



Colorimetric adenosine aptasensor based on DNA cycling amplification and salt-induced aggregation of gold nanoparticles

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Abstract

An aptamer based assay is described for the colorimetric detection of adenosine. The presence of adenosine triggers the deformation of hairpin DNA oligonucleotide (HP1) containing adenosine aptamer and then hybridizes another unlabeled hairpin DNA oligonucleotide (HP2). This leads to the formation of a double strand with a blunt 3' terminal. After exonuclease III (Exo III)-assisted degradation, the guanine-rich strand (GRS) is released from HP2. Hence, the adenosine-HP1 complex is released to the solution where it can hybridize another HP2 and initiate many cycles of the digestion reaction with the assistance of Exo III. This leads to the generation of a large number of GRS strands after multiple cycles. The GRS stabilize the red AuNPs against aggregation in the presence of potassium ions. If, however, GRS forms a G-quadruplex, it loses its ability to protect gold nanoparticles (AuNPs) from salt-induced AuNP aggregation. Therefore, the color of the solution changes from red to blue which can be visually observed. This colorimetric assay has a 0.13 nM detection limit and a wide linear range that extends from 5 nM to 1 μ M.

Keywords Adenosine detection · Hairpin DNA oligonucleotide · Low detection limit · Wide linear range · Colorimetric assay · Gold nanoparticle aggregation · Exonuclease III · Color change · G-quadruplex · DNA cycling amplification

Introduction

Adenosine is involved in neuromodulation [1], cell signaling [2], and neuroprotection [2]. In the brain, adenosine regulates cerebral blood flow and neurotransmission [3], offering an effective treatment means of ischaemia, epilepsy, inflammation, as well as chronic and neuropathic pain [4–6]. Therefore, it is very vital to detect adenosine for clinical diagnosis.

Massive detection methods have been developed for the determination of adenosine, such as fluorescence [7–9], electrochemiluminescence [10, 11], capillary electrophoresis [12], chromatography [13, 14], and surface-enhanced Raman

spectroscopy [15]. However, some of the approaches have some disadvantages, such as expensive instruments, tedious sensing process, limited detection range, as well as weak selectivity ability.

To address these defects, aptamer biosensors have been developed for the highly sensitive and selective detection of adenosine [16–23]. Hereinto, gold nanoparticles (AuNPs) based colorimetric aptasensors stand out from these adenosine aptasensors due to advantages of its rapidness, simplicity, and no need to use expensive analytical instruments [24–30]. These aptasensor exhibited excellent specificity for adenosine, however, sensitivity still needs to be improved.

Here, we designed a sensitive colorimetric aptasensor based on AuNP aggregation for detecting adenosine (Scheme 1). This sensor involves two hairpin DNA oligonucleotides (HP1 and HP2), where HP1 contains adenosine aptamer, and HP2 consists of the guanine-rich strand (GRS). With adenosine, the specific binding of HP1 to adenosine causes the deformation of hairpin-structured HP1, and further triggers HP1 and HP2 to hybridize to each other, thus forming a duplex DNA structure with a blunt 3' end. Therefore, Exo III can recognize the blunt 3' terminus and digest the segment (light blue) of HP2 [31, 32]. The adenosine-HP1 complex is released intact to hybridize with another HP2. After a series of

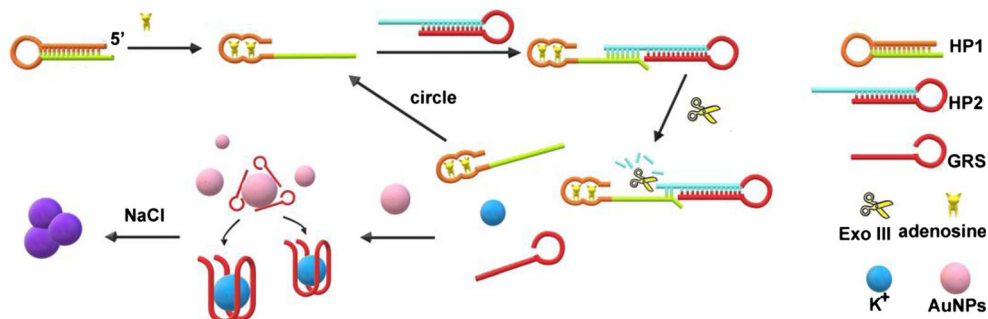
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Scheme 1 Schematic illustration of the colorimetric aptasensor for adenosine detection based on Exo III-assisted DNA cycling amplification and G-quadruplex-induced AuNP aggregation in salt solution



