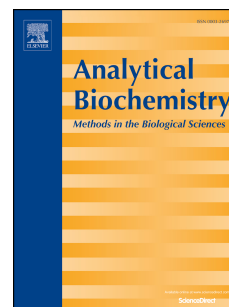


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Rapid and sensitive detection of *Salmonella* with reduced graphene oxide-carbon nanotube based electrochemical aptasensor

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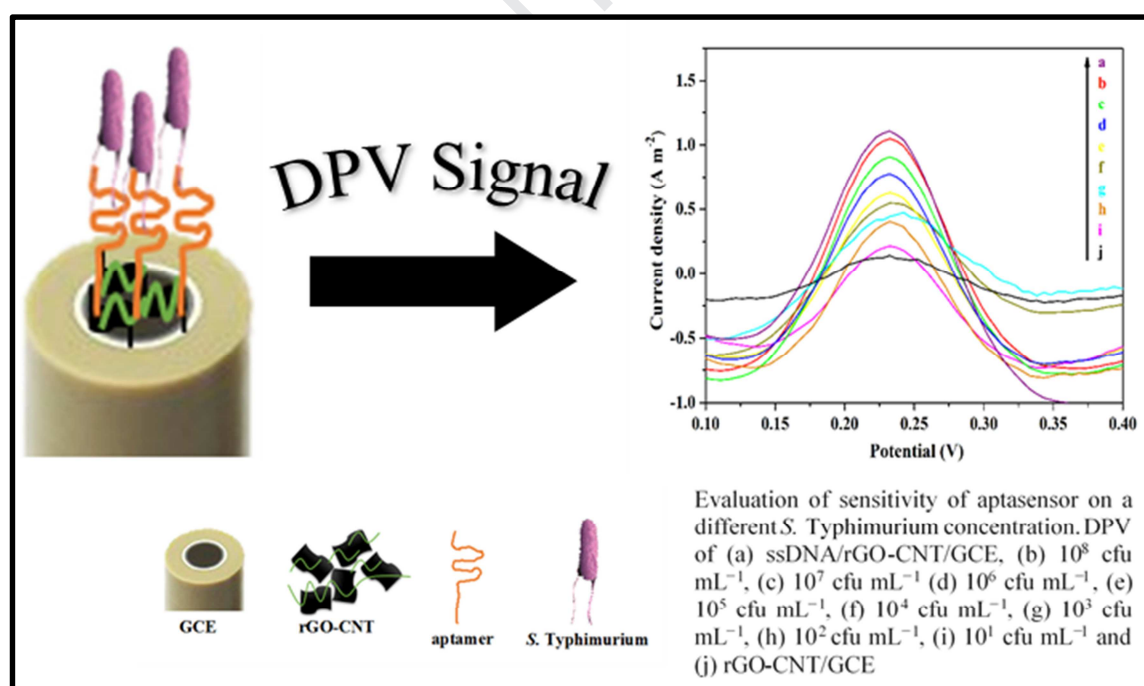
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Rapid and sensitive detection of *Salmonella* with reduced graphene oxide-carbon nanotube based electrochemical aptasensor

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Abstract

Rapid detection of foodborne pathogens is crucial as ingestion of contaminated food products may endanger human health. Thus, the objective of this study was to develop a biosensor using reduced graphene oxide-carbon nanotubes (rGO-CNT) nanocomposite via the hydrothermal method for accurate and rapid label-free electrochemical detection of pathogenic bacteria such as *Salmonella enterica*. The rGO-CNT nanocomposite was characterized using Fourier transform infrared spectroscopy, Raman spectroscopy, X-ray diffraction and transmission electron microscopy. The nanocomposite was dropped cast on the glassy carbon electrode and further modified with amino-modified DNA aptamer. The resultant ssDNA/rGO-CNT/GCE aptasensor was then used to detect bacteria by using differential pulse voltammetry (DPV) technique. Synergistic effects of aptasensor was evident through the combination of enhanced electrical properties and facile chemical functionality of both rGO and CNT for the stable interface. Under optimal experimental conditions, the aptasensor could detect *S. Typhimurium* in a wide linear dynamic range from 10^1 until 10^8 cfu mL⁻¹ with a 10^1 cfu mL⁻¹ of the limit of detection. This aptasensor also showed good sensitivity, selectivity and specificity for the detection of microorganisms. Furthermore, we have successfully applied the aptasensor for *S. Typhimurium* detection in real food samples.

Keywords: Aptasensor; Aptamer; Reduced Graphene Oxide; Carbon nanotubes; *Salmonella*

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1 Introduction

Salmonella enterica is a major foodborne pathogen causing foodborne illnesses globally, including Asian countries such as Singapore, Malaysia, Japan and Thailand [1]. In Malaysia, 48.51 cases of *Salmonella* associated foodborne illnesses per 100,000 people have been reported by the Ministry of Health, Malaysia [2]. *Salmonella enterica* is mainly transmitted through consumption of contaminated raw or uncooked food products (e.g. vegetables, fruits, poultry, meat and eggs) and may cause serious risk to consumers' health [3]. Lack of hygiene practice in production, handling and transportation of food products could lead to easy contamination with *Salmonella*. The increasing numbers of food poisoning cases due to *Salmonella enterica* implied that there is a crucial need for a more sensitive and rapid detection method to detect and quantitate the presence of this pathogen in food products [4]. Early detection of *Salmonella* in the food chain will help to prevent the spread of nontyphoidal salmonellosis [5].

Among the *Salmonella enterica*, *S. Typhimurium* is the most common serovar associated with food poisoning outbreaks in humans. Methods that are routinely used for the detection of *S. Typhimurium* include culture-based methods, PCR-based assays or immunoassays that are usually time-consuming, costly, require skilled personnel and sophisticated instruments [4]. Therefore, there is a concerted effort to develop alternative approach such as biosensors which are rapid, sensitive and accurate in detecting *S. Typhimurium* [6]. Although there are several types of biosensors, the electrochemical-based biosensors are favoured due to their low cost, miniaturization capacity, sustainability and minimal use of reagents and samples [7]. Furthermore, the use of aptamers in the electrochemical-based biosensor platform further improves sensitivity and affinity as aptamers are highly specific towards target molecules [8].

DNA or RNA aptamers are single-stranded oligonucleotides and highly selective towards the target molecule [9]. Typically, the aptamers are selected through an *in vitro* method referred to as SELEX (Systemic Evolution of Ligands by EXponential enrichment) [10]. The benefits of aptamer usage include easy modification, low cost, fast synthesis and its stability. Moreover, aptasensors (aptamer-based biosensors) have been coupled with an electrochemical-based platform to facilitate detection of pathogens and quantification especially for food safety requirements [1,11]. This aptasensor is very useful and highly desirable because of its high productivity, low limit of detection, simplicity, portability and most importantly, it requires a low volume of analytes [12]. Several aptamer sequences have been designed for the detection of various target bacteria including *S. Typhimurium*, *Staphylococcus aureus*, *Vibrio parahaemolyticus*, *Listeria monocytogenes* and *Shigella dysenteriae* [13,11]. The aptamer sequence used in this study was designed by Joshi *et al.* to complement an outer membrane protein that is found on *S. Typhimurium* as the specific binding target [14].

