



A novel fluorescence aptasensor based on mesoporous silica nanoparticles for selective and sensitive detection of aflatoxin B₁

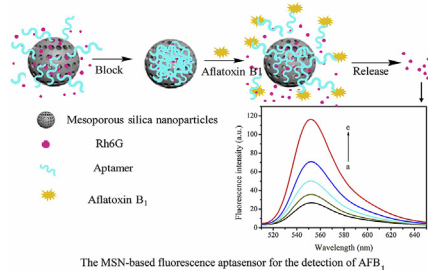
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HIGHLIGHTS

- High amino-modified concentration of mesoporous silica nanomaterials (MSN-NH₂) was synthesized.
- MSN-NH₂ loaded with label-free signal probe (Rh6G) could be regulated by gated molecular (AFB₁-aptamers).
- This aptasensor is designed by MSN-NH₂, Rh6G and AFB₁-aptamers to detect AFB₁ with the detection limit of 0.13 ng mL⁻¹.
- This design could be applied to other mycotoxins and contaminants if appropriate aptamers are available.

GRAPHICAL ABSTRACT



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ABSTRACT

Based on the mesoporous silica nanoparticles (MSN), a novel, simple and label-free aptamer biosensor was designed for the detection of aflatoxin B₁ (AFB₁). Here, the aptamers were used as molecular recognition probes and “gated molecules” while Rh6G was loaded into the interior of the particles as the signal probe. In the absence of AFB₁, the “gate” was closed to prevent the leakage of the signal probe because of the immobilization of aptamers on the surface of MSN-NH₂. With the presence of AFB₁, the “gate” could be opened to release the signal probe for the specific binding of aptamers to AFB₁. Our results showed that the fluorescence intensity was positively correlated with the concentration of AFB₁ (0.5–50 ng mL⁻¹), with the detection limit as low as 0.13 ng mL⁻¹. What's more, this design provides a new approach for rapid, sensitive and selective detection based on aptamers and it could be applied to numerous other analytes if appropriate aptamers are available.

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1. Introduction

Mycotoxins, as the toxic and carcinogenic secondary metabolites produced by various molds, can cause serious problems to

human health by contaminating numerous foods and animal feeds [1,2]. In recent years, aflatoxins often have been found in agricultural products, raising many global concerns due to a series of economic losses and potential risk to foods and feeds [3]. Among several types of aflatoxins, AFB₁, as the most toxic carcinogen, has been classified as Group I carcinogen by International Agency for Research on Cancer (IARC) [4]. Generally, agricultural products, such as, peanuts, corn and grains are particularly vulnerable to AFB₁

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contamination [5]. The maximum contamination levels of AFB₁ (2 µg kg⁻¹) and total aflatoxins (4 µg kg⁻¹) in all cereal and cereal-based products have been determined by European Commission to safeguard consumers' health and avoid food safety risk [6]. Up to date, the main methods for the detection of AFB₁ are typically instrumental approaches, such as, high performance liquid chromatography (HPLC) [7–9], liquid chromatography combined with mass spectrometry (LC-MS) [10], as well as rapid detection technologies, like enzyme-linked immunosorbent assays (ELISA) [11,12] and so on. These methods can accurately detect the presence of AFB₁, however, tedious pretreatments, expensive equipment and professional operator are required [13,14]. Besides, antibody-based rapid analyses require expensive, laborious and time-consuming antibody preparation and also may be susceptible to the instability of the antibody. In this case, the design of a simple, rapid, sensitive and cost-effective method for quantifying AFB₁ has become a research hotspot.

Aptamer, a single-stranded DNA or RNA oligonucleotide sequence, has high affinity and specificity for targets [15]. Owing to their higher affinity, lower cost, easier modification and higher stability [16], aptamers are being explored as a highly promising alternative to antibody [17]. Besides, aptamers can be functionalized with all kinds of fluorophores [18], functional groups [19], biotin and enzymes [20], which could simplify the inspection procedures and make the sensing platform more flexible. Thus, aptamers have been adopted as novel molecular recognition probes for the detection of mycotoxins. Up to now, aptamer-based biosensors have been developed for the detection of several mycotoxins, such as AFB₁ [21–23], AFM₁ [24,25], ochratoxin A (OTA) [26], zearalenone (ZEN) [27,28] and so on.

