



# Laser ablative TiO<sub>2</sub> and tremella-like CuInS<sub>2</sub> nanocomposites for robust and ultrasensitive photoelectrochemical sensing of let-7a

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## Abstract

A photoelectrochemical (PEC) biosensor based on a multiple signal amplification strategy was established for highly sensitive detection of microRNA (miRNA). TiO<sub>2</sub> was prepared on the surface of titanium sheet by laser etching to improve its stability and photoelectrical properties, and CuInS<sub>2</sub>-sensitized TiO<sub>2</sub> was used to form a superior photoelectrical layer, which realized the initial signal amplification. The electron donor dopamine (DA) was modified to H2 as a signal regulator, which effectively increased the photocurrent signal. To further amplify the signal, an enzyme-free hybridization reaction was implemented. When target let-7a and fuel-DNA (F-DNA) were present, the base of H1 specifically recognized let-7a and forced dopamine@AuNPs-H2 away from the electrode surface. Subsequently, the end base of H1 specifically recognized F-DNA, and let-7a was replaced and recycled to participate in the next cycle. Enzyme-free circulation, as a multifunctional amplification method, ensured the recycling of target molecules. This PEC sensor for let-7a detection showed an excellent linear response from 0.5 to 1000 pM with a detection limit of 0.12 pM. The intra-batch RSD was 3.8% and the recovery was 87.74–108.1%. The sensor was further used for clinical biomolecular monitoring of miRNA, showing excellent quantitative detection capability.

**Keywords** Photoelectrochemical biosensor · Laser ablative TiO<sub>2</sub> · CuInS<sub>2</sub> · MicroRNA

## Introduction

MicroRNA (miRNA) is a kind of endogenous, non-coding, small single-stranded RNA, with a size of about 22 nucleotides, which has a wide range of functions [1, 2]. The

function of miRNA is to regulate the stability or transcriptional efficiency of the target gene miRNA and participate in different levels of life activities. It plays an indispensable role in regulating almost all important life processes, such as cell proliferation [3, 4], apoptosis, development, differentiation, and metabolism [5, 6]. It is closely related to the occurrence and development of tumor and cancer. In recent years, many studies have confirmed that miRNA plays an important role in the pathogenesis of cancer. A large number of miRNA genes are located in cancer-related genomic regions, which are altered in cancerous diseases. Therefore, miRNA is expected to become a therapeutic target for a variety of diseases, used to monitor and treat cancer [7, 8]. Therefore, accurate and rapid monitoring of miRNA is particularly crucial in understanding the pathogenesis and clinical diagnosis of the disease. Up to now, with the rapid advance of science and technology, many methods of detecting miRNA have emerged such as surface plasmon resonance [9, 10], microarray analysis [11], and bioluminescence assay [12]. However, these analytical methods usually have some limitations, such as complex operation, low sensitivity, and expensive instruments. Photoelectrochemical biosensors [13, 14], as an important branch of electrochemical biosensors, have

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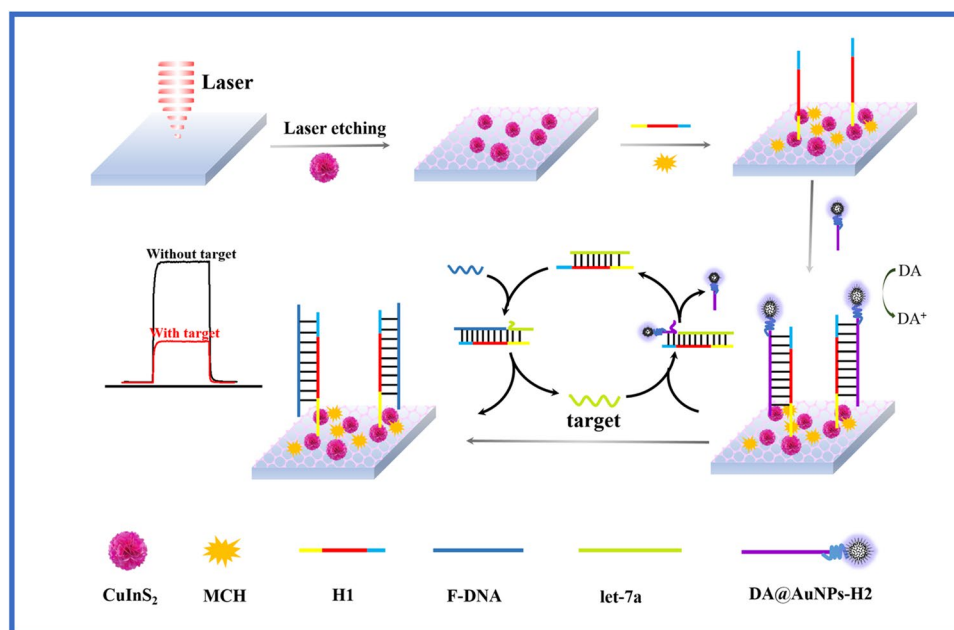
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**Scheme 1** Schematic of photoelectrochemical biosensor for detection of let-7a



attracted much attention in recent years due to their advantages of low background and high sensitivity.