Excellent physicochemical attributes of carbon-based nanomaterials enhanced the increasing use of these materials as a component in biosensors [15]. Specifically, high electrical conductivity and enhanced surface area of carbon-based nanomaterials in combination with aptamers as the recognition element improve the overall sensitivity of the aptasensors [16]. Graphene, a carbon-based nanomaterial, has emerged as an attractive nanomaterial in the field of electrochemical application mainly due to its excellent electrical and thermal conductivity [17]. However, the methods of graphene synthesis which include chemical vapor deposition and mechanical exfoliation are too costly and produce low yield [18]. Therefore, chemical or hydrothermal reduction of graphene oxide to generate reduced graphene oxide (rGO) has become an inexpensive procedure to imitate qualities of pristine graphene with similar mechanical and electronic properties [19]. Properties of rGO such as

increased current density and chemical inertness along with biocompatibility facilitate rGO surface modification through hydrogen bonding or π - π stacking [20,21]. Yet, less than optimal performances have been noted for rGO due to interlayer effects of van der Waals interactions which reduce the total surface area of rGO due to the restacking of graphene sheets [22]. Therefore, the hybridization of rGO with carbon nanotubes (CNT) has prevented the aggregation of graphene sheets as CNT acts as a one-dimensional spacer between the sheets [23]. Additionally, high surface area and excellent electrical conductivity of CNT further complement rGO for sensing purpose. The increased surface area in rGO-CNT hybrid nanocomposite allows covalent binding between the carboxyl group of the rGO-CNT and amino-modified aptamers, resulting in a highly stable and sensitive aptasensor [24].

The rGO-CNT has been developed recently as an aptasensor by some researchers to detect the biomarker prostate specific antigen in serum [25], breast cancer biomarker HER2 [26] and lysozyme [27]. Previously, we developed an electrochemical aptasensor using rGO-azophoxine which exhibited a wide detection range from 10^1 to 10^8 cfu mL⁻¹ for *S. Typhimurium* [1]. The physicochemical characteristics of rGO such as high electrical and thermal conductivity, and the fibrous structure of CNT with the ability to mediate electrons and act as an immobilization matrix prompted us to investigate the aptasensing ability of the rGO-CNT nanocomposite. Hence, the present work focuses on the synthesis and development of hybrid rGO-CNT nanocomposite and amino-modified aptamer as a sensitive electrochemical sensing platform for detecting *S. Typhimurium*, a major foodborne disease-causing pathogen.

The rGO-CNT aptasensor was designed using the hybridization technique between rGO and CNT without any chemical reduction method and dropped-cast onto glassy carbon electrode (GCE). In this study, amino-modified aptamer was used as a linker to ensure the binding between rGO-CNT-aptamer-bacteria. Under the optimal conditions, the aptasensor

could detect the *S. Typhimurium* via differential pulse voltammetry (DPV) technique. Furthermore, the aptasensor was also applied to detect the foodborne pathogen in real chicken samples. The developed aptasensor has several distinct advantages including high sensitivity, good selectivity and low cost. This study is the first to report the application of ssDNA/rGO-CNT/GCE as an aptasensor for detecting *Salmonella*.

2 Experimental

2.1 Materials and reagents

Graphite powder, Phosphoric acid (85 %), Nitric acid (65 %) Hydrochloric acid (37 %), Sulfuric acid (98 %), Potassium permanganate, Potassium ferricyanide, Hydrogen peroxide (30 %), 10 X Phosphate Buffer Saline (PBS), Potassium chloride and CNT were purchased in analytical grade from Sigma-Aldrich. The *Salmonella enterica* serovars that were used included *S. Typhimurium* (model organism), *S. Enteritidis*, *S. Albany*, *S. Weltevreden*, *S. Typhi*, and *S. Derby* while non-*Salmonella* bacterial species were represented by *Escherichia coli*, *Klebsiella pneumoniae*, *Pseudomonas aeruginosa*, *Staphylococcus aureus* and *Enterococcus faecalis*. All the bacterial strains used in this work were from the Biomedical Science Laboratory, University of Malaya, Kuala Lumpur, Malaysia. DNA sequence of the aptamer (recognition biomolecule) was as previously published [28].

2.2 Instrumentation

The functional groups of the GO, rGO, CNT and rGO-CNT were determined by Fourier transform infrared (FT-IR) (Bruker IFS 66V/S, USA), whereas the Raman analysis was done through Renishaw inVia Raman microscope, Gloucestershire, UK. The X-ray diffraction (XRD) study was conducted using X-ray diffractometer (Bruker d8 ADVANCE). The morphology of the prepared nanocomposite was characterized using a transmission electron microscope (TEM, LEO LIBRA 120, Carl Zeiss). The electrochemical experiments were

performed using PGSTAT302N (Metrohm AG, Switzerland), an electrochemical workstation. Three-electrode system was used along with GCE, the working electrode, a platinum wire, the counter electrode and silver/silver chloride (Ag/AgCl), the reference electrode. Exfoliation of GO was conducted using Ultrasonic bath sonicator. Centrifugation (HITACHI CR22N, Japan) and homogenization (IKA RW20 digital Homogenizer) methods were carried out to produce GO in the form of a slurry and the whole process was conducted in a fumehood (LABCONCO Protector Laboratory Hood). All bacterial culture work was carried out in a Biosafety cabinet (ESCO Streamline SC2-6A1).

2.3 Preparation of rGO-CNT

Graphene oxide (GO) was synthesized through a modified Hummers procedure [29,11]. In order to obtain a purified carbon nanotube (CNT), the CNT powder that was impurified (100 mg) was left to heat at 480 °C in the atmospheric pressure and followed by a 12 h of reflux using 3 mol L⁻¹ HNO₃/HCl solution to eliminate metal oxides [30]. The treated CNT was then washed and filtered a few times using deionized water until a neutral pH was measured. Reduced graphene oxide (rGO) was then prepared using the GO. In a typical preparation process, 30 mg of GO was suspended in 15 mL ethylene glycol and was sonicated for 1 h till it became a brown colloidal solution.