Significantly, giving the credit to the ease of modification, good biocompatibility and uniform nanometer size of nanoparticles (magnetic nanoparticles, silica nanoparticles, graphene, and etc.), many sensor strategies based on nanoparticles have been established to detect various mycotoxins [29]. Mesoporous silica nanomaterials (MSN) have been widely used for their large surface area, good thermal stability, easy modification, good porosity, large load capacity, and excellent reactivity [30,31]. Chen et al. [32] took advantage of the porosity of the MSN as a carrier molecule in conjunction with chemiluminescence to determine cocaine. And also, S. Dehghani et al. [33] established the MSN-based sensor to detect kanamycin.

Based on these ideas, we report herein for the first time that an aptasensor based on MSN has been developed for detection of AFB₁ in this paper. MSN was firstly synthesized and then the MSN-based controlled release system was combined with the specific aptamers of AFB₁ to construct an aptasensor. Here, the aptamers are used as molecular recognition probes and “gated molecules”. The affinity of the specific binding of aptamers-AFB₁ is greater than the electrostatic force of aptamers and MSN, which could make the “gate” open to release the Rh6G. The quantitative detection of AFB₁ is achieved by the enhancement of fluorescence intensity. The whole sensing procedures are simple, rapid and highly specific to AFB₁, which offers a great potential for on-site analysis.

2. Experimental

2.1. Materials and apparatus

Aflatoxin B₁ (AFB₁), Aflatoxin B₂ (AFB₂) were purchased from Sigma (USA). Aflatoxin M₁ (AFM₁), Deoxynivalenol (DON), Zearalenone (ZEN), Ochratoxin (OTA), Fumotoxin B₁ (FB₁), T-2 were purchased from Pribolab (Singapore). Other chemical reagents such as cetyltrimethylammonium (CTAB), tetraethyloxysilane (TEOS), 3-amino-propyltriethoxysilane (APTES), Rhodamine 6G (Rh6G) were obtained from Aladdin Reagent Co. Ltd., (Shanghai, China). Disodium phosphate (Na₂HPO₄), Sodium dihydrogen phosphate (NaH₂PO₄) and Sodium hydroxide (NaOH) were obtained from Cologne Chemical Co. Ltd (Chengdu, China). The samples of peanut oil, corn oil, corn, peanut, oatmeal, rice were purchased from the local supermarket (Chongqing, China). The AFB₁ aptamers (Apt1, Apt2, Apt3, Apt4, Apt5) used in this work (Table 1) were synthesized and purified with HPLC by Shanghai Sangon Biotechnology Co. Ltd. (Shanghai, China). All other chemicals were at analytical grade and aqueous solutions were prepared with ultrapure water (18.25 MΩ·cm).

The transmission electron microscopy (TEM) images were obtained with JEM-1200EX (Japan). FTIR spectra of the nanoparticles were obtained with Spectrum 100 (PerkinElmer, USA). Ultraviolet–visible (Uv–vis) absorption spectra were collected by UV-2450 (Shimadzu, Japan). The zeta potential was recorded on ZEN3690 (Malvern, England). Fluorescent intensity measurements were performed by F02500 (Shimadzu, Japan). Accuracy of this method was verified by HPLC on LC-20A (Shimadzu, Japan).

2.2. Synthesis of MSN

In this study, MSN was synthesized with the reported procedures with slight modifications [39–46]. Briefly, MSN was synthesized under alkaline conditions using TEOS as a silica precursor, CTAB as a surfactant template, NaOH as a catalyst and water as a solvent. Firstly, 0.5 g of CTAB was initially dissolved in 240 mL ultrapure water with slowly stirring at 80 °C to get a clear and transparent solution. Afterwards, 1.75 mL of NaOH (2 M) and 2.5 mL of TEOS were added dropwisely to the mixture solution, followed by vigorously stirring for 2 h. The mixture was allowed to stand for cooling, and then centrifuged at 10000 rpm for 10 min. The resulting product was washed successively with ethanol and water three times to remove the remaining solvent from the pores. MSN containing the template CTAB was obtained by vacuum drying overnight. The above prepared MSN nanoparticles were calcined at 550 °C for 5.5 h to remove the excess surfactant CTAB.

2.3. Synthesis of amino-modified MSN (MSN-NH₂)

0.5 g of MSN was homogeneously dispersed in 50 mL of ethanol by ultrasonication. After that, 1.5 mL of APTES was added into the mixture solution with slowly stirring for 10 h at 110 °C. Then, the white solid was recovered via centrifugation at 8000 rpm for 10 min, followed by washing with ethanol and water three times to

Table 1
Aptamer used in this study.

