



Development of Electrochemical Biosensor for miR204-Based Cancer Diagnosis

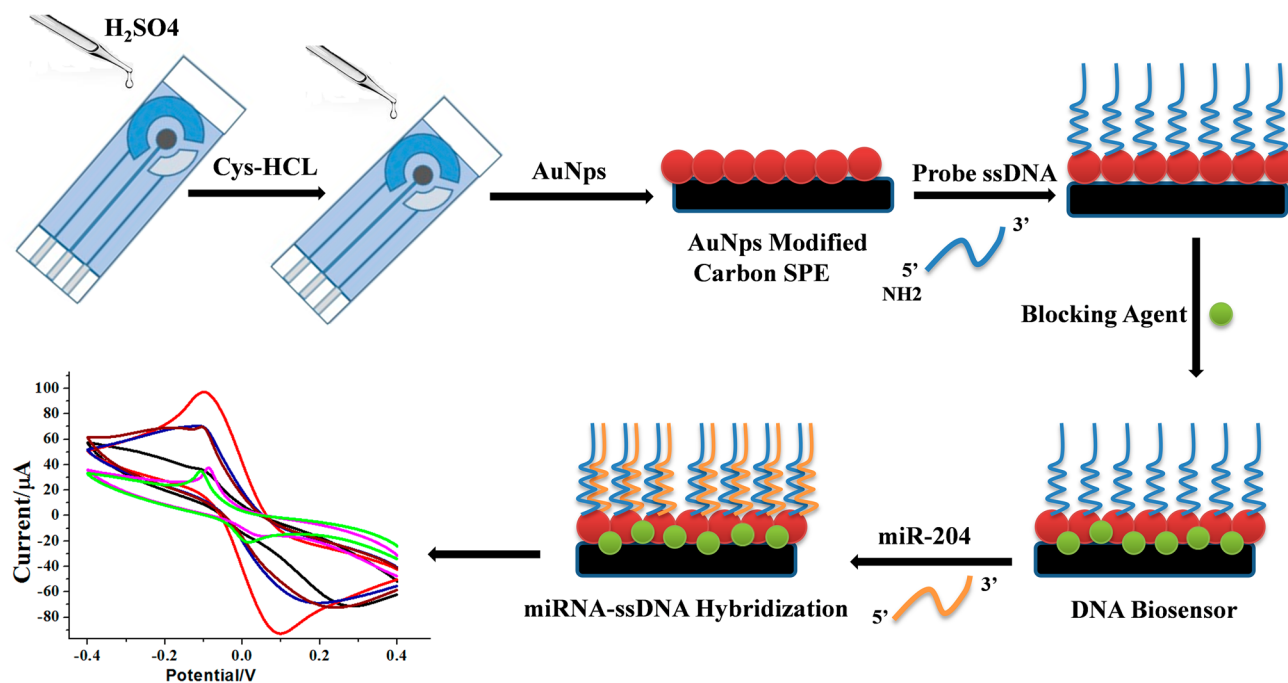
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Abstract

With increase in cancer burden worldwide and poor survival rates due to delayed diagnosis, it is pertinent to develop a device for early diagnosis. We report an electrochemical biosensor for quantification of miRNA-204 (miR-204) biomarker that is dysregulated in most of the cancers. The proposed methodology uses the gold nanoparticles-modified carbon screen-printed electrode for immobilization of single-stranded DNA probe against miR-204. Colloidal gold nanoparticles were synthesized using L-glutamic acid as reducing agent. Nanoparticles were characterized by UV–visible spectroscopy and transmission electron microscopy. Spherical gold nanoparticles were of 7–28 nm in size. Biosensor fabricated using these nanoparticles was characterized by cyclic voltammetry after spiking 0.1 fg/mL–0.1 µg/mL of miR-204 in fetal bovine serum. Response characteristics of the miR-204 biosensor displayed high sensitivity of 8.86 µA/µg/µL/cm² with wide detection range of 15.5 aM to 15.5 nM. The low detection limit makes it suitable for early diagnosis and screening of cancer.

Graphical abstract



Keywords Biosensors · Cancer detection · miR-204 · Electrochemical detection · Early cancer diagnosis

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1 Introduction

In global analysis, cancer is reported to be the second major cause of death and ranks fourth in India [1]. As per GLOBOCAN 2020 statistics, cancer burden is expected to rise from 19.3 million in 2020 to approximately 28.4 million cases in 2040 [2]. Challenges are far greater in developing countries that have limited access to effective diagnostic services such as pathology, laboratory and imaging—all key to timely detection and treatment. Survival chances significantly decrease when cancer is detected at later stages. Classical tumor marker genes, proteins and epigenetic markers have low positivity and poor sensitivity in early stages of cancer. Hence, early diagnosis of cancer requires new paradigms in diagnostic technique as well as biomarker identification [3].