Then, the solution was sealed using Teflon-lined autoclave and was heated in an oven for 18 h at 180 °C. The solids were washed with acetone and followed by distilled water to remove the additional GO. Finally, the rGO were dried in an oven (65 °C, 24 h). The rGO-CNT composites were then prepared by a hydrothermal method. In a typical procedure, a mixture of 10 mg of rGO and 10 mg of CNT was dispersed in distilled water and sonicated with a probe (Ultrasonic processor FS-250N, Shanghai, China). After a homogeneous black-brown dispersion was obtained, the mixture was shifted into a Teflon-lined autoclave, then

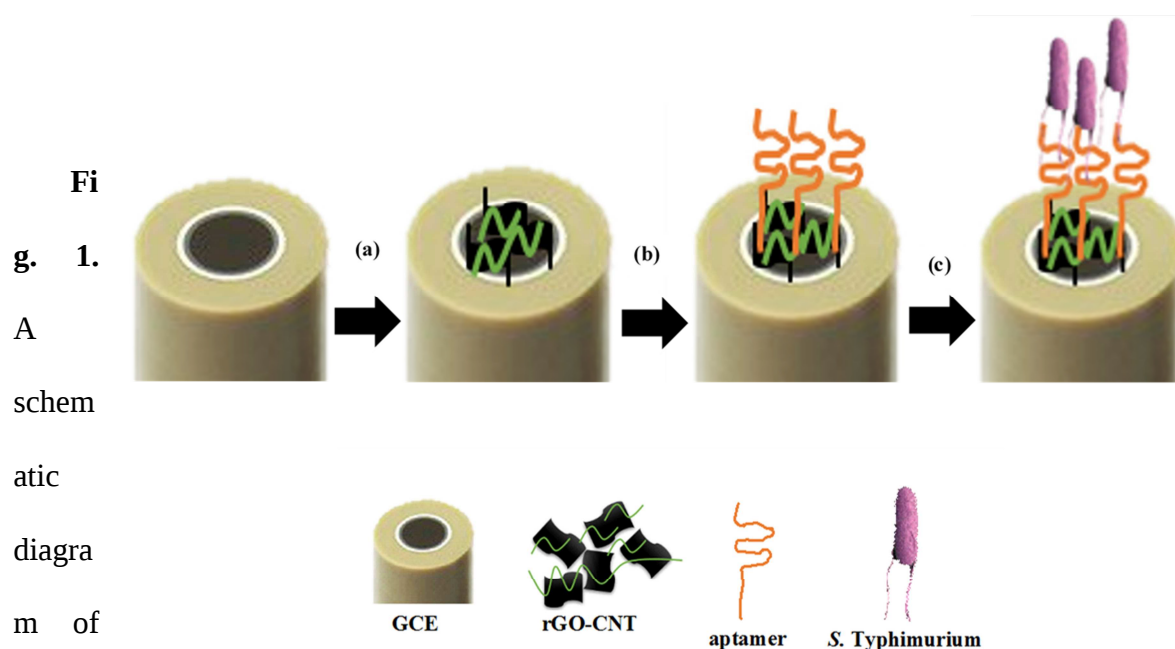
heated in an oven (180 °C, 24 h). Once it had cooled to room temperature, the resulting suspension was centrifuged to collect the precipitate. Finally, the composites were further washed 5 times with acetone and distilled water to remove the excess rGO and CNT.

2.4 Bacterial cultures

S. Typhimurium strain was cultured in Luria-Bertani broth for 24 hr with constant shaking. Bacterial cells from the overnight bacterial culture were harvested through centrifugation step at 5000 rpm for approximately 10 min. The cell pellets were washed once with 1X PBS (0.1M, pH 7.4) to remove traces of growth media. The bacterial cells were then resuspended in 1 mL of 1X PBS solution and diluted 10-folds to reach concentrations of 10^1 to 10^8 cfu mL⁻¹. The cell density of each dilution was then confirmed through conventional culture plating method. The procedure was repeated for *S. Albany*, *S. Enteritidis*, *S. Weltevreden*, *S. Typhi*, *S. Derby*, *E. coli*, *K. pneumoniae*, *P. aeruginosa*, *S. aureus* and *E. faecalis*.

2.5 Modification of the ssDNA/rGO-CNT/GCE aptasensor

GCE was initially polished using 0.05 µm-sized alumina fine powder to remove impurities from the electrode surface, and the electrode was then ultrasonicated, rinsed with distilled water and then left to dry at room temperature. For the modification of ssDNA/rGO-CNT/GCE aptasensor, firstly, 10 µL of the rGO-CNT suspension was dropped-cast on GCE surface and was left to dry under ambient conditions. Then, 5 µL of 5 µmol L⁻¹ of the amino-modified *Salmonella* aptamer was then dropped on the surface of the rGO-CNT/GCE and air-dried at room temperature to obtain ssDNA/rGO-CNT/GCE aptasensor. The fabricated electrode was then stored at 4 °C until future use. The overall modification process is schematically shown in Fig. 1.



the modification process of GCE. a) drop casting of rGO-CNT on GCE, b) rGO-CNT/GCE modified with amino-aptamer, c) ssDNA/rGO-CNT/GCE incubation with *S. Typhimurium*.

2.6 Analysis of the working electrodes through electrochemical method

CV technique was applied to determine the electrical conductivity of bare GCE, rGO/GCE, rGO-CNT/GCE and ssDNA/rGO-CNT/GCE using Zobel's solution (0.1 mol L^{-1} of KCl solution containing 3 mmol L^{-1} of $\text{K}_3[\text{Fe}(\text{CN})_6]$) at 100 mVs^{-1} scan rate. To optimize the condition of the sensing platform, the effects of rGO-CNT with concentrations ranging from 1 to 11 mg mL^{-1} and the aptamer concentrations (1 to $20 \text{ } \mu\text{mol L}^{-1}$) were also studied using CV.

To quantitate the presence of *S. Typhimurium*, the aptasensor was incubated with different concentrations of bacterial cultures ranging from 10^1 until 10^8 cfu mL^{-1} for approximately 5 min in PBS solution. Then, the corresponding DPV was carried out in Zobel's solution (electrolyte) at 6 mVs^{-1} scan rate. For the comparison studies, the DPV of ssDNA/rGO-CNT/GCE and rGO-CNT/GCE without any bacteria incubation were tested. To determine the

selectivity and specificity of the aptasensor, another five serovars of *Salmonella enterica* (*S. Albany*, *S. Enteritidis*, *S. Weltevreden*, *S. Typhi* and *S. Derby*) and five non-*Salmonella* bacteria such as *E. coli*, *K. pneumoniae*, *P. aeruginosa*, *S. aureus* and *E. faecalis* were cultured as previously mentioned. A standard concentration of 10^2 cfu mL⁻¹ of each culture was used for the test. All the electrochemical measurements were done in a clean environment in the Biosafety cabinet (ESCO Streamline SC2-6A1).

2.7 Aptasensor evaluation using food samples

The developed aptasensor was further evaluated using raw chicken meat sample ($n = 5$) purchased from a supermarket. Approximately 10 g of chicken meat was homogenized using 90 mL of buffer peptone water (BPW) for 3 hr at 37 °C. The pre-enriched food sample homogenate was then centrifuged, and the remaining cell pellet was resuspended in the PBS buffer. The cell suspension was serially diluted 10-folds. The colony forming units of the diluted bacterial suspension was determined through viable plating method on BrillianceTM Agar for *Salmonella* isolation (Oxoid). In parallel, an aliquot of 1 mL of pre-enriched food sample homogenate (in PBS buffer) was used to evaluate the working of the aptasensor. The food homogenate was left at room temperature for 5 min prior to electrochemical detection. For each food sample, three consecutive tests were carried out to confirm the reproducibility of the results.