NO	Sequence (5'-3')	Reference
Apt 1	AGTTGGGCACGTGTTGTCTCTGTGTCTCGTCCCTTCGCTAGGCCACA	[34]
Apt 2	GTT GGG CAC GTG TTG TCT CTC TGT GTC TCG TGC CCT TCG CTA GGC CC	[35]
Apt 3	CAG AGA GAC AAC ACG TGC CCA AC	[36]
Apt 4	ATA GGA GTC ACG ACG ACC AGA AAG TAA TGC CCG GTA GTT ATT CAA AGA TGA GTA GGA AAG ATA TGT GCG TCT ACC TCT TGA CTA	[37]
Apt 5	TTG GGC ACG TGT TGT CTC TCT GTG TCT CGT GCC CTT CGC TAG GCC CAC A	[38]

remove the excess APTES. Finally, MSN-NH₂ was obtained in a vacuum freeze-drying oven. The concentration of amino-modified on the MSN was calculated [32].

2.4. Preparation of the fluorescent aptasensor for AFB₁ detection

To fabricate the MSN-based aptasensor, 30 mg of MSN-NH₂ was incubated with 500 μ L of Rh6G solution (1 mM) with shaking for 14 h at 25 °C. After that, the MSN-NH₂/Rh6G complex was incubated with 200 μ L of a 100 μ M aptamer capture probe for 5 h at 25 °C. The complexes of aptamer/MSN-NH₂/Rh6G were formed, centrifuged and washed repeatedly with ultrapure water to remove excess aptamers and Rh6G dye. Finally, the above complexes were resuspended in phosphate saline (PBS, 0.2 M, pH 7.4) ($C_{\text{MSN-NH}_2}$ = 60 mg mL⁻¹) for further using.

The process of AFB₁ detection was carried out as follows. Firstly, 50 μ L of the above particle dispersion solution was added into 600 μ L of AFB₁ solution diluted in PBS. 600 μ L of PBS was added as a blank control. After fully shaking, the mixture solution was then incubated at 37 °C for 3 h. The corresponding fluorescent intensity at excitation/emission wavelength of 490/552 nm was recorded at room temperature (25 °C).

2.5. Sample analysis

In order to evaluate the recovery of the MSN-based fluorescence aptasensor, the samples of peanut oil, corn oil, corn, peanut, oatmeal and rice were obtained from supermarket. First of all, the samples were detected by HPLC to verify the presence of AFB₁. Afterwards, each of corn oil and corn samples was spiked with AFB₁ at the concentration of 1, 3, 5, ng mL⁻¹, respectively. Each spiked sample (5.00 ± 0.005 g) and 25 mL of the extraction solution (methanol:water = 7:3) were mixed to extract AFB₁. The corn, peanut, oatmeal and rice were vortexed for 20 min and then extracted using ultrasonic for 30 min. Subsequently, the solution was collected. The peanut oil, corn oil were vortexed for 20 min and then extracted using ultrasonic for 30 min. Then, the filtered solution was further extracted with 10 mL of n-hexane and 10 mL of 70% methanol in a separatory funnel. Finally, all extracts were collected, dried with nitrogen and diluted in PBS for AFB₁ detection.

In order to evaluate analytical reliability and applicability of the proposed MSN-based fluorescence aptasensor, the detection of AFB₁ in natural samples were carried out in accordance with the procedures above.

3. Results and discussion

3.1. The strategy of the proposed fluorescence aptasensor for AFB₁ detection

Schematic illustration for the design of MSN-based fluorescent aptasensor for AFB₁ detection was shown in Fig. 1A. MSN-NH₂ was selected as the carrier for the controlled release system and Rh6G was loaded into the interior of the particles as signal probe. As “gated molecule” and molecular recognition probes, the aptamers were immobilized on the surface of MSN-NH₂ particles based on the interaction of positively charged MSN-NH₂ with negatively charged aptamers. The aptasensor (aptamers-modified MSN-NH₂ loaded with Rh6G) was prepared for the detection of the target AFB₁. In the presence of AFB₁, the aptamers could be removed from MSN-NH₂ by the specific binding between aptamers and AFB₁, thus the “gate” was opened and signal probe (Rh6G) is released. In the case, the fluorescence intensity was positively correlated with the concentration of AFB₁ and the quantitative detection of AFB₁ could be finally achieved by the change of fluorescence intensity.

To confirmed the feasibility of the aptasensor, series concentration of AFB₁ were respectively added into the aptamer/MSN-NH₂/Rh6G complexes to obtain their corresponding fluorescent intensities. As shown in Fig. 1B, the fluorescence intensity was enhanced with the increasing concentration of AFB₁. Therefore, this principle could be applied to develop a sensitive MSN-based fluorescence aptasensor for the AFB₁ detection.

