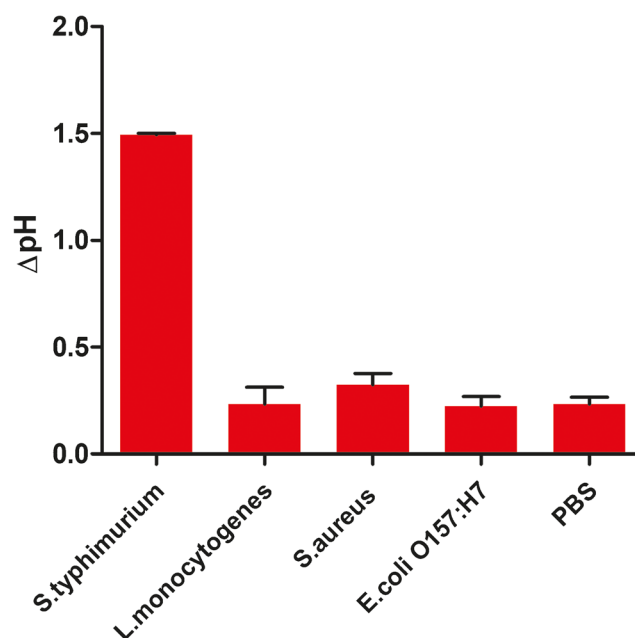


Fig. 3 Specificity assay using different pathogenic bacteria. The density of pathogenic bacteria was  $1 \times 10^7$  CFU/mL

under the same experimental conditions, the signal response of the target bacteria presented a significant change, when compared with other pathogenic bacteria. These results effectively demonstrated the good specificity of the established approach. However, the practical application of the developed biosensors remains challenging and requires more efficient strategies to simplify the experimental process.

### Detection of *S. typhimurium* in milk

To further investigate its capability to determine foodborne pathogens, the fabricated biosensor was used to detect real food samples. Different concentrations of *S. typhimurium* were respectively added into sterilized milk samples, and then recovery was assayed using the developed method. As presented in Table 2, the range of analytical recoveries was from 91.4 to 106.6%, which were within the acceptable criteria for bioanalytical method validation. The most important requirements are the detection limit (1 CFU/25 g) for *S. typhimurium* in milk powder [45, 46]. These results revealed that the fabricated biosensor is an excellent tool for assaying pathogenic bacteria in milk.

### Conclusion

In summary, based on ALDH signaling system and portable pH meter readout, a novel signal sensing platform was fabricated for detecting *S. typhimurium*. To the best of our knowledge, this is the first study to employ ALDH as a signal detection platform for the determination of pathogenic bacteria.

**Table 2** Recoveries of *S. typhimurium* from spiked milk sample

| Original bacterial concentration (CFU/mL) | Added bacterial concentration (CFU/mL) | Detection (CFU/mL)               | Recovery (%) |
|-------------------------------------------|----------------------------------------|----------------------------------|--------------|
| 0                                         | 10 <sup>5</sup>                        | (1.066 ± 0.07) × 10 <sup>5</sup> | 106.6 ± 7    |
| 0                                         | 10 <sup>6</sup>                        | (0.914 ± 0.05) × 10 <sup>6</sup> | 91.4 ± 5     |
| 0                                         | 10 <sup>7</sup>                        | (1.028 ± 0.06) × 10 <sup>7</sup> | 102.8 ± 6    |

With its excellent signal amplification efficiency and portable pH meter readout, the developed method presented three advantages. First, the detection method only needs 50 min for efficient completion, which is relatively faster than the other reported methods. Second, the sensitivity of 46 CFU/mL is at least an order of magnitude or 1–2 orders of magnitude higher than that of conventional detection method, which guarantees effectiveness and feasibility of the fabricated detection method. Finally, the developed biosensor can be extended to detect other pathogens by simply replacing the corresponding aptamer. Overall, the fabricated biosensor shows strong application potential in food safety and clinical diagnosis.

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## Compliance with ethical standards

**Conflict of interest** The authors declare that they have no competing interests.

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