3 Results and discussion

3.1 Surface characterization of rGO and rGO-CNT

The interaction between rGO and CNT was studied using FT-IR spectra. Fig. 2(a) shows the FT-IR spectra of GO, rGO, CNT and rGO-CNT. Based on FT-IR spectrum of GO, several peaks that corresponded to C=O stretching at 1724 cm⁻¹, C–O–C stretching at 1391 cm⁻¹, and C–O stretching at 1082 cm⁻¹ were reduced or absent in the rGO [1]. This

observation clearly indicated that the GO has been partially reduced by the decomposition method. Due to the partially reduced GO, the peak corresponding to the -OH functional groups remained in the rGO which help to interact with CNT. In rGO spectrum, the broad band at $\sim 3253\text{ cm}^{-1}$ corresponded to the hydroxyl group (-OH). Interestingly, this band was slightly shifted to $\sim 3204\text{ cm}^{-1}$ and was shrunken after the functionalization of CNT. There are three possibilities the existence of hydroxyl group (-OH). Firstly, the initial GO already contain huge amount of -OH groups shown by the shoulder at around 3200 cm^{-1} appears in the green spectrum. So, the reduction using the ethylene glycol (EG) only reduces carboxylic acid groups and not hydroxyl group. This was supported by the disappearance of 1724 cm^{-1} signal. Secondly, the use of EG under thermal condition only partially reduce the GO, not as good as strong acids. Finally, EG substance is quite hygroscopic and able to hold water. The presence of the broad signal maybe also due to the wet condition and trace amount of EG in the rGO sample.

In the FT-IR spectrum of rGO, the band at 1631 cm^{-1} can be attributed to the C=C stretching which is present due to sp^2 hybridization being shifted to 1557 cm^{-1} in rGO-CNT. Moreover, the peak at 2340 cm^{-1} which is found in rGO-CNT can be related to the presence of carbon dioxide (CO_2). CO_2 cannot be eliminated as it can be presence in all samples. This was also supported by the missing 1724 cm^{-1} and 3400 cm^{-1} signals. The C-O-C stretching vibrations in rGO-CNT are shown in the two absorption peaks at about 1323 cm^{-1} and 1027 cm^{-1} . Since this is partial reduction, C-O-C is possible especially condensation of adjacent -OH groups. All these FT-IR bands present in rGO-CNT provided direct evidence of a successful immobilization of rGO and CNT.

The results of the Raman spectroscopy further elucidate the vibration bands of the rGO and rGO-CNT nanocomposite. Fig. 2(b) shows Raman spectra of the GO, rGO, CNT

and rGO-CNT nanocomposite. The spectrum of rGO showed two significant vibrations at 1310 cm^{-1} (D band) and 1600 cm^{-1} (G band) representing the existence of structural defects in sp^2 -hybridized carbon system and first-order scattering of E_{2g} phonons of sp^2 carbon atoms. For rGO-CNT nanocomposite, both D and G bands are shown at 1349 and 1600 cm^{-1} correspondingly and two other peaks of 2D (2690 cm^{-1}) and D+G (2941 cm^{-1}) designated successful incorporation of rGO and CNT. The formation of rGO-CNT nanocomposite increases the peak intensity of D and G bands as compared to rGO peak intensities. However, the individual CNT peak showed the respective D and G bands at higher peak intensities as compared to rGO and rGO-CNT. These differences are due to the decrease of in-plane sp^2 domain sizes resulting from the introduction of oxygen-containing groups and increased defects in its graphitic domains [31].

To further support the FT-IR and Raman studies, the high angle XRD analyses for GO, rGO, CNT and rGO-CNT are depicted in Fig. 2(c). The observed amorphous peak at 2θ of 23.7° corresponds to the (002) reflection plane of rGO which is consistent with the values reported in the literature [32]. The XRD pattern of rGO-CNT nanocomposite has more than one peak which confirms the interaction between rGO and CNT forming a rGO-CNT nanocomposite. The XRD pattern of rGO-CNT had a similar diffracted pattern peak (002) with CNT and rGO sheet. Nevertheless, this peak slightly shifted and appeared as an intense broad peak at 25.4° . In addition, the peak defining in-plane regularity (100) appeared at $2\theta = 40.4$ shows that the CNT has been successfully incorporated with rGO. There are other small peaks which presence are due to the catalyst that was present during CNT preparation. However, these impurities did not negate the performance of the aptasensor.