Exo III-assisted cycles of DNA cleavage reaction, more amounts of GRS are generated. The binding of GRS to potassium ions prompts GRS to fold into stable G-quadruplex structure, thus preventing the stabilization of individual AuNPs, resulting in purple-blue AuNP aggregates under high salt conditions. Therefore, the target adenosine can be identified by the bare eye or quantified sensitively by UV-vis spectrometry. The above significant changes were confirmed by TEM images of the AuNPs at different conditions as shown in Fig. 1. As presented, the AuNPs are well-dispersed in aqueous solutions, whereas the presence of target adenosine caused AuNP aggregates.

CAG GTA TTT TTT-3', HP2: 5'-TGC GAT GGG TTG GGC GGG ATG GGA ACC GCC CAA CCC ATC GCA ATA CCT GG-3') were synthesized by Sangon Biotech Co., Ltd. (www.sangon.com, Shanghai, China). All of the reagents were of analytical reagent grade. Ultrapure water was provided by a Direct-Q3 system and used as a solvent in all experiments.

Instrumentation

UV-vis absorption spectra were recorded by a Shimadzu UV-2550 spectrophotometer (Japan). Transmission electron microscope (TEM) was performed on a JEM-2010 transmission electron microscope at 80 kV.

Synthesis of gold nanoparticles

The information is available in Supporting Information (Synthesis of gold nanoparticles, Fig. S1).

Protocol for adenosine detection

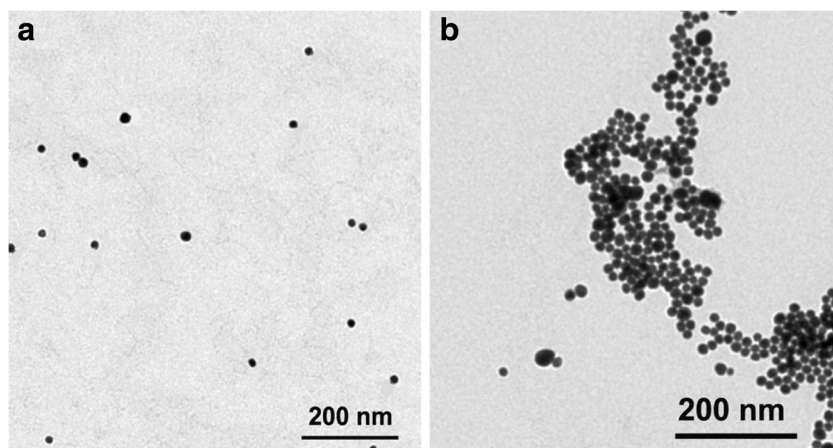
Prior to use, DNA strands were processed as follows: Hairpin HP1 (1 μ M) and HP2 (1 μ M) were respectively heated to 85 $^{\circ}$ C for 10 min, and then slowly cooled down to room

Experimental section

Materials and reagents

Trisodium citrate, chloroauric acid ($\text{HAuCl}_4 \cdot 3\text{H}_2\text{O}$), Adenosine (A), uridine (U), cytidine (C), and Exonuclease III (Exo III) were purchased from Aladdin Industrial Corporation (www.aladdin.com, Shanghai, China). The DNA sequences (HP1: 5'-AGA GAA CCT GGG GGA GTA TTG CGG AGG AAG GTC GCA ATA CTC CCC

Fig. 1 Typical TEM images of AuNP solution containing (a) 0, and (b) 100 nM adenosine. NaCl concentration: 25 mM, Exo III concentration: 0.5 U/ μ L, HP1 concentration: 40 nM, HP2 concentration: 100 nM, and pH = 7.0