3.2. Characterization of the synthesized MSN-NH₂

The formation of mesopores mainly refers to the process of removing surfactants by high temperature treatment or other physicochemical methods [47]. As could be seen from Fig. 2A, strong absorbing peaks at 2921 cm⁻¹, 2851 cm⁻¹ (C–H stretching vibration) and 1478 cm⁻¹ (C–H deformation) were shown in line a, which could be caused by a large amount of surfactants residue in the porous pores. Besides, in line b, these peaks were found to be disappeared, suggesting that the mesopores were formed after the removal of surfactants [48]. Compared with MSN, MSN-NH₂ (line c) showed new peaks at 1636 cm⁻¹ and 1554 cm⁻¹, corresponding to the flexural stretching vibrations of amide I and –NH₂, respectively. After modification of the MSN with amino groups, vibrational peaks were displayed at 1340 cm⁻¹ and 1410 cm⁻¹, which could be attributed to the stretching band of the C–N group [32]. Furthermore, the Si–OH band at 962 cm⁻¹ in the MSN-NH₂ spectrum became weaker after amino modification. These results suggested

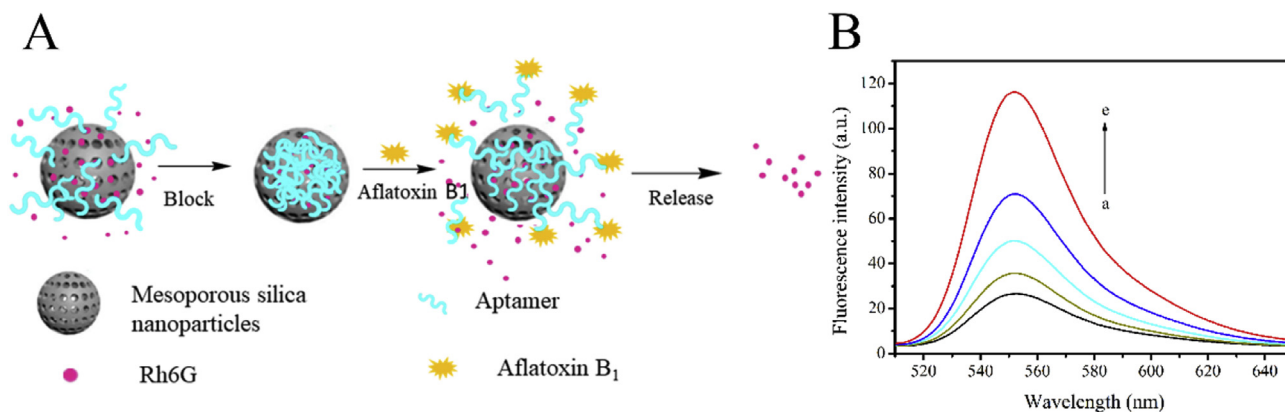


Fig. 1. (A) Principles of the MSN-based fluorescence aptasensor for the detection of AFB₁. (B) Fluorescence emission spectra of the MSN-based fluorescence aptasensor in the presence of AFB₁ (a: 0, b: 5 ng mL⁻¹, c: 10 ng mL⁻¹, d: 20 ng mL⁻¹, e: 40 ng mL⁻¹). Excitation and emission are at $\lambda_{\text{ex}}/\lambda_{\text{em}}$ = 490/552 nm.