MicroRNAs (miRs) have emerged as a promising biomarker for early detection of tumors as well as other nonmalignant diseases [4]. These miRs are a short stretch of 19–25 nucleotide present in solid tissues as well as cell-free miRNA in different body fluids [5]. The abundant availability and high stability of miRNAs in tissue, blood, urine and other body fluids make it easy to detect with high sensitivity [6]. The miRNA can be upregulated or downregulated in a given pathological condition depending on the body fluid or solid tumor being studied. Presently, these miRNAs are detected by quantitative real-time RT-PCR, northern blotting, bead-based microarray, and deep-sequencing techniques which require trained manpower and laboratory facilities with high end costly instruments [7]. Electrochemical (bio)sensors seem to be a promising cost-effective alternative as point-of-care devices that can be easily operated by any individual [8].

The miR-204 has been reported in various cancers such as colon cancer, gastric cancer, breast cancer, glioma cancer, non-small cell lung cancer, endometrial cancer, papillary thyroid carcinoma, retinoblastoma intrahepatic cholangiocarcinoma, neuroblastoma, nephroblastoma, cervical cancer, bladder cancer, and colorectal cancer, as shown in Table 1. While other novel miRNA biomarkers that are reported are either specific or limited to few cancers only, miR-204 can be a potential candidate for screening of cancers. The dual role of miR-204 as tumor suppressor and oncomiR is reported in studies of upregulation/downregulation of specific genes by miR-204. In most of the studies, miR-204 is reported to be downregulated and its expression levels are inversely correlated with tumor progression. However, there are conflicting reports of over/underexpression of miR-204 in some of the solid tumors such as pancreatic [9], ovarian [10], endometrial [11], breast [12], prostate [13], and colorectal cancer [14]. Segregating studies based on samples studied and analysis of the results show downregulation of miR-204 in serum samples in different carcinomas [15–19], while it is upregulated in other diseases [20–23]. This agrees with cell lines-based studies that report

downregulation of miR-204 in different cancer cell lines [29, 36–40], except one report on breast cancer cell line [41]. However, urine sample analysis shows elevated levels of miR-204 in cancerous condition [24, 25] and downregulated in other diseases [26]. However, tissue sample analysis shows conflicting reports, and based on miR-204 levels it is difficult to differentiate cancerous and other conditions [9, 12, 14, 27–35]. Hence, quantification of miR-204 levels in blood, serum, plasma or urine samples and evaluation of the deviation from standard/control samples can be used for screening of cancer.

The present study reports the development of an electrochemical biosensor for quantification of miR-204. Gold nanoparticles-modified carbon screen-printed electrodes were employed for immobilization of probe DNA against miR-204. The fabricated biosensor could detect miR-204 from 15.5 aM to 15.5 nM. The achieved low detection limit and role of miR-204 in almost all cancers make it a promising tool for screening of cancer.

2 Experimental

2.1 Materials

Tetrachloroauric(III) acid and bovine serum albumin (BSA) from Himedia, trisodium citrate (TSC), potassium ferrocyanide, sodium dihydrogen phosphate, anhydrous sodium carbonate, disodium hydrogen phosphate, potassium chloride, sodium chloride, sodium bicarbonate, Tris-hydrochloride, ethylenediaminetetraacetic acid, polyethylene glycol 4000, and polyethylene glycol 6000 were purchased from CDH chemicals Ltd., cysteamine hydrochloride from Sigma-Aldrich and gelatin from Merck, USA. The ssDNA probe and miRNA were synthesized by Sigma-Aldrich: single-stranded probe DNA 5'-NH₂-AGGCATAGGATGACAAAG GGAA3' and the miRNA-204 with the sequence of 5'UUC CCUUUGUCAUCCUAUGCCU3'.

2.2 Apparatus

The electrochemical measurements were performed using CHI-604E electrochemical workstation from CH Instruments, USA. It was employed for recording the cyclic voltammetric curves after addition of 100 µL of potassium ferri/ferrocyanide solution covering the entire three electrodes. Carbon screen-printed electrodes were from Gwent, comprising a three-electrode system consisting of a working electrode (C2070424P2 carbon graphite ink containing Prussian blue mediator, 3 mm in diameter) as well as a counter electrode made up of carbon and a reference electrode of Ag/AgCl. All electrochemical measurements were carried out in 100 µL of 5 mM of potassium ferri/ferrocyanide 0.01 M PBS, pH 7.4.

Table 1 Representation of dual regulatory role of miR-204 in various cancers/diseases