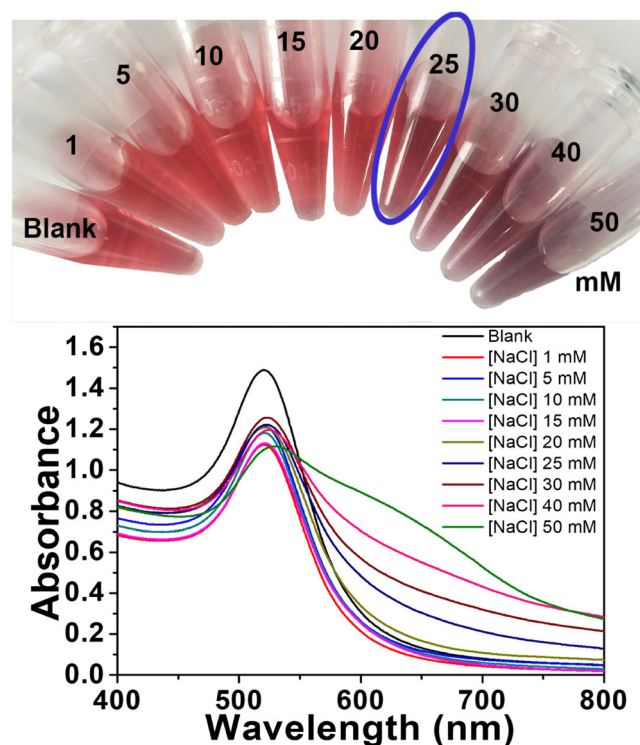


Fig. 2 The effect of different concentrations of NaCl on (a) The color, (b) UV-vis spectra of HP1-AuNP solution. Exo III concentration: 0.5 U/ μ L, HP1 concentration: 40 nM, HP2 concentration: 100 nM, and pH = 7.0

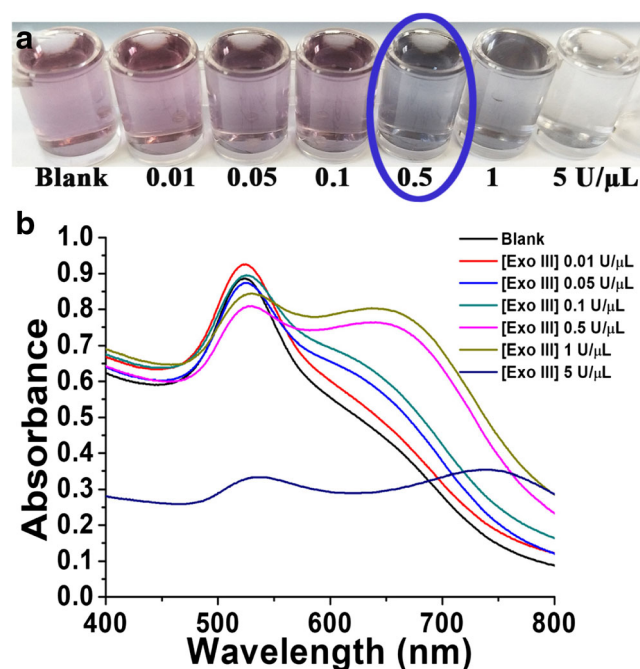


Fig. 3 The effect of different concentrations of Exo III on (a) The color, (b) UV-vis spectra of HP1-adenosine-HP2-AuNP solution. NaCl concentration: 25 mM, HP1 concentration: 40 nM, HP2 concentration: 100 nM, and pH = 7.0