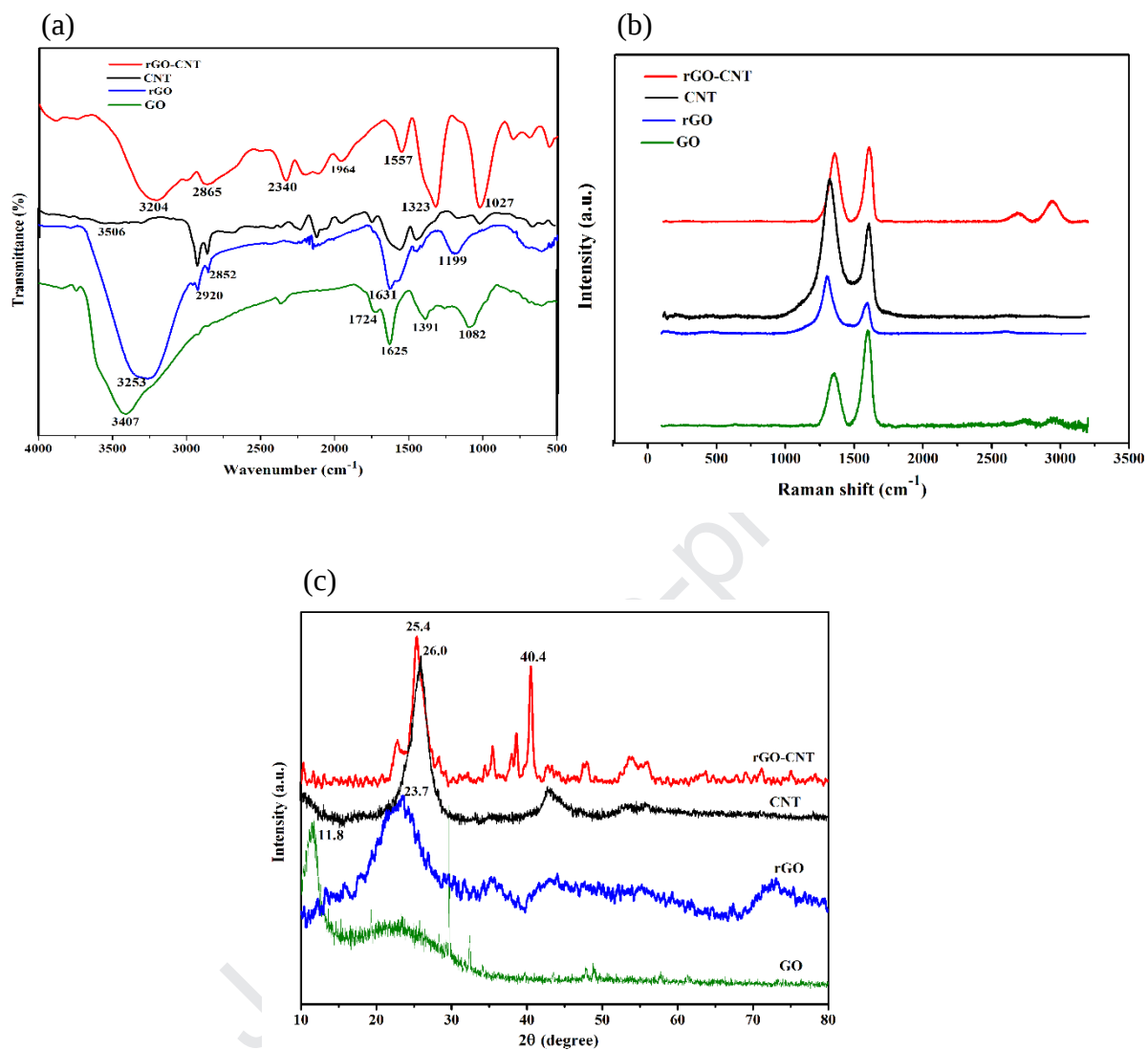


Fig. 2. The (a) FT-IR spectra of GO, rGO, CNT and rGO-CNT, (b) Raman spectra of GO, rGO, CNT and rGO-CNT and (c) High angle XRD pattern of the GO, rGO, CNT and rGO-CNT nanocomposites.

The TEM images of GO, rGO, CNT and rGO-CNT are presented in Fig.3a, 3b, 3c and 3d, respectively. The presence of thin multilayers of graphene oxide nanosheet is clearly demonstrated in Fig.3a. This result further supports the Raman data. As can be seen from Fig.

3b, the surface of rGO is flat, entangled and almost transparent morphology. This morphology is due to the ultrathin character of the rGO nanosheets. Fig. 3c shows the morphology of carbon nanotubes. This CNT consists of multiwalled nanotubes with hollow internal channel at the tip. In Fig. 3d, rGO-CNT presents a few-layered graphene sheet, coupled with a carbon nanotube. The average diameter of carbon nanotube was 14 nm. The mixture of rGO and CNT morphology confirmed the hybridization of the nanocomposite.

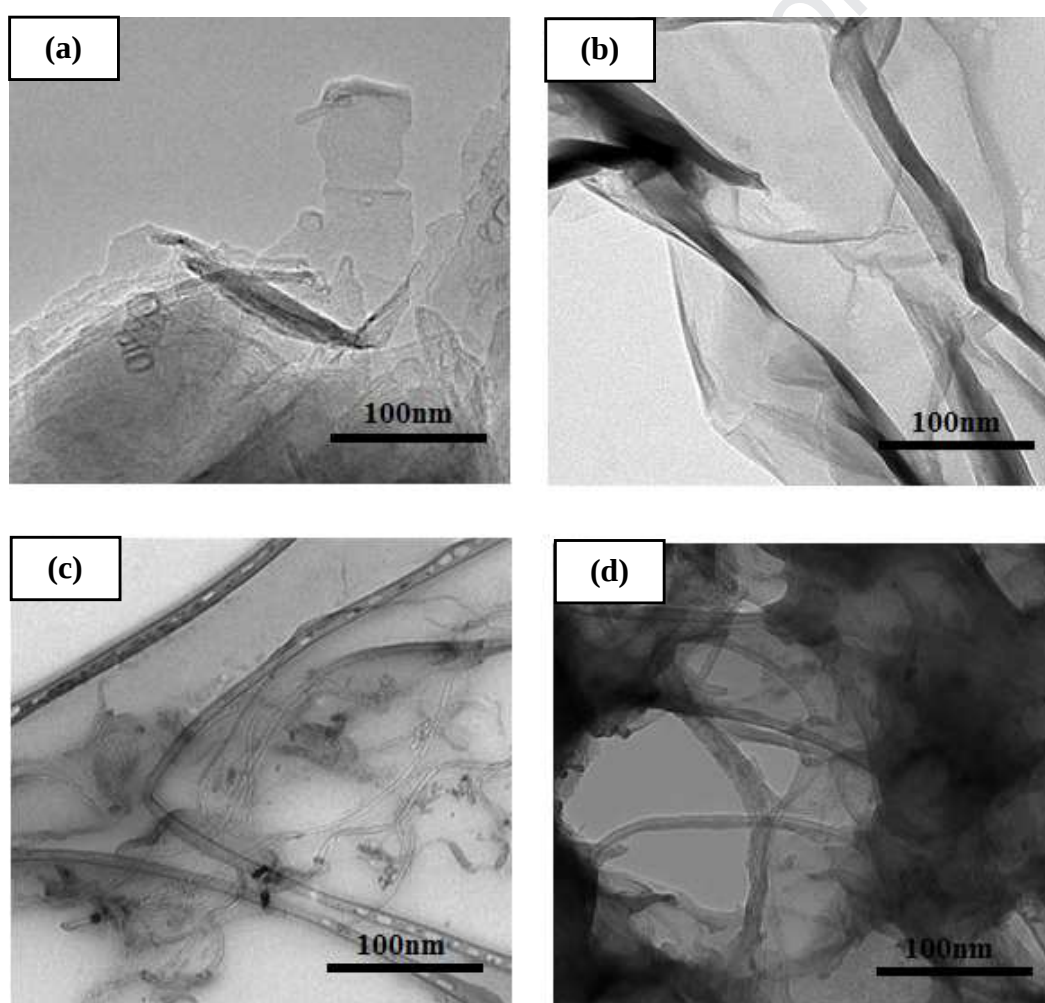


Fig. 3. The TEM images of the synthesized (a) GO, (b) rGO, (c) CNT and (d) rGO-CNT.

3.2 Electrochemical characterization of ssDNA/rGO-CNT/GCE bioelectrode

3.2.1 Optimization of working electrodes

Two main parameters have been optimized in this study to achieve the best sensing performance of our developed electrode. Firstly, the concentration effect of rGO-CNT on the electrodes was investigated as shown in Figure 4(a). The current density increased with the increasing concentration from 1 to 11 mg mL⁻¹ and reached the maximum current density at 5 mg mL⁻¹ of rGO-CNT. The current density decreased drastically when the loading amount exceeded 5 mg mL⁻¹. This phenomenon was due to the development of thick layers on the surface of electrodes which block the electron from reaching the electrode and thus lowering its sensitivity. Lastly, the concentration of rGO-CNT was optimized to 5 mg mL⁻¹.