Reference	Type of cancer/disease	Expression patterns of miR-204	Reported samples
Blood/serum/plasma samples			
Chen et al. [15]	Gastric cancer (GC)	Downregulated	Serum samples of GC patient
Wu et al. [16]	Endometrial cancer	Downregulated	serum
Zheng et al. [17]	Lung cancer	Downregulated	Serum sample of patient lung cancer
Nie et al. [18]	Acute myeloid leukemia (AML)	Downregulated	AML serum samples and Kasumi-1 and HL-60 cells
Guo et al. [19]	Non-small cell lung cancer (NSCLC)	Downregulated	Plasma patient of NSCLC
Gaddam et al. [20]	Non-dipping hypertension in diabetic patients (heart and kidney failure)	Upregulated	Blood
Cheng et al. [21]	Chronic kidney diseases	Upregulated	Blood
Chiu et al. [22]	Parkinson's diseases (PD)	Upregulated	Serum samples of PD patient
Xu et al. [23]	Type 1 diabetes	Upregulated	Serum sample
Urine samples			
Kurahashi et al. [24]	Renal cell carcinoma	Upregulated	Urinary exosomes of renal PRCTFE3 Tg mice
Debernardi et al. [25]	Stage 1 pancreatic ductal adenocarcinoma	Upregulated	Urine sample of PDAC patient
Tapia-Castillo and Guanzon et al. [26]	High blood pressure	Downregulated	Urinary exosomes
Tissue samples			
Hofbauer et al. [27]	Kidney patients with hypertension, hypertensive nephrosclerosis, or diabetic nephropathy	Downregulated	Human kidney tissue
Scian et al. [28]	Chronic allograft dysfunction (CAD)	Downregulated	Kidney tissue samples
Turner et al. [29]	Prostate cancer	Upregulated	Tumor of prostate cancer patient
Li et al. [12]	Breast cancer	Downregulated	Breast cancer tissue
Lee et al. [30]	Breast cancer	Downregulated	Breast cancer tissue
Hong et al. [31]	Breast cancer	Downregulated	Breast cancer tissue
Hoffmann et al. [32]	Secondary cataract (SC)	Upregulated	Lens tissue
Sümbül et al. [14]	Colorectal cancer	Upregulated	Colorectal cancer tissue
Yin et al. [33]	Colorectal cancer	Downregulated	Colorectal cancer tissue
Khalid et al. [9]	Pancreatic ductal adenocarcinoma (PDAC)	Downregulated	PDAC tissue and cell lines SW1990 and PANC-1
Conte et al. [34]	Inherited retinal dystrophy associated with ocular coloboma	Downregulated	Eye tissue samples
Qiu et al. [35]	Intrahepatic cholangiocarcinoma (ICC)	Downregulated	Human ICC tissue and cell lines ICC-9810
Cell lines			
Shen et al. [36]	Breast cancer	Downregulated	Breast cancer MCF-7 cells
Guo et al. [37]	Bladder cancer (BC)	Downregulated	Urine sample of BC patients
Ding et al. [29]	Prostate cancer	Downregulated	PAC and NPEC cells
Ying et al. [38]	Glioma	Downregulated	SNB19 and LN382T glioma cell lines
Ryan et al. [39]	Neuroblastoma	Downregulated	Neuroblastoma cell lines
Zhang et al. [40]	Gastric Cancer(GC)	Downregulated	AGS and BGC gastric cancer cell lines
Findlay et al. [41]	Breast cancer	Upregulated	Breast cancer cell line MCF7 and human breast tumor

2.3 Gold Nanoparticle Synthesis Using L-Glutamic Acid

In the proposed methodology, the gold nanostructures were synthesized using modified Turkevich method as described by Shikha et al. [42]. In the experiment, 100 mL of 0.5 mM

tetrachloroauric(III) acid prepared in deionized water (18.2 MΩ) in a conical flask was brought to boiling condition on a magnetic stirrer with heating. To the boiling chloroauric acid precursor solution, 3 mL of 100 mM L-glutamic acid was injected at a flow rate of 200 µL/min, while heating and vigorously stirring the solution. To minimize the evaporation

and contamination during synthesis, the flask was covered with a Petri dish which helps in achieving the objective.

2.4 Biosensor Fabrication

The schematic representation of the step-by-step biosensor fabrication for quantification of miR-204 is shown in Fig. 1. The working electrode of 2 mm diameter and counter electrode having area 5 mm (ratio of area working/counter electrode = 3) were of carbon, while reference electrode was made up of silver. The screen-printed carbon electrode was cleaned by treating the working electrode surface by 0.1 M sulfuric acid to remove pre-adsorbed impurities. The sulfuric acid-treated electrode was then washed with deionized water and completely air dried. This was followed by treatment of the working electrode surface with 7 μ L cysteamine hydrochloride with the desired concentration (2 mg/mL) for 1 h under dark condition. This leads to the thiolene reaction ($C=C-SH$) thereby yielding thio-functionalized electrode surface. Excess of cysteamine hydrochloride was washed by rinsing the electrode with deionized water. Addition of 7 μ L of synthesized gold nanoparticles (AuNPs) onto the working electrode and incubation for 1 h at room temperature result in nanoparticles-catalyzed amide bond formation between the COO^- groups on the surface of AuNPs and NH_3^+ groups of cysteamine hydrochloride on the working electrode surface. Unbound gold nanoparticles are removed by washing with deionized water. The AuNPs-modified

electrode surface is treated with 7 μ L of 0.5 μ g/mL of ssDNA probe and incubated for 1 h at room temperature. The free citrate groups on the surface of AuNPs form amide linkage with amino-modified free 5' end of the ssDNA probe. Unbound or physically adsorbed probe was removed by washing with deionized water, followed by air drying under ambient conditions. Finally, the fabricated biosensor was treated with 1% gelatin, as a blocking agent to avoid non-specific interactions on the electrode surface. After 1 h of incubation with blocking agent at 37 $^{\circ}C$, the electrodes were washed with deionized water and air dried. The total time taken for complete fabrication of DNA-based biosensor is about 4 h.