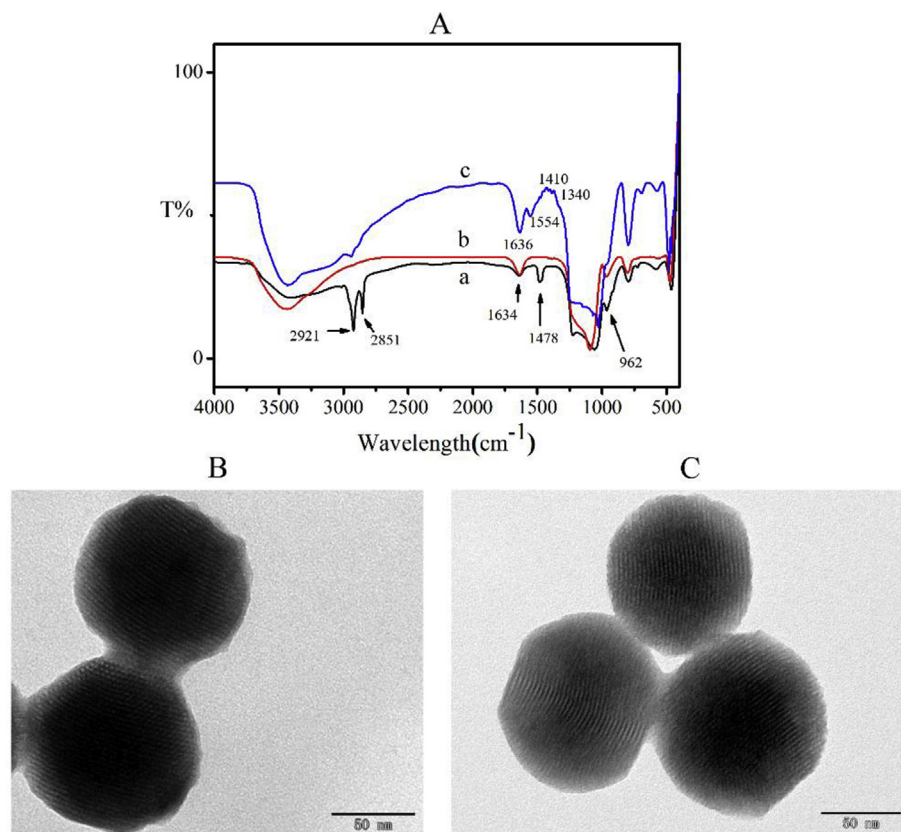


Fig. 2. (A) FTIR spectra of MSN with surfactant, MSN and MSN-NH₂ (from a to c). (B) TEM micrograph of MSN. (C) TEM micrograph of MSN-NH₂.

that amino groups were successfully modified onto MSN.

Moreover, the pores structures of MSN and MSN-NH₂ were further proved by TEM. Fig. 2B and C showed typical images of the synthesized MSN and MSN-NH₂, the average size of the as-synthesized nanoparticles was 100 nm. The TEM image showed that the MSN and MSN-NH₂ possessed excellent mesoporous channel structures. A large number of pores on the MSN-NH₂ indicated that the MSN-NH₂ nanoparticles had the potential applicability as a carrier to contain more fluorochrome.

The concentration of amino-modified on the MSN was 1.235 mmol g⁻¹, which was 10 times of other MSN-NH₂ nanoparticles in the reported [32]. The modification of amino groups to MSN and the binding of aptamers to MSN-NH₂ nanoparticles were verified by zeta potential and UV spectra. As shown in Fig. 3A, the zeta potential of MSN was increased from -32 mV to 20.8 mV after modification of the MSN with amino groups. The change of the zeta potential decreased from 20.8 mV to -34.3 mV after the binding of MSN-NH₂ with aptamers, which confirmed the conjugation between aptamers and MSN-NH₂.

Moreover, the UV-vis spectra showed that aptamers were successfully bound to the MSN-NH₂. As shown in Fig. 3B, the disappearance of the characteristic absorption peak at 260 nm could be attributed to the formation of aptamer/MSN-NH₂ complex. Besides, characteristic absorption peak at 260 nm was found in the aptamer + MSN-NH₂ spectrum (Fig. 3C), illustrating the successful binding of aptamers on MSN-NH₂ nanoparticles.

3.3. Screening out AFB₁-specific aptamer suitable for MSN-based aptasensor strategy

As a recognition molecular, specificity and affinity of aptamers

are the bases of the proposed aptasensor. A large number of aptamers have been selected for AFB₁ recognition by SELEX methods. Therefore, a total of five AFB₁ aptamers (Apt1, Apt2, Apt3, Apt4, Apt5) were selected to investigate its specificity and affinity in the proposed method. They are commonly used and reported in the previous literature. The specificity of this aptasensor for AFB₁ and its analogues, aflatoxin B₂ (AFB₂), OTA, ZEN and deoxynivalenol (DON) was evaluated using the proposed aptasensor strategy at the same test conditions (Fig. 4A). As shown in Fig. 4A, the specificity of aptamers in the proposed biosensor was evaluated with 30 ng mL⁻¹ of the mycotoxins. AFB₁ induced the highest fluorescence intensity with binding to Apt1 (the black column in Fig. 4A) while the other analogues (AFB₂, DON, ZEN or OTA) induced a low fluorescence intensity. This indicated that Apt1 can strongly bound to AFB₁ but only slightly bind with other mycotoxins. In particular, AFB₂ contains almost the same carbon skeleton as AFB₁, except for the double bond on the difuradin ring. However, the fluorescence intensity of AFB₁ was still up to fourteen-fold higher than that of AFB₂ when Apt1 (black column in Fig. 4A) was used as the recognition sequence. The other four aptamers tested (Apt2, Apt3, Apt4, Apt5) were also able to recognize AFB₁ (the other color columns except black), but showing much poorer specificity than Apt1. So Apt1 is the best one of the five aptamers to recognize AFB₁.