Photoelectrochemical detection is a burgeoning and promising method for biological analysis [15–17]. On the one hand, photoelectric nanomaterials are very important for the construction of photoelectric chemical sensors [18–20]. On the other hand, due to the low abundance and easy degradation of miRNA in biological samples, it is urgent to improve the ultrasensitive detection of miRNA. Therefore, selecting the suitable photoelectric nanomaterials can improve the sensitivity of the sensor [21, 22]. As a typical photosensitive material,  $\text{TiO}_2$  has been widely used on account of its excellent performance [23–25]. Nevertheless, the wide band gap of  $\text{TiO}_2$  results in its ability to utilize only ultraviolet (UV) light in the UV range, which limits the application of  $\text{TiO}_2$  in sensors [26, 27].  $\text{CuInS}_2$  is a classic I–III–VI ternary metal sulfide with a band gap of 1.5 eV, with a wide light absorption range and tunable band gap energy, causing wide attention [28].  $\text{CuInS}_2$  can be well coupled with  $\text{TiO}_2$  and have excellent photoelectrochemical properties compared to other sensing materials, abundant charge transfer properties, and good energy level matching, so as to realize the directional flow of photo-generated carriers in the composite semiconductor, widen the absorption range of light, and improve the utilization ratio of excited light [29, 30]. However,  $\text{TiO}_2$  nanomaterials prepared by conventional synthetic methods were difficult to achieve large-scale periodic preparation, long-term storage, good stability, and patterning electrode, so it was necessary to introduce new technologies. By using the significant local fast photothermal effect induced by focused laser,  $\text{TiO}_2$  nanomaterials were prepared as the working electrode in

a patterned way without interfering with the surrounding materials, replacing the conventional thermal treatment method for the nucleation and growth of nanomaterials. The resulting  $\text{TiO}_2$  not only have better photoelectric properties, larger specific surface area, and more active sites, but also can be manufactured on a large scale and synthesized rapidly. Laser etching synthesis has the advantages of low energy consumption, simplicity, high scalability, mass production, and easy control [31], and has an inestimable application prospect in the field of electrochemistry.

Based on the above advantages, a new PEC sensing platform was developed based on the enzyme-free chain replacement reaction for highly sensitive detection of let-7a. As shown in Scheme 1, firstly, tremella-like  $\text{CuInS}_2$  with the layered nano-flakes were modified onto the laser-etched  $\text{TiO}_2$  flakes. The resulting  $\text{CuInS}_2/\text{TiO}_2$  photoelectric layer had highly photogenerated carrier separation efficiency to visible light. Subsequently, H1 and the sealant 6-mercaptopropionamide (MCH) were introduced. Moreover, with dithiobis (succinimide propionate) (DSP) as the bridge, DA as the electron donor was chemically bonded to H2.  $\text{DA@AuNPs-H2}$  could be used as a probe to amplify the signal of PEC biosensor. With the addition of the target and F-DNA, let-7a replaced H2. The F-DNA was hybridized to H1, and the target let-7a was released from the electrode surface to participate in the next cycle reaction. The signal amplification strategy used in this paper, which has no additional powering or enzymes involved, can better transport carriers and improve photoelectric conversion efficiency. Excitingly, the as-prepared  $\text{TiO}_2$  and  $\text{CuInS}_2$  indeed exhibited distinct PEC response. We greatly hope that this research would inspire interest in the fabrication and development of laser ablative

TiO<sub>2</sub> for sensitive and selective PEC bioanalysis in complex samples.

## Experimental

### Materials and apparatus

The materials and apparatus are described in the Supplementary Material.

### Preparation of TiO<sub>2</sub> nanosheets

Firstly, the original titanium sheet was cleaned by ultrasonic treatment with diluted hydrochloric acid and ultrapure water for 30 min. The computer was connected with the laser system and the experimental parameters were set. The linear scanning mode with laser wavelength of 1064 nm, pulse duration of 100 ns in air atmosphere, average power of 10 W, and scanning speed of 2000 mm s<sup>-1</sup> was used for laser graphics. Subsequently, a size of 5 × 5 mm TiO<sub>2</sub> electrode was produced [32].