2.5 Electrochemical Measurements

Cyclic voltammetry (CV) was employed to determine the electrochemical properties of the fabricated DNA-based biosensor. Electrochemical measurement was done by adding 100 μ L of 5 mM ferri/ferrocyanide solution onto the fabricated biosensor surface covering all three working, reference and counter electrodes. A sweeping potential of -0.9 to 0.9 V at a scan rate of 100 mV/s was applied and a cyclic voltammogram was recorded. Response characteristics of the biosensor were evaluated by preparing different concentrations of miR-204 (0.1 fg/mL–0.1 μ g/mL) spiked in fetal bovine serum (FBS) and recording the cyclic voltammogram in a suitable potential window.

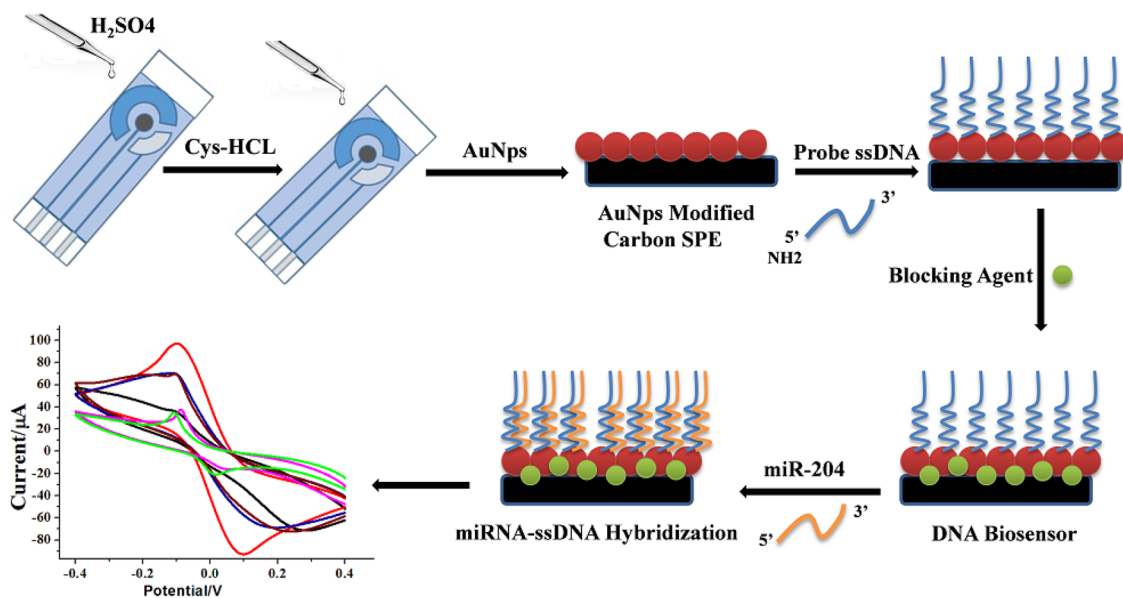


Fig. 1 Schematic representation of stepwise DNA-based biosensor fabrication

3 Results and Discussion

3.1 Gold Nanoparticles

Colloidal gold nanoparticles synthesized using L-glutamic acid were characterized by transmission electron microscopy and UV–visible absorption spectroscopy. Transmission electron microscopic image shows spherical nanoparticles of 7–28 nm in size as shown in Figure 2a. The absorption spectra of the colloidal suspension show an absorption peak with maxima at 528 nm. The choice of the gold nanoparticles for biosensor development was dictated by performance parameters such as sensitivity, detection range, and stability or reproducibility of the biosensor. Gold nanoparticles improve the sensitivity of the biosensor owing to high conductivity and at the same time not having oxidation issues like silver and copper nanoparticles. In addition to this, the usage of nanoparticles in general for biosensor fabrication leads to increase in the effective surface area of the electrode. This helps in increasing the loading capacity of the probe DNA (biorecognition molecule) without increasing the size of the biosensor. Increased loading capacity is in turn responsible for a wider detection range of the biosensor.