AFB₁ aptamers with different sequences (Apt1, Apt2, Apt3, Apt4, Apt5) were also investigated in affinity of aptamer recognition probe to target with proposed aptasensor. As shown in Fig. 4B, with the increasing of AFB₁ concentration, the fluorescence intensity of the tested group increased gradually. In particular, the fluorescence intensity of Apt1 and Apt2 increased more in the linear range of 0.5–50 ng mL⁻¹. It indicated that Apt1 and Apt2 had a strong affinity for AFB₁ in this aptasensor. Based on the specificity and

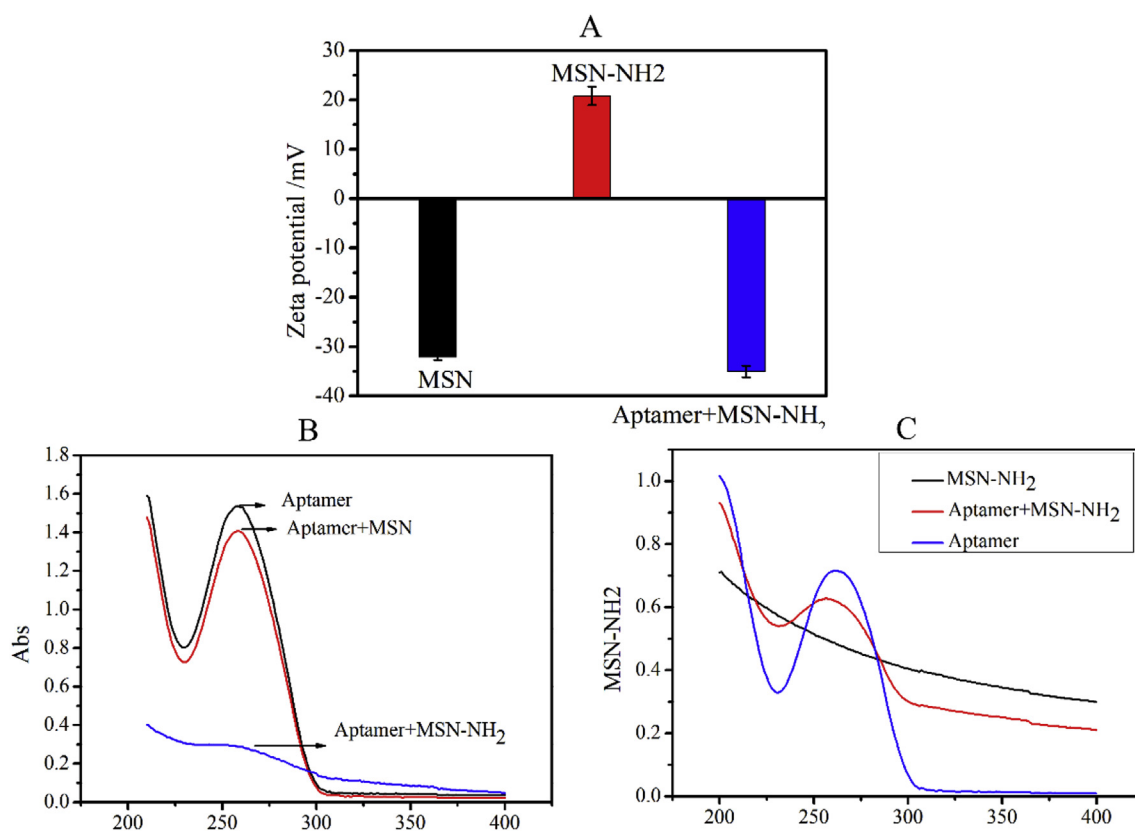


Fig. 3. (A) Zeta potential of MSN, MSN-NH₂, aptamer/MSN-NH₂ complex. (B) UV-Vis spectra of supernatant of pure aptamer, aptamer/MSN and aptamer/MSN-NH₂ complexes. (C) UV-Vis spectra of aptamer, MSN-NH₂ and aptamer/MSN-NH₂.