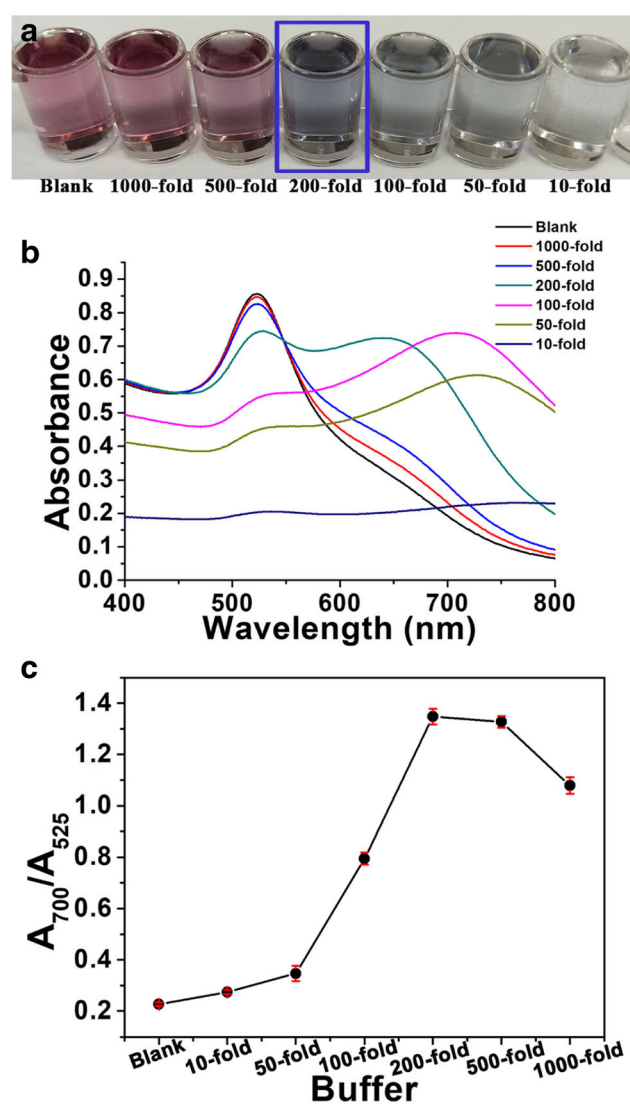


Fig. 4 The effect of different dilution multiples on (a) solution color, (b) UV-vis spectra, and (c) the A_{700}/A_{525} of HP1-adenosine-HP2-AuNP solution. The error bars represent standard deviation obtained from three independent measurements. NaCl concentration: 25 mM, Exo III concentration: 0.5 U/ μ L, HP1 concentration: 40 nM, HP2 concentration: 100 nM, and pH = 7.0

temperature to form the hairpin structures. 70 μ L of HP1 (40 nM) was mixed with adenosine at various concentrations (final concentration: 0–5 μ M) for incubation of 1 h at 37 $^{\circ}$ C. Afterwards, 70 μ L of HP2 (100 nM) was injected to the solution and mixed thoroughly, followed by the addition of 70 μ L of Exo III (0.5 U/ μ L) (pH 6.5) containing 200 mM KCl, 5 mM KPO_4 , and 0.05 mM EDTA for another 2 h incubation at 37 $^{\circ}$ C. Then, 400 μ L of AuNPs (10 nM) was added to the above solution. After incubation of 3 h at 37 $^{\circ}$ C, 18 μ L of 1 M NaCl was added and mixed. The mixture was rest at room temperature for 5 min to finish the coloration process. Finally, the solutions were used for absorbance measurement at wavelengths of 525 and 700 nm.

Results and discussion

Optimization of experimental conditions

Herein, the effect of NaCl concentration on the sensing performance was first investigated. As shown in Fig. 2, the absorbance of AuNP-HP1 solution at 525 nm gradually decreased with the increasing NaCl concentration from 0 to 50 mM. It can be found that when NaCl concentration was 25 mM, color began to change from red to blue-purple, which indicates 25 mM NaCl can induce AuNP aggregation in the presence of 40 nM HP1. Thus, 25 mM NaCl was used for further studies. Exo III concentration is an important factor due to its effect on the performance of the signal amplification. From Fig. 3, it is obvious that the absorbance of AuNPs at 520 nm was decreased with increasing Exo III concentration from 0 to 5 U/ μ L. Beyond 0.5 U/ μ L, the color of the solution became colorless, this is because that excessive Exo III was adsorbed on the surface of AuNPs, resulting in further AuNP aggregation. Hence, 0.5 U/ μ L Exo III was employed. The effect of the multiplier of the dilution of the buffer (10 mM Bis Tris Propane-HCl, 10 mM MgCl_2 , 1 mM DTT, pH 7.0) on its absorbance are shown in Fig. 4. The result shows that maximum signal at A_{700}/A_{525} was observed when the buffer was diluted 200 times. Beyond 200-fold dilution, the A_{700}/A_{525} value began to decrease on account of lower ionic strength. Thus, 200-fold diluted buffer was selected for all subsequent assays. We also investigated the absorbance change using the above buffer between pH 6 to pH 7.5 (Fig. 5). We find that when solution pH was 7.0, the A_{700}/A_{525} value reached the maximum. This indicates that the solution of pH = 7 was the most conducive to the binding of the HP1 to target adenosine. Finally, we investigated the effect of GRS concentration on the absorbance. As displayed in Fig. S2, the A_{700}/A_{525} increased with the increase of GRS