### Preparation of tremella-like CuInS<sub>2</sub>

Tremella-like CuInS<sub>2</sub> was prepared based on previous work [33]. Detailed steps are described in the Supplementary Material.

### Preparation of DA@AuNPs-H2

Ten microliters of AuNPs was mixed with 20-μL 2 mM DSP aqueous solution, then 10 μL 10 mM DA was added to the mixed solution and shaken overnight. The obtained solution was mixed with 10 μL 100 nM H2 and placed on the shaker overnight to prepare DA@AuNPs-H2.

### Fabrication of the enzyme-free PEC biosensor

A PEC biosensor mechanism for enzyme-free replacement cycle detection of microRNA is shown in Scheme 1. First of all, based on the advantages of laser etching synthesis, TiO<sub>2</sub> nanosheets (5 × 5 mm) were obtained by laser etching titanium sheets in air atmosphere. After that, the TiO<sub>2</sub> nanosheets were placed in ultrapure water for ultrasonic treatment. An aliquot of 20 μL of CuInS<sub>2</sub> was dripped onto TiO<sub>2</sub> nanosheets for 4 h at room temperature. Then, 10 μL H1 (10 μL) was incubated onto surface CuInS<sub>2</sub>/TiO<sub>2</sub> at room temperature for 16 h. H1 was assembled on the modified electrode by sulfhydryl affinity and washed with ultrapure water to remove excess H1. The nonspecific binding site was sealed with 200 μL MCH (2.0 mM) for 40 min. Thereafter, DA@AuNPs-H2 (10 μL) was

hybridized with H1. However, when the mixed solution of let-7a and F-DNA was incubated onto the-prepared electrode for 2 h at 4 °C, let-7a was recognized by the yellow fragment of H1 to hybridize with it, thus replacing DA@AuNPs-H2. Meanwhile, F-DNA was recognized by the blue part of H1 to replace let-7a; let-7a was released to participate in the next cycle. Accordingly, an enzyme-free circulating PEC biosensor was successfully constructed.

## Results and discussions

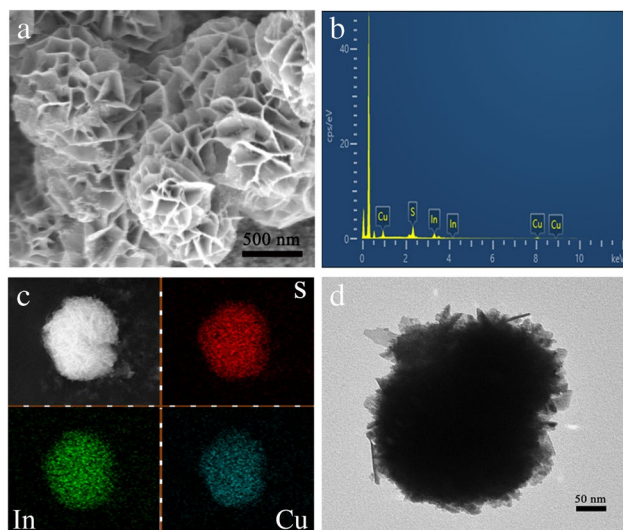
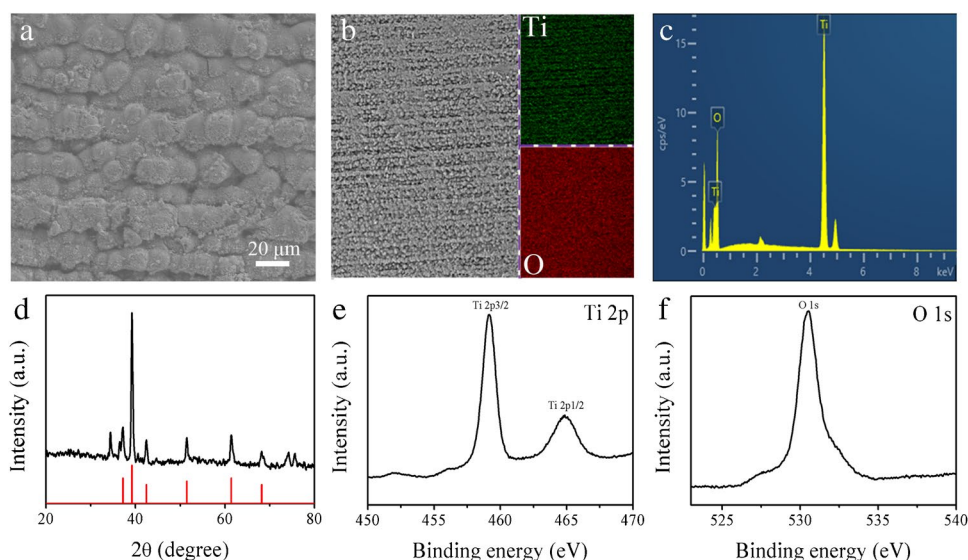
### Characterization of the laser etching TiO<sub>2</sub>