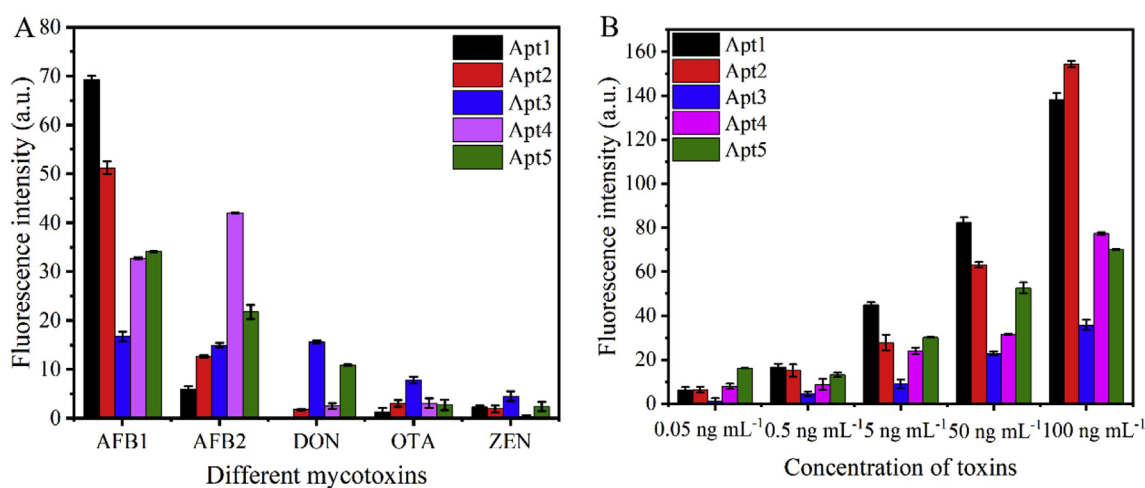


Fig. 4. (A) Evaluation of selectivity of the proposed MSN-based aptasensor for AFB₁ (each of mycotoxins was 30 ng mL⁻¹). (B) Evaluation of affinity of the proposed MSN-based aptasensor for AFB₁.

affinity of aptamers for the proposed method, Apt1 was selected as the molecular recognition probe of this designed aptasensor for the following experiments.

3.4. Study of experimental parameters

In order to improve the performance of the aptasensor, we investigated the recovery efficiency of the fluorescence intensity after the combination of different concentrations of Apt1 and MSN-NH₂. As shown in Fig. 5A, with the increase of Apt1 concentration

from 0 μ M to 120 μ M, the recovery value of the fluorescence intensity in the sensor system was gradually enhanced. The recovery value of the fluorescence intensity would be reached a plateau at a concentration over 100 μ M. The reason lied in that all MSN-NH₂ nanoparticles had electrostatically combined with Apt1 when the concentration of Apt1 was 100 μ M. Therefore, 100 μ M was selected as the Apt1 concentration for the following experiments.

The effect of the binding time of Apt1-MSN-NH₂ on fluorescence intensity was further investigated. As shown in Fig. 5B, the fluorescence intensity was gradually enhanced and reached a