3.2 Biosensor Characterization

The biosensor fabrication/surface functionalization at each step was confirmed by cyclic voltammetry (CV) response analysis. Figure 3 shows voltammogram of carbon screen-printed electrode (CSPE) recorded in 0.5 mM ferri/ferrocyanide solution in a potential window of -0.4 to 0.4 V. As observed from Fig. 3, bare CSPE (black solid line) shows redox peaks at cathodic potential of -0.115 V (E_{PC}) and anodic potential of 0.2623 V (E_{PA}) and the corresponding

peak currents were $37.02 \mu\text{A}$ (I_{PC}) and $-60.59 \mu\text{A}$ (I_{PA}). After surface functionalization of carbon screen-printed electrode with cysteamine hydrochloride, the cathodic and anodic peak current increased to $96.2 \mu\text{A}$ (I_{PC}) and $92.7 \mu\text{A}$ (I_{PA}), due to ionic character of cysteamine hydrochloride (red-colored dash). Immobilization of gold nanoparticles onto cysteamine-modified CSPE results in decrease in the peak current to $69.54 \mu\text{A}$ (I_{PC}) and $-68.14 \mu\text{A}$ (I_{PA}) (blue-colored dot) due to the semiconducting nature of gold nanoparticles. Finally, after immobilization of the single-stranded probe DNA, negligible change in peak current is observed (I_{PC} : $-69.4 \mu\text{A}$ and I_{PA} : $-58.83 \mu\text{A}$) since ssDNA is a charged biomolecule (brown-colored dash dot). Hence, despite having insulating character of biomolecules, the expected reduction in current is not observed since DNA is a charged molecule. Blocking the electrode surface with gelatin increases the insulating character, hence a sharp decline

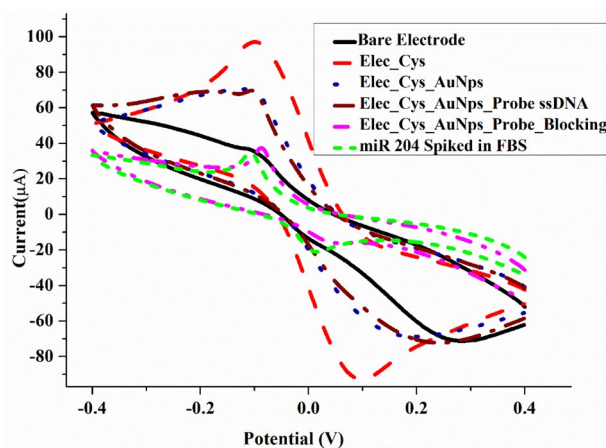


Fig. 3 Cyclic voltammetry of carbon screen-printed electrodes after modification with CysHCl, AuNPs, probe, blocking buffer and miR-204

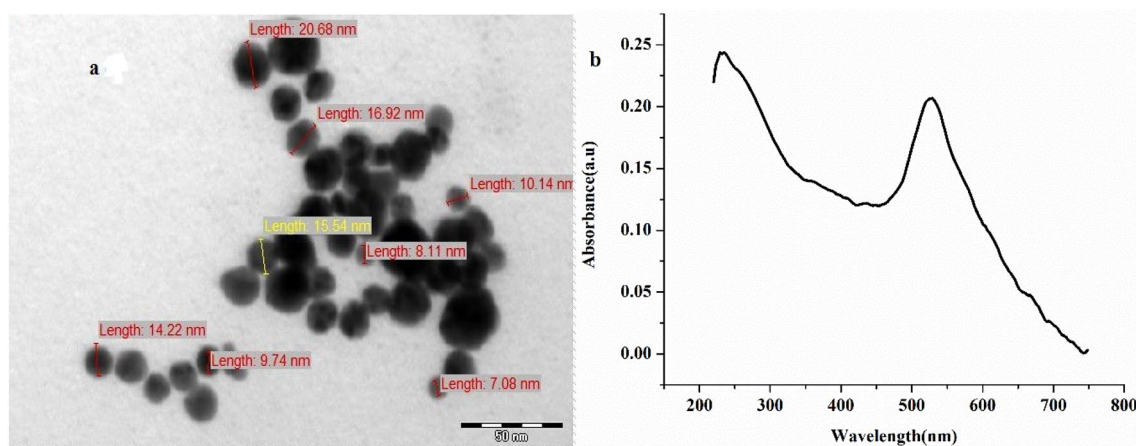


Fig. 2 **a** TEM micrograph of gold nanoparticles synthesized using L-glutamic acid as reducing agent, **b** UV–visible absorption spectra of synthesized gold nanoparticles

in peak current is observed with I_{PC} as 30.4 μA and I_{PA} as 12.78 μA (magenta-colored dash dot dot).

The fabricated biosensor was further confirmed by incubating it with a sample containing miR-204 spiked in FBS. After hybridization, the peak potential shifts to lower potential, i.e., cathodic potential of -0.108 V (E_{PC}) and anodic potential of 0.014 V (E_{PA}). Also, the corresponding peak currents of 38.22 μA and 20.8 μA , respectively (Green coloured short dash), were observed. The calculated ΔE_p value is 0.13 V which shows that the system follows Nernst equation. Also, low redox potential decreases the probability of undesirable redox reactions from other redox species (like uric and ascorbic acid) present in the sample and hence decreased interference is expected. The variation in the potential and

current response after addition of the sample is indicative of the modification of the surface with gold nanoparticles and probe ssDNA.