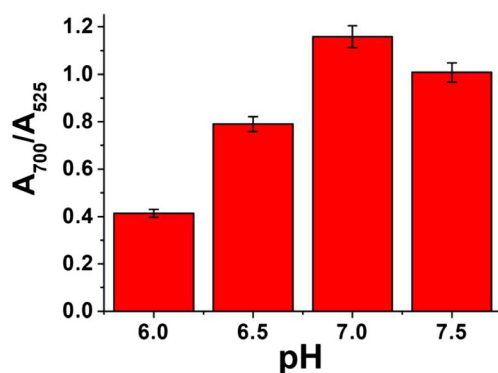


Fig. 5 The effect of solution pH on the absorbance change (A_{700}/A_{525}). NaCl concentration: 25 mM, Exo III concentration: 0.5 U/ μ L, HP1 concentration: 40 nM, and HP2 concentration: 100 nM. The error bars represent standard deviation obtained from three independent measurements

concentration from 40 to 100 nM, followed by the decrease beyond GRS concentration of 100 nM. This is because that a surplus of GRS generated more negative charges, which cause greater repulsion between negatively charged AuNPs and GRS. Thereby, the 100 nM GRS was selected for the subsequent research.

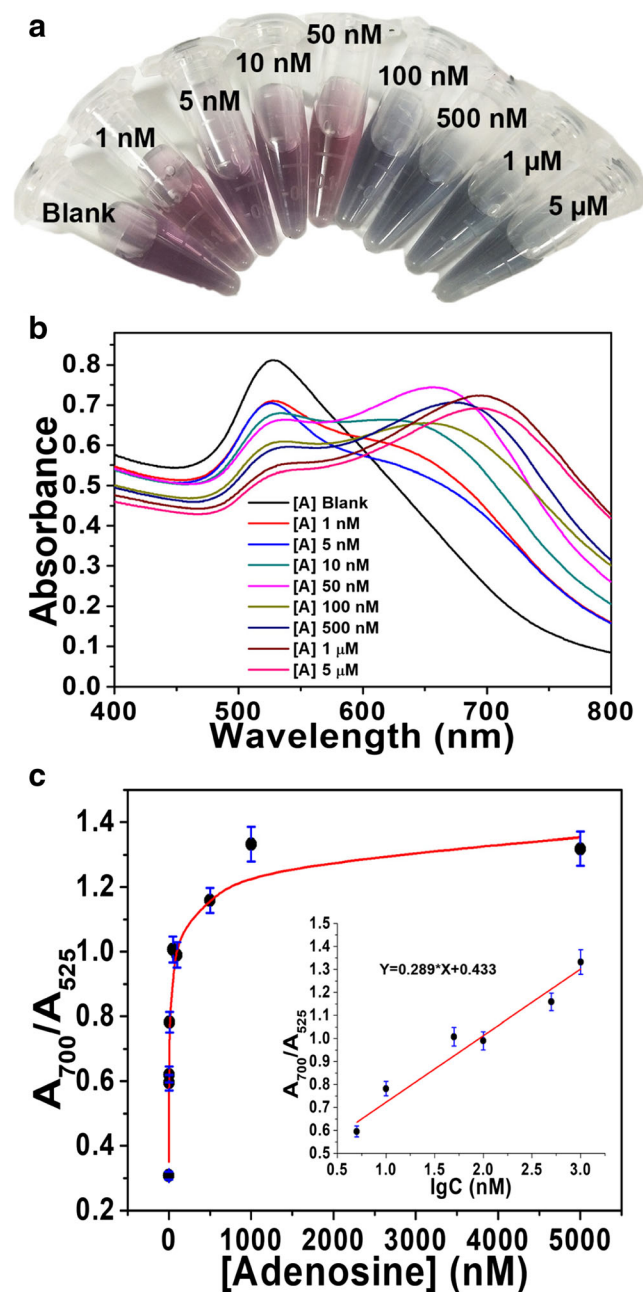


Fig. 6 **a** Photographs of the solution with different concentrations of adenosine. **b** UV-vis spectra of the solution upon the addition of various concentrations of adenosine from 0 to 5 μ M. **c** Inset: calibration graph between A_{700}/A_{525} and log adenosine concentration (5 nM–1 μ M). The error bars represent standard deviation obtained from three independent measurements