The properties of a substance are closely related to the performance of photoelectrochemical biosensor. It is crucial for the performance of TiO<sub>2</sub> as the initial substrate to build a PEC biosensor. TiO<sub>2</sub> as the substance for working electrode was prepared by laser ablation. As can be seen from the scanning electron microscope (SEM) of Fig. 1a, the etched TiO<sub>2</sub> was arranged in a linear pattern with a width of 20 μm. This concave-convex arrangement allowed TiO<sub>2</sub> with more specific surface area and more sites, resulting in the amplification of photocurrent signal. The element mapping image (Fig. 1b) and energy-dispersive spectrum (Fig. 1c) of TiO<sub>2</sub> showed that Ti and O elements were evenly distributed, confirming the success of laser ablation of TiO<sub>2</sub>. In order to prove the formation of TiO<sub>2</sub>, X-ray diffraction (XRD) technique was used. As shown in Fig. 1d, the XRD diagram of TiO<sub>2</sub> clearly indicated that all of the peaks obtained by laser ablation was located at 2θ = 20–80°, which could be indexed from reflections (JCPDS card no. 34–0180) of TiO<sub>2</sub>. The pattern showed a sharp diffraction peak, indicating high purity and high crystallinity. Finally, X-ray photoelectron spectroscopy (XPS) further provided powerful evidence of the successful laser ablation of TiO<sub>2</sub>. The XPS spectra distinctly showed that the prepared TiO<sub>2</sub> contained Ti and O elements which were two distinct peaks, 459.15 eV and 464.85 eV (Fig. 1e), corresponding to Ti 2p<sub>3/2</sub> and Ti 2p<sub>1/2</sub>, respectively. As shown in Fig. 1f, the O (1 s) peak situated at 530.55 eV was observed, which resulted from surface-bound oxygen-containing species. The results of XPS were consistent with those of XRD and confirmed that the high purity of TiO<sub>2</sub> was successfully prepared.

### Characterization of CuInS<sub>2</sub>

The surface morphology and structure of CuInS<sub>2</sub> nanomaterial were analyzed by SEM characterization. It was observed that CuInS<sub>2</sub> were tremella-like and have average diameters of 1.5 μm (Fig. 2a), which were composed of

**Fig. 1** SEM image (a), element mapping images of TiO<sub>2</sub> (b), and energy dispersive spectrum of TiO<sub>2</sub> (c). **d** XRD patterns of TiO<sub>2</sub>, **e** XPS spectra of element of Ti, **f** XPS spectra of element of O



**Fig. 2** SEM image of the CuInS<sub>2</sub> (a), EDS patterns of the CuInS<sub>2</sub> (b), the corresponding element mapping of S, In, and Cu respectively (c), TEM image of the CuInS<sub>2</sub> (d)