The effect of aptamer concentration loaded on the rGO-CNT electrode was also optimized as shown in Figure 4(b). As the aptamer concentration increased from 1 to 20 μmol L⁻¹, the current density also increased. However, with further increase in aptamer concentration, the increase in current density was almost constant because the aptamer fixed on the electrode has reached a saturation point. Therefore, 5 μmol L⁻¹ of aptamer concentration was selected for subsequent experiments.

The electrochemical characterization of the developed electrode using Cyclic voltammetry in Zobel's solution at 100 mVs⁻¹ scan rate is shown in Figure 4(c). A pair of redox peaks was detected for bare GCE electrode. A slight increase in redox peak current, upon the coating of rGO on the electrode's surface was observed. Subsequently, the redox peak current for rGO-CNT/GCE is higher than rGO/GCE and bare GCE. This can be attributed to the uniform attachment of CNT on rGO surface which provides a high surface area and excellent electron transfer which enhances the overall conductivity [33].

After immobilization of aptamer, the peak current increased significantly due to the interactions of negatively charged DNA on the electrode surface contributes to ionic conductance and increases the overall charge transfer [34].

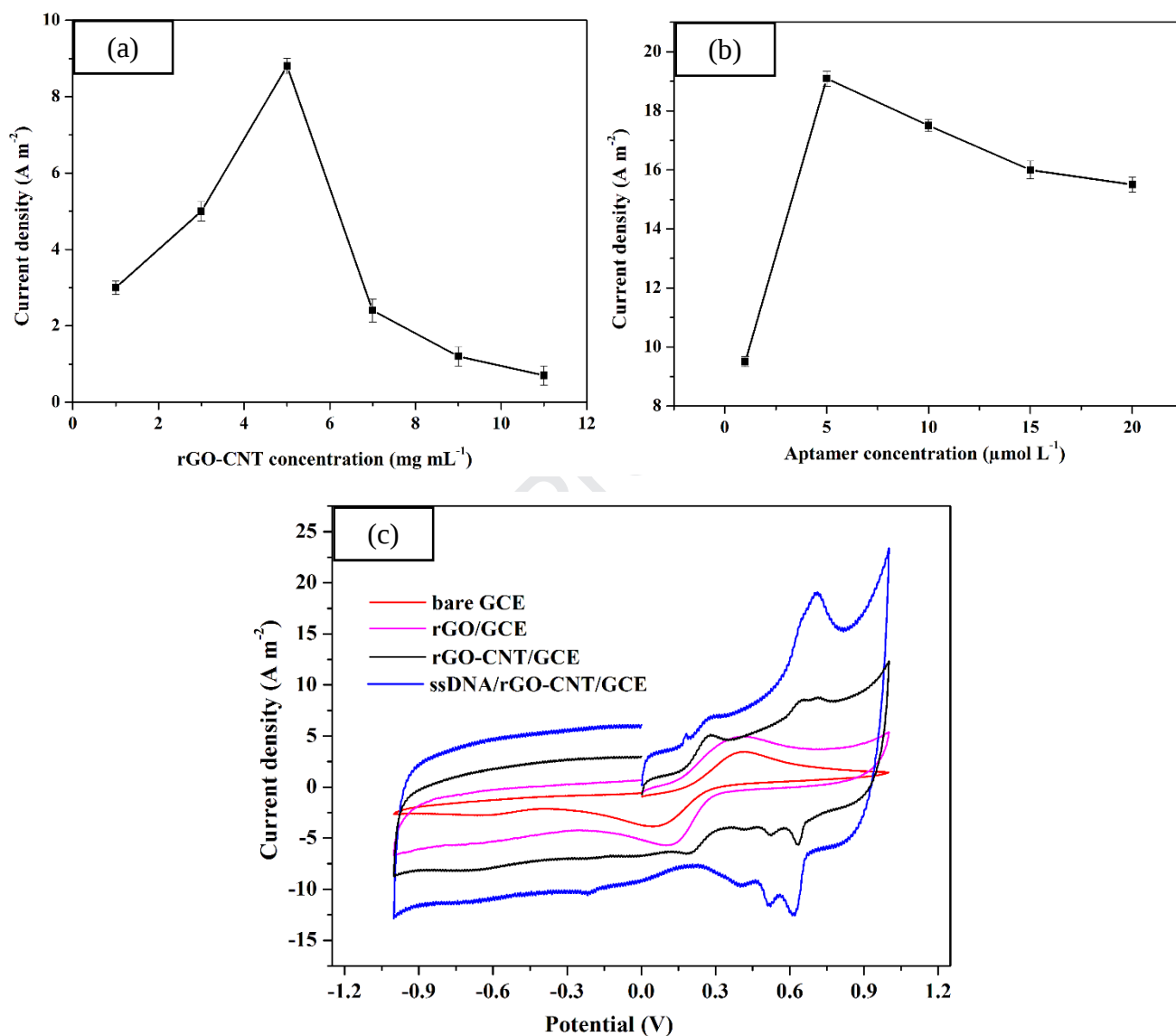


Fig.4. The effects of (a) rGO-CNT concentrations (mg mL⁻¹), (b) aptamer concentrations (μmol L⁻¹) and (c) electrochemical properties of the fabricated electrodes. The CV of bare GCE, rGO/GCE, rGO-CNT/GCE, and ssDNA/rGO-CNT/GCE, in Zobel solution (pH 7.0) at 100 mVs⁻¹ scan rate.

3.3 Sensitivity test

The electrochemical detection ability of the rGO-CNT nanocomposite was studied by exposing the aptamer-modified ssDNA/rGO-CNT/GCE electrode (aptasensor) to the bacterial concentration (10^1 to 10^8 cfu mL⁻¹). Fig. 5(A) shows that the current density increases with increasing concentrations of *Salmonella* cells. There is a notable change in the electrochemical behaviour of each assembly step in the aptasensor preparation that could be observed from the DPV response at 100 mVs⁻¹ scan rate. A prominent oxidation peak for the rGO-CNT/GCE can be observed at the potential of 0.23 V indicating excellent electron transferability and good electrical conductivity of rGO-CNT nanocomposite. After immobilization of aptamer, a large increase in oxidation current density of ssDNA/rGO-CNT/GCE is visible in the voltammogram. This is due to the increase in charge transfer as a result of the ionic conductance and π - π DNA interaction which proves the binding of aptamer on the rGO-CNT surface with high affinity [35]. In the presence of bacteria, the oxidation peak current increased proportionally to the concentration of bacteria cells. Nevertheless, the current decreased prominently as compared to ssDNA/rGO-CNT/GCE after binding of bacterial onto the electrode's surface. This is ascribed to the negatively charged cell membrane that creates hindrances and blocks the electron transfer to the electrode surface [36]. When the bacterial concentration increased, the current density also increased due to the kinetics of interfacial electron-transfer and increase in the electron-transfer resistance from the medium to the electrode's surface. This causes an electro-permeation of conductive ions from inside the bacterial cells into the electrolyte [37]. More bacteria cells cause leakage of more conductive ions which increases the current density.

The plot of current density vs. the logarithm of the concentration of bacteria (Fig. 5B) showed a linear relationship with a correlation coefficient of 0.9855. The limit of detection of this sensor was 10^1 cfu mL⁻¹.