Table 1 Comparison of our method with other methods for adenosine detection

Material	Method	Detection limit (nM)	Linear range	Ref.
AuNPs, aptamer, and Exo III	Colorimetry	0.13	5 nM–1 μ M	This work
Carbon dots and aptamer	Fluorescence	0.63	2–50 nM	[33]
CdTe quantum dots	Fluorescence	0.217	0.333–167 nM	[34]
Aptamer, Exo I and a 2-aminopurine-modified DNA	Fluorescence	300	10–600 μ M	[35]
DNA/Ag nanoclusters	Fluorescence	103.4	0.5–7.0 μ M	[7]
Cu/Ag nanoclusters	Fluorescence	19	0–1 μ M	[36]
Exo III and aptamer	Colorimetry	17	50 nM–6 μ M	[37]
Endonuclease, aptamer, and magnetic beads	Electrochemistry	0.032	0.05–20 nM	[38]

Analytical performance

To assess the sensitivity of the colorimetric sensor for determination of adenosine, under the optimized experimental conditions, as shown in Fig. 6a, b, we can observe that the AuNP solution undergoes a red-to-dark blue-to-colorless color change, and the absorbance values (A_{700}/A_{525}) gradually increase with the increase of adenosine concentration from 0 to 5 μ M. However, at higher concentrations (1 μ M), the detection signals gradually level off. The absorbance intensity (A_{700}/A_{525}) shows a good linear relationship with the logarithm of adenosine concentration in the range of 5 nM–1 μ M (inset of Fig. 6c). The limit of detection (LOD) was estimated as 0.13 nM, calculated by three times signal to noise. Compared with other reported methods for adenosine detection, as shown in Table 1, the sensitivity of the assay was comparable or more sensitive.

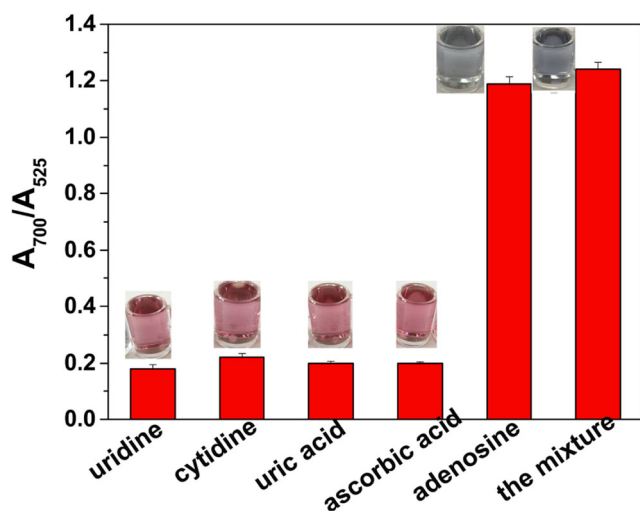


Fig. 7 Change in absorbance of the AuNP solution in the presence of adenosine and other interfering substance (uridine, cytidine, uric acid, and ascorbic acid). Inset: photograph of AuNP solutions in the presence of adenosine and its analogues. The concentrations of analogues (500 μ M) were 100-fold higher than that of adenosine (5 μ M). The error bars represent standard deviation obtained from three independent measurements