Further, biosensor response characteristics were analyzed by cyclic voltammetric studies varying the scan rate from 10 to 150 mVs^{-1} . As observed from Fig. 4a, with increasing scan rate, the peak current increases to keep the charge constant as given by Eq. (1),

$$\text{Charge(coulomb)} = \text{current(ampere)} \times \text{time(second)}. \quad (1)$$

Also, at low scan rates, the ratio of $I_{PC}/I_{PA} < 1$ is indicative of the quasi reversible nature of the reaction, though the ratio approaches one at 100 mV/s . The plot of anodic

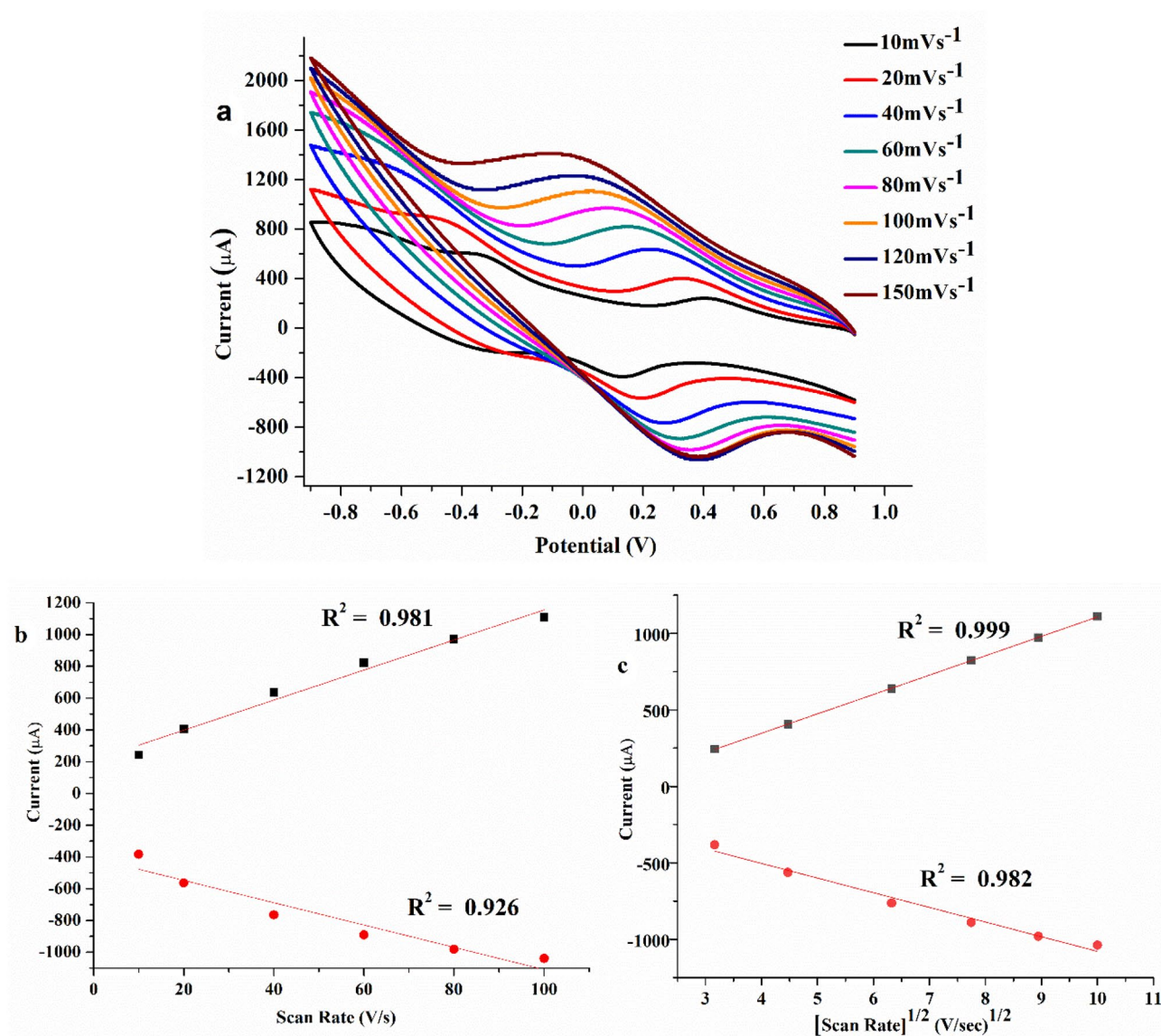


Fig. 4 a Cyclic voltammogram of fabricated biosensor recorded by varying scan rates for 0.1 μg of miR-204 spiked in FBS. b Variation of peak current as a function of scan rate and c variation of peak current as a function of square root of scan rate