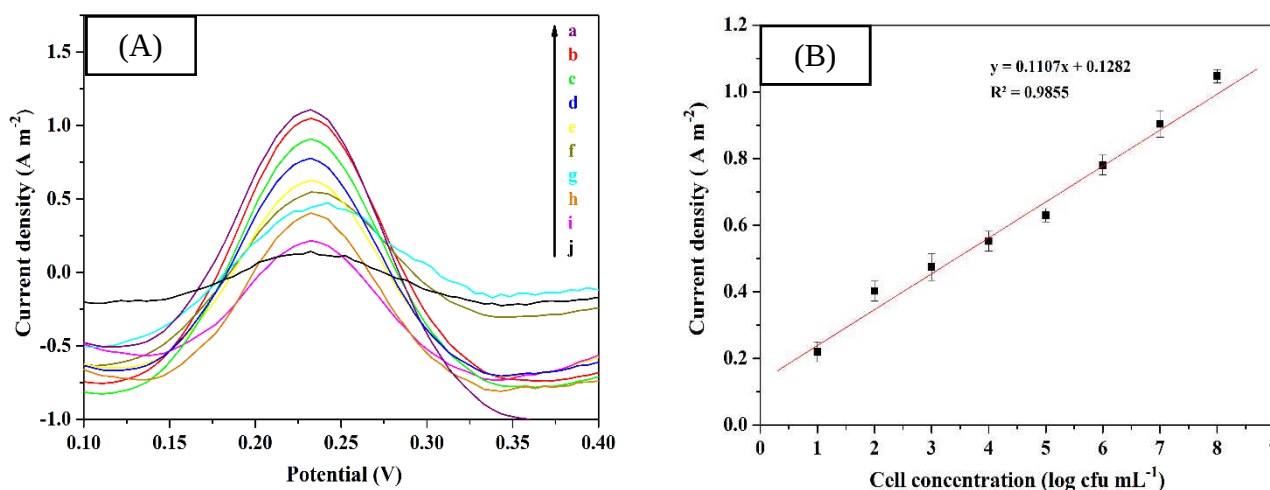


Fig.5. Evaluation of sensitivity of aptasensor on a different *S. Typhimurium* concentration. (A) DPV of (a) ssDNA/rGO-CNT/GCE, (b) 10^8 cfu mL⁻¹, (c) 10^7 cfu mL⁻¹, (d) 10^6 cfu mL⁻¹, (e) 10^5 cfu mL⁻¹, (f) 10^4 cfu mL⁻¹, (g) 10^3 cfu mL⁻¹, (h) 10^2 cfu mL⁻¹, (i) 10^1 cfu mL⁻¹ and (j) rGO-CNT/GCE in 0.1 mol L⁻¹ of KCl solution containing 3 mmol L⁻¹ K₃[Fe(CN)₆] at pulse height 50 mV, pulse width 70 ms. (B) The graph depicts linear relationship between current density and cell concentration (logarithm).

3.4 Selectivity and specificity tests

The selectivity and specificity tests were studied using bacteria concentrations of 10^2 cfu mL⁻¹. Fig. 6(a) shows the selectivity test of the aptamer against other serovars of *Salmonella enterica* such as *S. Albany*, *S. Enteritidis*, *S. Weltevreden*, *S. Typhi*, and *S. Derby*. All the bacteria exhibited peak currents like *S. Typhimurium* which indicated the aptamer was specific for all tested *Salmonella* serovars. Nevertheless, when the aptasensor was tested against non-*Salmonella* bacteria (*E. coli*, *K. pneumoniae*, *P. aeruginosa*, *S. aureus* and *Enterococcus faecalis*), the current density was lower than that of the limit of detection (LOD) of the electrode (Fig. 6(b)). This indicates non-specificity of the aptamer to non-

Salmonella bacterial cells. The intermolecular folding of aptamers enables them to identify and bind to their targets with good affinity and specificity discriminating the binding with non-*Salmonella* bacteria.

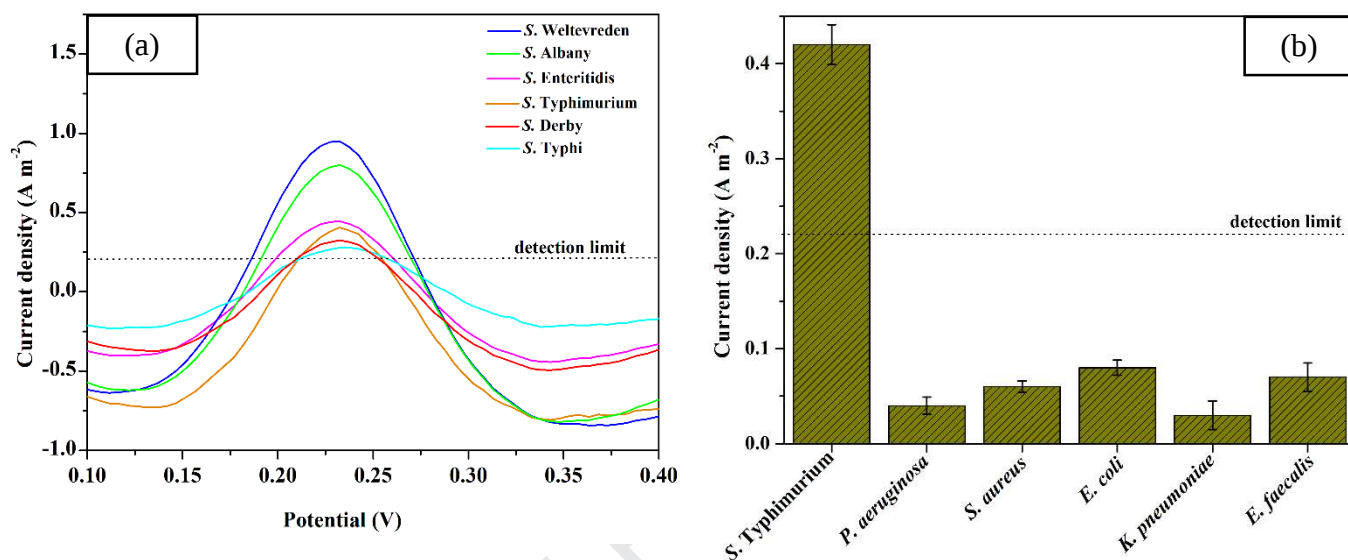


Fig.6. Selectivity and specificity evaluation of the aptasensor in detecting different bacterial strains. (a) DPV of *S. Weltevreden*, *S. Albany*, *S. Enteritidis*, *S. Typhimurium*, *S. Derby* and *S. Typhi* in 0.1 mol L^{-1} KCl solution with 3 mmol L^{-1} of $\text{K}_3[\text{Fe}(\text{CN})_6]$ at pulse height of 50 mV, pulse width of 70 ms. (b) Specificity of aptasensor in various samples. Data are expressed as the mean $\pm$ standard deviation ($n = 3$).