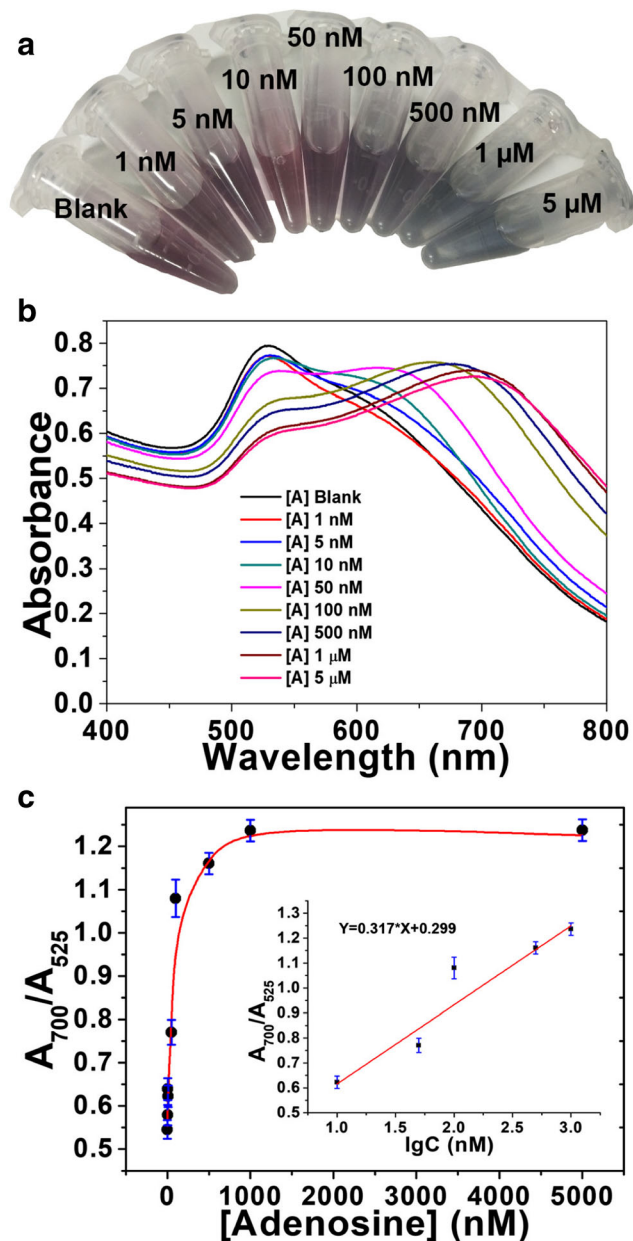


Fig. 8 **a** Photographs of AuNP solution before and after addition of different concentrations of adenosine. **b** Changes of absorption spectra after the addition of adenosine with various concentrations. **c** The different absorbance intensity (A_{700}/A_{525}) corresponding to different concentrations of adenosine. Inset: linear calibration plot for adenosine detection in urine samples. Error bars indicate standard deviation of the mean of three experiments

Table 2 The values of adenosine spiked in human urine samples and determination using the sensor ($n = 3$)

Urine samples	Adenosine spiked (nM)	Detected by this sensor (nM)	Recovery rate (%)
1	50	46.88	93.76
2	500	495.65	99.13
3	1000	947	94.70

Selectivity

To test the selectivity of this adenosine detection method, the absorbance intensity values (A_{700}/A_{525}) of two analogues of adenosine including uridine (U) and cytidine (C), uric acid, and ascorbic acid were measured, respectively. As shown in Fig. 7, the addition of 500 μM these interfering substance, which was 100-fold higher than that of adenosine, to the solutions did not produce remarkable absorbance, which was negligible compared with the absorbance intensity of adenosine. In addition, the coexistence of adenosine and these interfering substance show almost identical absorbance generated by adenosine, demonstrating that the specificity of this sensor was reasonably satisfactory for target adenosine. The good selectivity was ascribed to a high affinity of the HP1 for adenosine.

Analytical application

To demonstrate the feasibility of this sensing system in practical applications, human urine samples were collected from our laboratory staff. The need for working in the UV makes the probe prone to interferences by biomatter (in blood/serum, cells, marine water, wastewaters etc.) which always display strong background UV absorption and fluorescence. In addition, the UV light used for excitation may be screened off by UV absorbing biomolecules. In view of the above considerations, prior to measurement, the urine samples were centrifuged at 12000 rpm for 10 min and diluted 100 times to reduce the impact of **interfering substance** on adenosine detection. Then the urine samples were spiked with adenosine at different concentration levels (final concentrations: 0, 1, 5, 10, 50, 100, 500, 1000, and 5000 nM). The measurement procedure of adenosine detection in urine samples is the same as that in buffer. A linear regression equation for adenosine, $A_{700}/A_{525} = 0.3171\text{lgC} + 0.299$ ($R = 0.97$), was obtained when the concentration of adenosine was in the range of 1 nM–1 μM (Fig. 8). In addition, according to the recovery rates obtained from urine samples, as shown in Table 2, it shows that the detection results were consistent with the spiked adenosine concentrations, satisfying the practical adenosine detection in real samples.

Conclusion

In summary, a novel colorimetric aptasensor for adenosine detection, which is based on Exo III-assisted DNA cycling amplification and G-quadruplex-induced AuNP aggregation in salt solution. The developed sensor can sensitively and selectively detect adenosine due to the signal amplified effect and the recognition function of aptamers. The red-to-blue color change can be monitored or visualized with the satisfying detection limit of 0.13 nM, and the assay can discriminate adenosine from other interfering substance with high selectivity. Although the aptasensor possesses excellent sensitivity and selectivity, its application in real samples may be hampered by the relatively long reaction time (2 h), which requires further study.

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Compliance with ethical standards The author(s) declare that they have no competing interests.

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