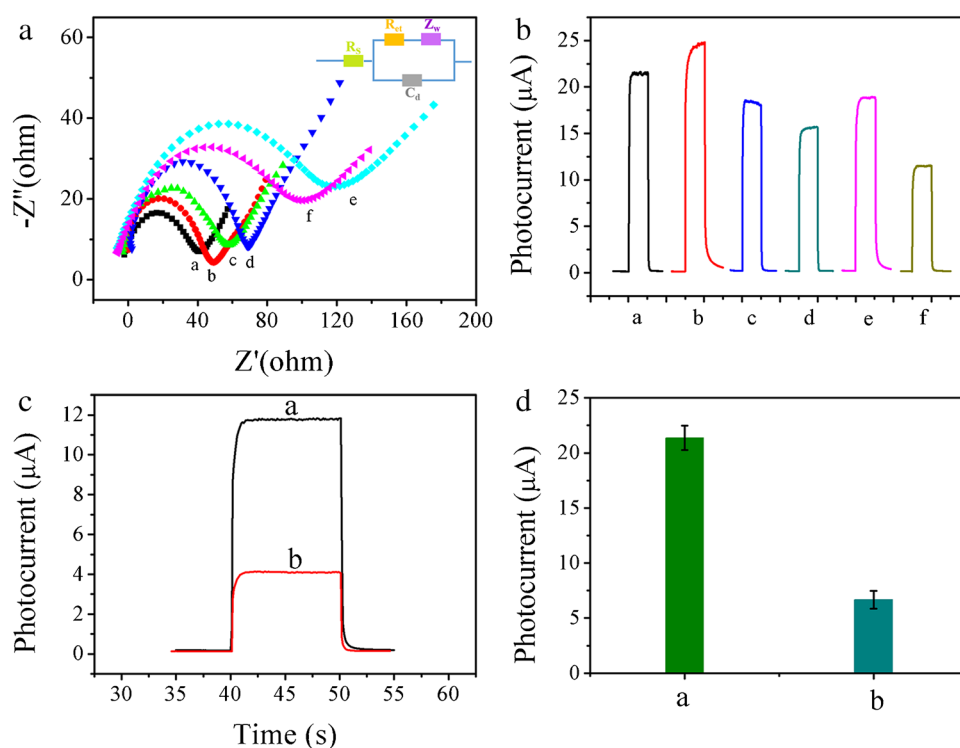
numerous slick nanosheets. This structure greatly increased the specific surface area. Figure 2b was the energy-dispersive spectroscopy (EDS) of tremella-like CuInS<sub>2</sub>, from which the element peak of S, In, and Cu could be clearly observed. Moreover, the EDS mapping results showed the compact location of S, In, and Cu distributed on tremella-like CuInS<sub>2</sub> as shown in Fig. 2c. In addition, as shown in in Fig. 2d, the TEM image of tremella-like CuInS<sub>2</sub> showed that there were many thin sheets on the edge of it, and it was concluded that it was composed of thin nanosheets with a large surface area. And the XPS spectra, XRD spectra, and Raman spectra of CuInS<sub>2</sub> were described in supplementary

materials (Fig. S1). The above results proved that the tremella-like CuInS<sub>2</sub> were synthesized successfully.

### EIS and PEC characterization of the fabricated biosensor

To demonstrate the manufacturing process of PEC biosensor, the electrochemical characterization of the stepwise-modified electrode was executed by electrochemical impedance spectroscopy (EIS) in 5.0 mM [Fe(CN)<sub>6</sub>]<sup>3-/4-</sup> solution containing 0.1 M KCl. The charge transfer resistance ( $R_{et}$ ) corresponds to the semicircle diameter in the higher part of the EIS diagram, and the change in resistance is related to the electrode modification. Figure 3a shows the electrochemical impedance diagram of the modified electrode, and the illustration was the corresponding equivalent circuit diagram. The working electrode of unmodified TiO<sub>2</sub> sheet showed a small semicircle (curve a). After CuInS<sub>2</sub> was assembled on the electrode, the  $R_{et}$  value increased obviously due to the weak conductivity of the semiconductor (curve b). When H1 was fixed to the modified electrode, the  $R_{et}$  value began to increase (curve c). After sealing the electrode with MCH, the  $R_{et}$  value continued to increase (curve d), which was due to the fact that the blocking agent hindered the electron transfer on the surface of the electrode. When the electrode (MCH/H1/CuInS<sub>2</sub>/TiO<sub>2</sub>) was modified by the DA-DSP@AuNPs-H2, the  $R_{et}$  value was distinctly increased (curve e) in virtue of the non-conductivity of the biomolecule and the organic molecule. In the end, when the modified electrode was incubated with the mixed solution of F-DNA and let-7a, the  $R_{et}$  value declined (curve f). The EIS results indicated that the PEC biosensor was successfully fabricated. To

**Fig. 3** **a** EIS characterizations and **b** photocurrent responses of the stepwise modified electrodes: (a)  $\text{TiO}_2$ , (b)  $\text{CuInS}_2/\text{TiO}_2$ , (c)  $\text{H1/CuInS}_2/\text{TiO}_2$ , (d)  $\text{MCH/H1/CuInS}_2/\text{TiO}_2$ , (e)  $\text{DA@AuNPs-H2/MCH/H1/CuInS}_2/\text{TiO}_2$  (f)  $\text{F-DNA/MCH/H1/CuInS}_2/\text{TiO}_2$ . The EIS measurement was performed in 5 mM  $[\text{Fe}(\text{CN})_6]^{3-/4-}$  solution containing 0.1 M KCl. Photocurrent detection was carried out in PBS solution (PH 7.4) containing 0.01 M DA. **c** Photocurrent responses of (a)  $\text{DA@AuNPs-H2/MCH/H1/CuInS}_2/\text{TiO}_2$ , (b)  $\text{F-DNA/MCH/H1/CuInS}_2/\text{TiO}_2$  in PBS solution (PH 7.4). **d** Photocurrent responses of the different methods to obtain electrodes: (a) laser etching of  $\text{TiO}_2$ , (b)  $\text{TiO}_2/\text{GCE}$ . Error bars, S.D.,  $n = 3$