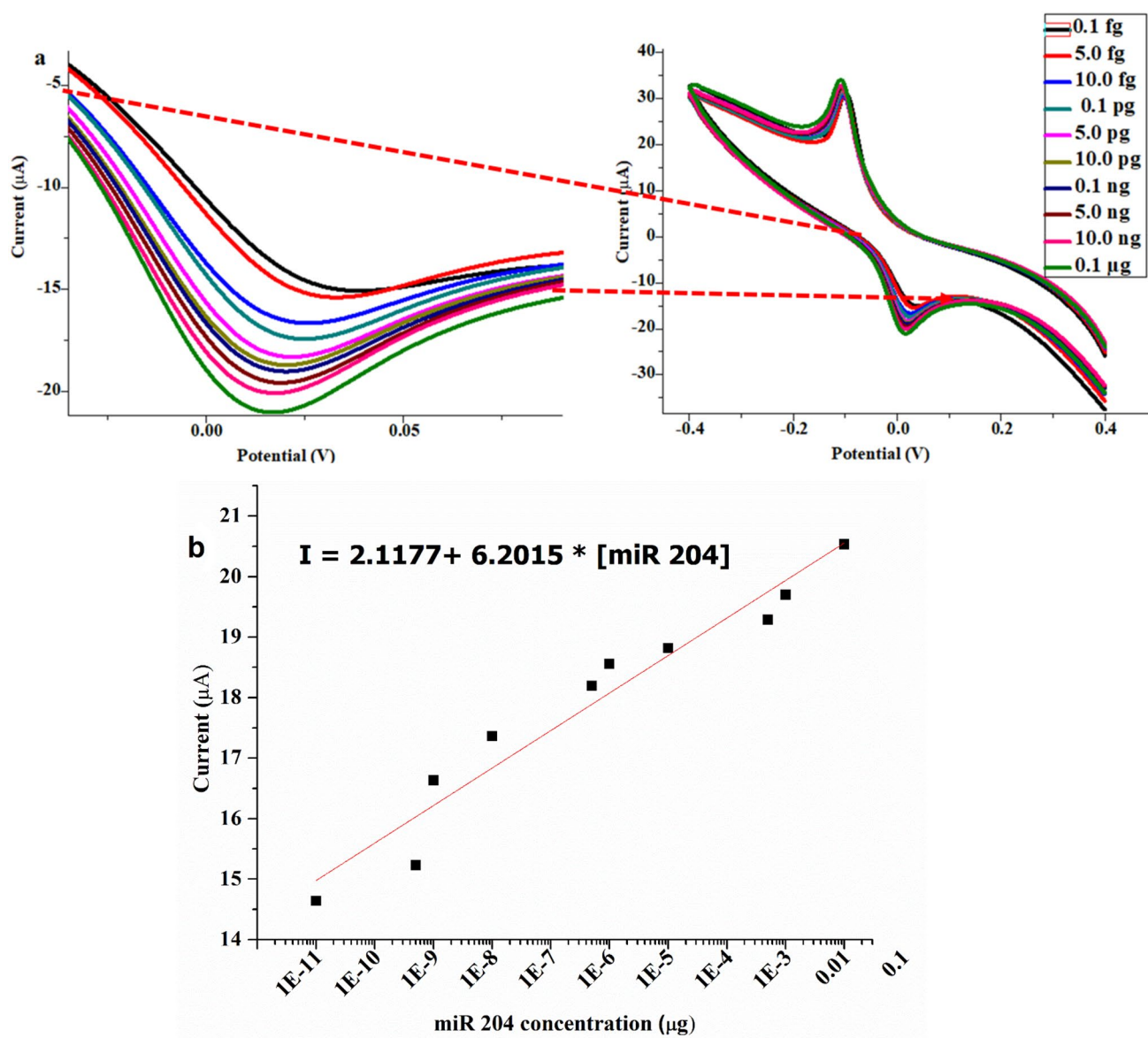


Fig. 5 **a** Cyclic voltammetry response of fabricated biosensor after hybridization with different concentrations ranging from (0.1 fg to 0.1 μg) of miR-204 spiked with FBS. **b** Calibration plot showing the peak current as a function of miR-204 concentration

and cathodic peak current as a function of scan rate (ν) and square root of scan rate ($\nu^{1/2}$) is shown in Fig. 4b and c, respectively. The linear regression analysis of the two curves shows R^2 of 0.99 for variation of peak current with $\nu^{1/2}$, while poor linear fit was observed for peak current vs scan rate ν . This suggest the reaction to be diffusion controlled.

3.3 Quantification of miR-204

Quantification of miR-204 was carried out by recording the cyclic voltammetric responses at a scan rate of 100 mV/s

from -0.4 to 0.4 V for different concentrations (0.1 fg/mL–0.1 μg/mL) spiked in fetal bovine serum (FBS) (as shown in Fig. 5a).