The performance of the developed rGO-CNT aptasensor for bacteria detection was compared to previous studies and the important characters are summarized in Table 1. The obtained LOD was much lower using the prepared biosensor in this study compared to those published data. The comparison table shows that the developed rGO-CNT aptasensor exhibited high sensitivity with improved detection limit. Hence this provides an alternative approach for rapid detection of *S. Typhimurium*.

Table 1. A comparison of different methods of bacterial detection with the current study.

Materials	Targeted bacteria	Detection methods	LOD (cfu mL ⁻¹)	Time	Ref
rGO-MWCNT	<i>Salmonella</i>	EIS	25	1 h	[38]
CNT	<i>S. aureus</i>	Potentiometry	8 x 10 ²	Not stated	[39]
rGO-Au	<i>S. aureus</i>	EIS	10	Not stated	[40]
-	<i>Salmonella</i> spp.	Multiplex-PCR	10 ³	3 h	[41]
Gold Nanoparticle-Based Enzyme-Linked Antibody	<i>S. Typhimurium</i>	Immunological assay	10 ³	2 h	[42]
Three-Dimensional Interdigitated Electrode Array (3D-IDEA)	<i>E. coli</i> O157:H7	EIS	2.9 x10 ²	30 min	[43]
rGO-CNT	<i>S. Typhimurium</i>	DPV	10 ¹	10 min	This work

3.5 Reproducibility and stability properties of the aptasensor

Under the optimal conditions, the electrochemical reproducibility and stability of rGO-CNT nanocomposite (data not shown) were investigated by repetitive cycling procedure (21 cycles), which was observed to produce changes that were insignificant for the peak current density. Also, ssDNA/rGO-CNT/GCE was stored for 20 days in ultrapure water at 4 °C and it showed good to excellent stability as an aptasensor. The slight decrease in signal was noted at day 20 and therefore the aptasensor has good storage stability.

3.6 Food samples evaluation

Fig. 7 shows the response of the aptasensor with five chicken meat samples. Out of 5 chicken meat samples, 3 chicken samples (CS1, CS2, CS3) exhibited a positive response to the presence of different concentration of *Salmonella* ranging from 10^1 to 10^3 cfu mL⁻¹ while another two samples (CS 4 and CS5) gave a response below the detection limit. When the same food samples were tested using the conventional culture method, presumptive *Salmonella* isolates were detected on Brilliance™ *Salmonella* agar (Oxoid) for samples CS1, CS2 and CS3 only (Data not shown). Hence the data obtained by the aptasensor was validated by the culture method. The results obtained in this food sample analysis were comparable to the standard graph obtained in Fig. 5(B). This aptasensor showed good recovery illustrating its applicability for food sample analysis.

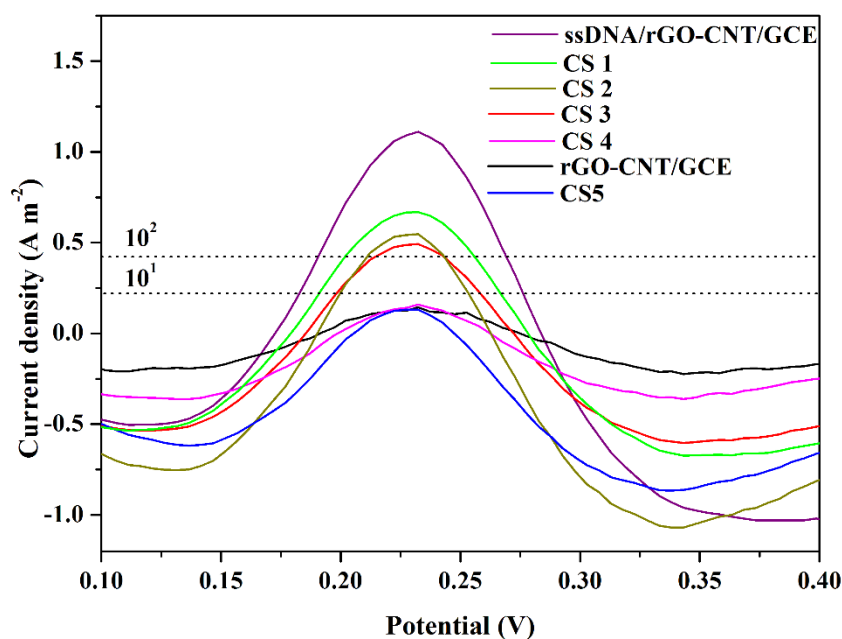


Fig.7. DPV comparison between bacterial detection in five different chicken samples with rGO-CNT/GCE and ssDNA/rGO-CNT/GCE. Food samples: CS1 – CS5

4 Conclusion

In summary, an electrochemical aptasensor rGO-CNT nanocomposite for the detection of *Salmonella* was successfully developed. The nanocomposite has been characterized by physicochemical methods to show the hybridization of rGO and CNT. The incorporated amino-modified DNA aptamer to the rGO-CNT successfully detected whole *Salmonella* bacterial cells without any pre-treatment or DNA extraction steps. The complete working electrode ssDNA/rGO-CNT/GCE demonstrated significant current peaks for various concentrations of *S. Typhimurium* ranging from 10^1 until 10^8 cfu mL⁻¹ with a LOD value of 10^1 cfu mL⁻¹. More remarkably, the detection of *Salmonella* was able to be completed in 10 min. Furthermore, the prepared aptasensor successfully distinguished *Salmonella* from non-*Salmonella* cells. The overall results indicated that the rGO-CNT nanocomposite exhibited good electrochemical activity attributed to the presence of reduced graphene oxide which synergistically improves electron transfer. Moreover, the carbon nanotube has acted as the spacer and provided a large specific surface area for aptamer immobilization. Thus, the designed ssDNA/rGO-CNT/GCE showed high selectivity, sensitivity and specificity, cost-effective approach for foodborne pathogen analysis. Nevertheless, there are some limitation which must be considered for the development of the aptasensor such as pH, ionic strength, viscosity and temperature. Thus, in future study, more focus will be given on the design of aptasensors against unexplored target analyte with different types of aptamer.

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Compliance with Ethical Standards:

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Conflict of Interest: Jimmy Nelson Appaturi, Kwai Lin Thong, Thiruchelvi Pulingam, Shalini Muniandy, Noraini Binti Ahmad and Bey Fen Leo declare that there are no conflict of interest.

Ethical approval: This article does not contain information from any studies that were carried out with human or animals.

Informed consent: No informed consent was obtained as human participants were not involved in this work.

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Research Highlights

- rGO-CNT nanocomposite was developed for detection of *Salmonella*.
- rGO-CNT nanocomposite demonstrating the synergistic effect.
- Detection limit of 10^1 cfu mL⁻¹ for *S. Typhimurium*.
- A sensitive, selective and specific aptasensor with a wide detection range was produced.
- The ssDNA/rGO-CNT/GCE was successfully applied to real food samples.