verify the successful construction of the PEC biosensor, the photocurrent response monitoring is shown in Fig. 3b. The experiment was carried out in a PBS solution (pH 7.4) containing 0.01 M DA. The photocurrent signal produced by laser-etched  $\text{TiO}_2$  film under light is shown in the Fig. 3b (curve a). When  $\text{CuInS}_2$  was modified on  $\text{TiO}_2$ , the photocurrent signal increased significantly (curve b). On the one hand, the unique structure of tremella-like  $\text{CuInS}_2$  exhibited photoactivity enhancement under a visible-light irradiation. On the other hand,  $\text{TiO}_2$  and  $\text{CuInS}_2$  can well match energy levels, thus promoting electron transfer and improving photocurrent signal. These two advantages resulted in high photoelectric conversion efficiency. After the immobilization of H1 and MCH, the photocurrent response decreased (curves c, d), which may be attributed to the fact that biomolecules and organic molecules hindered the transfer of electrons, which was not conducive to the separation of photogenerated electron holes. However, the photocurrent response increased followed by the  $\text{DA@AuNPs-H2}$  immobilization (curve e), which was attributed to the change of DA to provide electrons under visible light irradiation, introducing an electron donor as a capture body for light-generated holes. When F-DNA and let-7a were recombined on the modified electrode, the photocurrent signal decreased (curve f). The photocurrent response experiment proved the successful construction of the biosensor.

The DA acted as an electron donor to signal amplification through available inhibiting the electron–hole recombination. To clarify the contribution of DA to signal amplification, a set of tests was implemented in PBS solution (pH 7.4). As can be seen from Fig. 3c, after immersing  $\text{DA@AuNPs-H2/MCH/H1/CuInS}_2/\text{TiO}_2$  electrode in PBS solution (pH 7.4), DA as an electron donor provided electrons to the structure of  $\text{CuInS}_2/\text{TiO}_2$  to produce holes under light excitation; thus, the photocurrent was enhanced (curve a). However, the photocurrent response evidently decreased after modifying the mixed solution with let-7a and F-DNA (curve b). The let-7a and F-DNA hybridized with the H1.  $\text{DA@AuNPs-H2}$  was replaced by let-7a and hybridized with H1, and then let-7a was replaced by F-DNA. The hybridization process caused the release of  $\text{DA@AuNPs-H2}$  from the electrode surface, so the photocurrent signal was reduced. The experimental results showed that the signal amplification strategy was feasible.

The photoelectric properties of substrate materials are particularly important for the construction of photoelectrochemical biosensors. The photocurrent responses of  $\text{TiO}_2$  nanomaterials obtained by different methods were carried out.  $\text{TiO}_2$  materials were prepared by hydrothermal method and the photocurrent of  $\text{TiO}_2/\text{GCE}$  was measured according to the reported literature [34, 35]. As shown in the Fig. 3d, under the same conditions, the photocurrent signal of  $\text{TiO}_2$  directly as electrode obtained by laser etching was larger than that of  $\text{TiO}_2$  modified by hydrothermal method on glassy carbon electrode. Therefore,  $\text{TiO}_2$  obtained by laser

ablation is selected as the electrode of photoelectrochemical biosensor.

### Photoelectrochemical detection for target RNA

Under the best conditions (see supplementary material), the linear relationship of the target let-7a concentration toward PEC signal was studied. Figure 4a displays the photocurrent responses measured at different concentration of let-7a. With the concentration of let-7a increasing, more DA@AuNPs-H2 was replaced by let-7a and away from the electrode surface, leading to a decrease in photocurrent signal. As seen from Fig. 4b, there was a good linear relationship between the photocurrent and the logarithm value of let-7a concentration in the range from 0.5 to 1000 pM. The linear regression equation was  $I = 21.01 - 1.867 \log C_{\text{let-7a}}$ , with a correlation coefficient of 0.994. The limit of detection (LOD) was estimated to be 0.12 pM ( $S/N = 3$ ). Compared with many reported results of study let-7a (Table 1), the detection limit of this PEC sensor is better than them.