The peak current increases with increasing concentration of the miR-204 in sample. The cathodic peak shows poorer resolution as compared to the anodic peak. The calibration plot of anodic peak current with concentration of miR-204 is shown in Fig. 5b. Figure 5a inset shows the enlarged image of the anodic peak. The sensitivity of the fabricated biosensor having working electrode of 3 mm diameter was found to be:

Table 2 RNA-based electrochemical biosensor for detection of cancer

Cancer Type	miRNA detected	Sample	Detection methodology	LineatDetection range	Limit of detection	Time for Hybridization	Type of Electrode Biosensor	Ref
Breast cancer	miR-21	Breast cancer cell line	DPV	–	6 pM	20 min	Graphite pencil electrode Voltammetric measurements of oxidation signals are recorded by the enzymatic reaction product	Kilic et al. [43]
General cancer	Let-7b	Hela cell lines and A549 cell lines	DPV	10 fM–0.1 nM	3.55 fM	4 h	Pd–Pt supported functionalized gold SPE, triple amplification signal of redox cycling of enzyme phosphates	Chen et al. [44]
Prostate cancer	miR-141	T24 cell line	EIS and SWV	0.1 fM–0.2 pM	2.3×10^{-13} M	2 h	Methylene blue-labeled DNA probe is attached on modified SPE, dual amplification strand displacement reaction by Ca^{+2} -dependent DNase	Yang, et al. [45]
General cancer	miR-21	Esophageal tumoral tissue and adjacent normal tissues collected	Chronoamperometry	10 fM–10 nM	10 fM	5 h	TSP modified gold electrode, cyclic voltammetric and amperometric detection of redox reaction by avidin-HRP or high-activity poly-HRP80	Wen et al. [46]
General cancer	miR-122b	Chemically synthesized lyophilized powder, Shanghai, China	Chronoamperometric	10 aM–1 nM	10 aM	2 h	SAM-modified surface of gold electrodes with microRNA helper, electroreduction current of miR-122b is measured by chronoamperometry	Wen et al. [47]
avian leukemia	miR-21	DF-1 chicken fibroblast cells	DPV	0.5–100 fM	0.17 fM	2 h	AuNPs-modified gold SPE, enzymatic amplification of T7 exonuclease for ultrasensitive detection of miRNAs	Ge et al. [48]
Liver cancer	miR-122	Chemically synthesized lyophilized powder	DPV	0.005–1 mM	0.1 pM	1 h	Graphite SPE, electrochemical genosensor detection of label-free guanine oxidation during hybridization of DNA–RNA	Lusi, E. A., et al [49]

Table 2 (continued)

Cancer Type	miRNA detected	Sample	Detection methodology	Linear/Detection range	Limit of detection	Time for Hybridization	Type of Electrode Biosensor	Ref
General cancer	miR-155	Chemically synthesized from TaKaRa, Dalian, China	CV	5.6 pM–0.56 μ M	1.87 pM	16 h	Glassy carbon electrode	Wu et al. [50]
General cancer	Let-7b	Chemically synthesized from TaKaRa, Dalian, China	Amperometry	100 pM–100 nM	3.4 aM	2 h	Modified gold SPE allos- teric molecular beacons, aptamer that generates electrochemical signal for displacement of complicated DNA hair- pin structure	Cai et al. [51]
Breast cancer	miR-21	Chemically synthesized from Sangon, Shanghai, China	SWV	10 fM–1 nM	2.9 fM	1.5 h	Glassy carbon electrode; the SWV signals were generated by the reduc- tion reaction of methyl- ene blue (MB)	Ma et al. [52]

$$\begin{aligned}
 \text{Sensitivity} &= \frac{\text{Slope of calibration plot}}{\text{Area of working electrode}} \\
 &= \frac{0.62 \mu\text{A}/\mu\text{g/ml}}{0.07 \text{ cm}^2} \\
 &= 8.86 \mu\text{g}/\mu\text{L}/\text{cm}^2.
 \end{aligned} \quad (2)$$

The linear range of the biosensor was observed to be 15.5 aM–15.5 nM with lower detection limit less than 15 aM (0.1 fg/mL). Till date, to the best of our knowledge, no biosensor literature is available for miR-204; hence, we have compared the performance of the fabricated biosensor with other miRNA-based biosensor reported in literature. As shown in Table 2, the detection limit of the developed miR-204 biosensor is comparable to the best known biosensor for miRNA (miR-122b), while the linear range of the fabricated biosensor for miR-204 is much wider than that of any of the listed miRNA biosensors. The wide linear range makes it a suitable diagnostic tool for stage I to stage IV cancer.

4 Conclusion

The present study describes L-glutamic acid-synthesized gold nanoparticles which were fabricated on carbon screen-printed electrode. This modified carbon screen-printed electrode was attached with probe ssDNA. Later, to avoid non-specific interactions and matrix effects, the surface is blocked with gelatin. Next, the hybridization process is carried out by adding miR-204 for 1 h at room temperature out of 4 h of the total fabrication process. The generated current response was analysed by using cyclic voltammetry (CV). The results of this experiment showed that the proposed biosensor displays a wide linear range with lower detection limit less than 0.1 fg/mL, making it suitable for early diagnosis and prognosis of cancer. DNA-based biosensor developed using these nanoparticles can be used for detection and quantification of other interesting miRNAs in biosamples. Future research will move toward effective, easy, miniaturized, miRNA-based biosensors for clinical applications.

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Declarations

Conflict of Interest The authors state that there exists no conflict of interest toward this work.

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