### Stability, reproducibility, and selectivity of the PEC biosensor

It is vital for PEC biosensor to evaluate the performance by the stability. As a part of PEC biosensor, the stability of the  $\text{CuInS}_2/\text{TiO}_2$  electrode was also very important during the measurement. Under the condition of repeated light irradiation in PBS with DA, the stability of the sensor was detected by testing the stability of the photocurrent signal of  $\text{CuInS}_2/\text{TiO}_2$  electrode. As could be seen from Fig. 5a, the produced photocurrents on the modified electrode were almost unchanged in “on – off” runs by controlling the light irradiation switch. These results indicated that the prepared photosensitive materials had good stability and could be used to construct PEC sensors. In addition, under the same other conditions, the prepared sensing electrodes were stored at 4 °C and measured at an interval of 1 week. Figure 5b illustrates that the photocurrent response kept at 88.7% of its

**Table 1** Performance of miRNA detection contrasted with other works

| Analytical method                         | Detection limit | Linear range | Refs <sup>a</sup> |
|-------------------------------------------|-----------------|--------------|-------------------|
| Electrochemistry                          | 1 pM            | 1–800 pM     | 1                 |
| Fluorescence                              | 0.4 pM          | 0.5–500 pM   | 2                 |
| Fluorescence                              | 0.78 nM         | 1.56–400 nM  | 3                 |
| Colorimetry                               | 21.7 pM         | 50–10 nM     | 4                 |
| Mesoporous silica/<br>PtNps nanomaterials | 2.56 nM         | 5–100 nM     | 5                 |
| PEC                                       | 0.12 pM         | 0.5–1000 pM  | This work         |

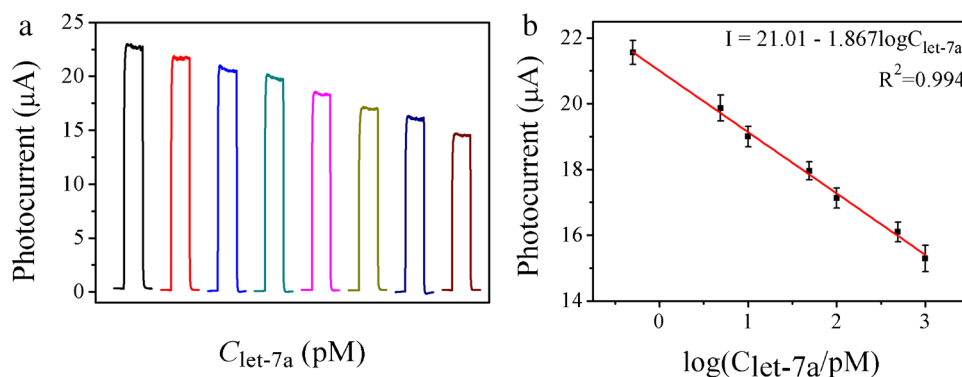
<sup>a</sup>References, see the Supplementary Material

original response within 4 weeks. All results demonstrated the excellent stability of the designed biosensor.

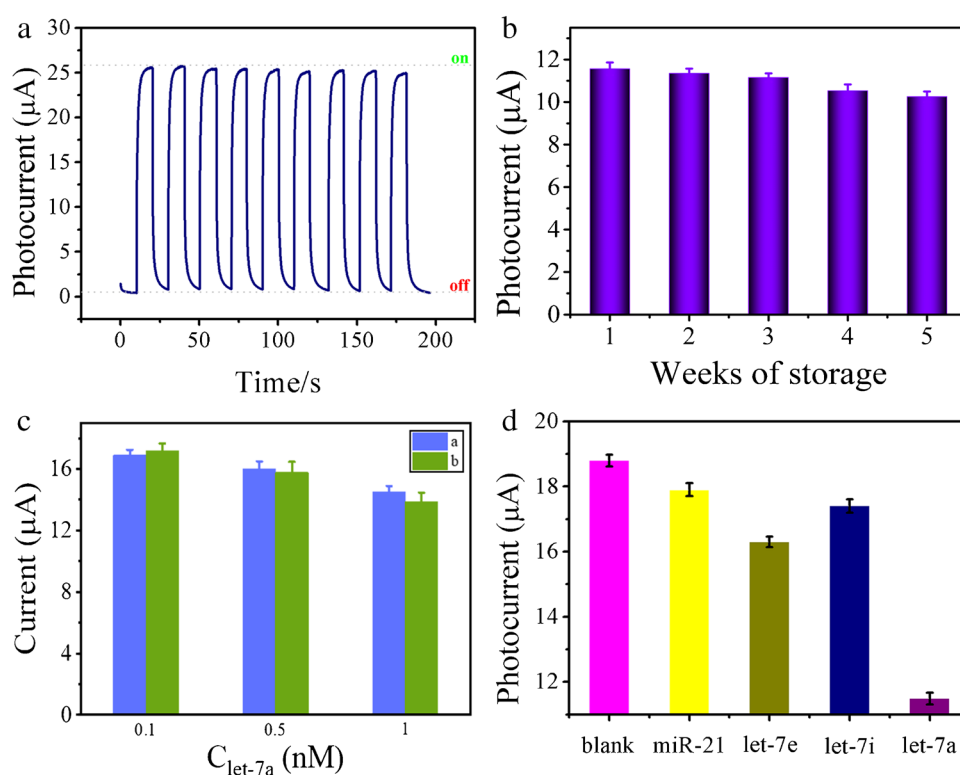
The reproducibility of PEC biosensor was proved by evaluating the relative standard deviation (RSD) of intra-batch and inter-batch. In the process, we used the let-7a at concentrations of 0.1, 0.5, and 1 nM to evaluate reproducibility. Figure 5c presents that the RSD value of intra-batch was 3.8%, 5.6%, 4.5%. The sensor RSD value of inter-batch was 7.8%, 6.6%, 7.5%, respectively. Therefore, the proposed PEC biosensor had a good reproducibility.

Selectivity is one of the most important properties for a biosensor. Whether in point-of-care testing or clinical applications, the high selectivity of biosensors is very significant. Under the best experimental condition, the possible disruptors, such as miR-21, let-7e, let-7i, and let-7a, were incubated with biosensors. As could be seen from Fig. 5d, compared to the let-7a, high current responses of the above compounds were observed, which was in agreement with of the blank. It seemed beyond dispute that with the decrease of paired bases, the chain replacement was difficult to react, and the photocurrent signal had no obvious change. Hence, the experimental results suggested that the designed biosensor had a better selectivity of the target. However, this method still has the limitation that it cannot detect multiple objects simultaneously.

**Fig. 4** **a** Photocurrent responses of the PEC biosensor at various concentrations of let-7a in 6 mL of PBS (pH 7.4) containing 0.01 M DA (0 pM, 0.5 pM, 5 pM, 10 pM, 50 pM, 100 pM, 500 pM, 1000 pM). **b** Relationship between the peak current value and let-7a concentration. Error bars, S.D.,  $n = 3$



**Fig. 5** **a** The stability of the CuInS<sub>2</sub> nanoflowers modified TiO<sub>2</sub> electrode and **b** the proposed PEC biosensor in the refrigerator (4 °C). **c** The photocurrents response of PEC biosensor intra (a) and inter (b) the groups. **d** Photocurrent responses of the PEC biosensor with different sequences of target (10  $\mu$ L, 0.1  $\mu$ M): blank, miR-21 (0.5  $\mu$ M), let-7e (0.5  $\mu$ M), let-7i (0.5  $\mu$ M), and let-7a (0.1  $\mu$ M). Error bars, S.D.,  $n=3$



## Conclusions

A multiple signal amplification PEC biosensor was successfully established for ultrasensitive detection of microRNA with laser ablative TiO<sub>2</sub> and tremella-like CuInS<sub>2</sub> nanocomposites. This sensor had the advantages of ease operation, high sensitivity, strong anti-interference ability, which were attributed to the following points: (1) TiO<sub>2</sub> obtained by laser etching as the substance of the experiment had a more stable photocurrent. (2) CuInS<sub>2</sub> could not only sensitize TiO<sub>2</sub> electrode, but also significantly enhanced photocurrent signal because of its high visible light absorption capacity, which made the PEC biosensor bright in the way of detecting biomolecules. (3) The electron donor DA as a signal amplifier was assembled with a carefully designed H2 (DSP@AuNPs-H2) and applied to a PEC biosensor by base specific recognition. Meanwhile, CuInS<sub>2</sub> can consume electron donor DA efficiently under visible light irradiation because of its excellent light absorption ability. In addition, this highly specific enzyme-free hybridization avoided the effects of the introduction of other enzymes and enabled the target to be recycled. Although this method cannot detect multiple targets simultaneously, it provides a new PEC pathway for the detection of biomolecules with high sensitivity and selectivity.

**Supplementary Information** The online version contains supplementary material available at <https://doi.org/10.1007/s00604-022-05178-9>.

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## Declarations

**Conflict of interest** The authors declare no competing interests.

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