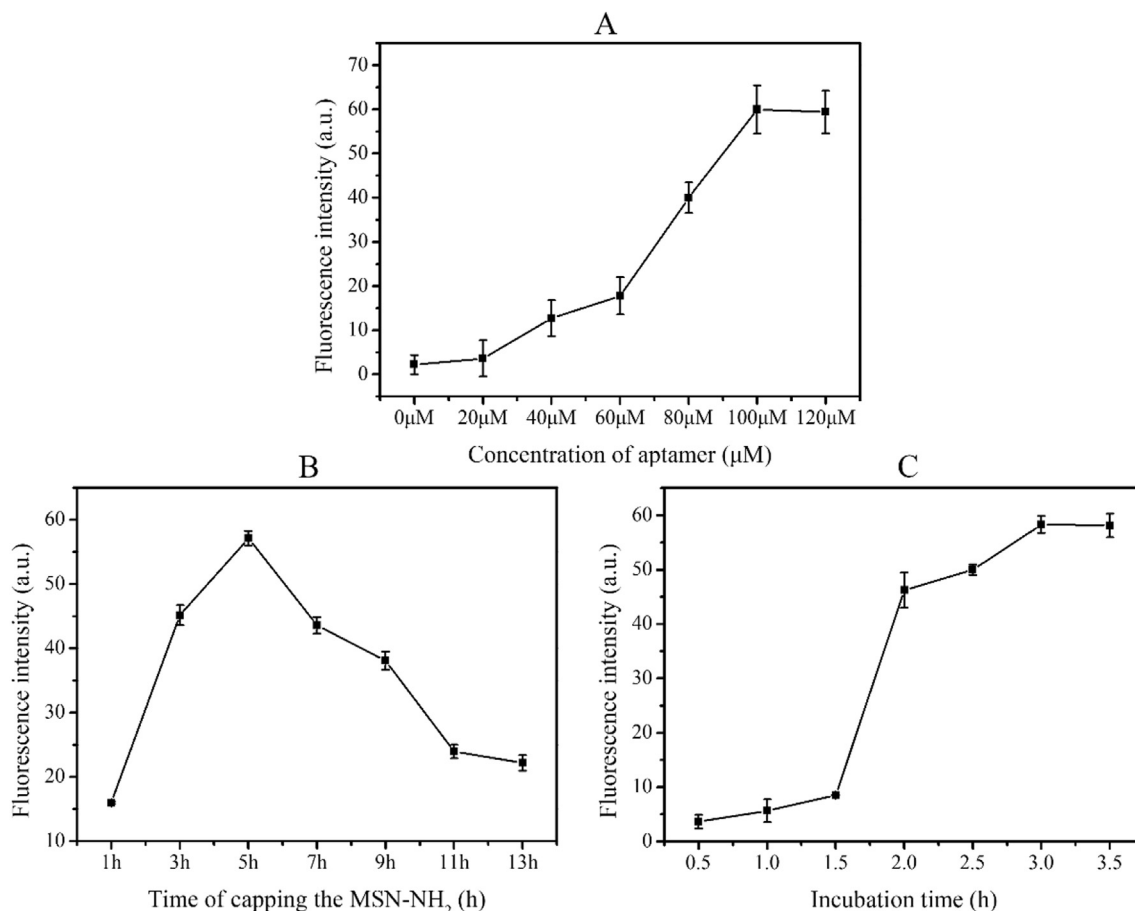


Fig. 5. (A) The effect of Apt1 concentration on the fluorescence intensity. (B) The effect of the capping time for the Apt1-MSN-NH₂ on the fluorescence intensity. (C) The effect of incubation time on the fluorescence intensity. The concentration of AFB₁ was 10 ng mL⁻¹.

maximum at 5 h. Then, the fluorescence intensity was decreased after 5 h. The phenomenon could be that the increasing Apt1 attached to the surface of MSN-NH₂ within 5 h, which improved the performance of the aptasensor. After 5 h, the interaction force between Apt1 and MSN-NH₂ was so strong that the release of signal probe Rh6G was affected. Therefore, the capping time of 5 h was selected as the Apt1 capping time for its beneficial to the sensitivity of the aptasensor.

The effect of incubation time for Apt1-AFB₁ in aptasensor on fluorescence intensity was investigated and shown in Fig. 5C. The fluorescence intensity was enhanced significantly with the incubation time consuming of AFB₁. The aptasensor had its best function based on the incubation time of 3 h for AFB₁.

3.5. Detection of AFB₁ based on MSN-NH₂ aptasensor

As shown in Fig. 6A, the fluorescence intensity of supernatant solution increased with the increasing of AFB₁. The fluorescence intensity reached a plateau when the AFB₁ concentration was up to 160 ng mL⁻¹. Besides, Fig. 6B showed a good linear correlation between the fluorescence intensity and the concentration of AFB₁ in the dynamic range of 0.5–50 ng mL⁻¹. The limit of detection (LOD) was calculated by $3\sigma_b/\text{slope}$ (σ_b , standard deviation of the blank samples) to be 0.13 ng mL⁻¹.

The characteristics of the proposed MSN-based aptasensor were compared with other methods reported previously and summarized in Table 2. The LOD of the proposed method was barely comparable to or even lower than most of the reported methods

previously. Moreover, the proposed MSN-based aptasensor was also compared with other methods based on fluorescence [49–51] and electrochemistry [52–54]. Results showed that the analytical method in this study had wider linear range, higher sensitivity, easier sample preparation, simpler operation and less time-consuming, holding great potentials in AFB₁ rapid detection.

3.6. Specificity of the aptasensor

In order to evaluate the selectivity and specificity of this aptasensor, 20 ng mL⁻¹ of mycotoxins (AFB₁, AFB₂, AFM₁, DON, OTA, ZEN, T-2) were tested under the optimized experimental conditions. As shown in Fig. 7, under the same conditions, strong fluorescence intensity of the aptasensor was only induced by AFB₁ while weak intensity was detected in the groups of both control and other mycotoxins. It was supposed that other mycotoxins could not specifically bind to AFB₁ aptamers and couldn't disrupt the electrostatic binding between MSN-NH₂ and aptamers. Thus, almost no fluorescent dye could be released into the solution to generate significantly fluorescent enhancement. Based on the results above, the MSN-based aptasensor method could eliminate the interference resulted from other mycotoxins and had a good specificity for AFB₁ determination.

3.7. Accuracy of aptasensor in samples detection

In order to evaluate the accuracy in samples detection, the proposed aptasensor was applied to determine AFB₁ in spiked corn

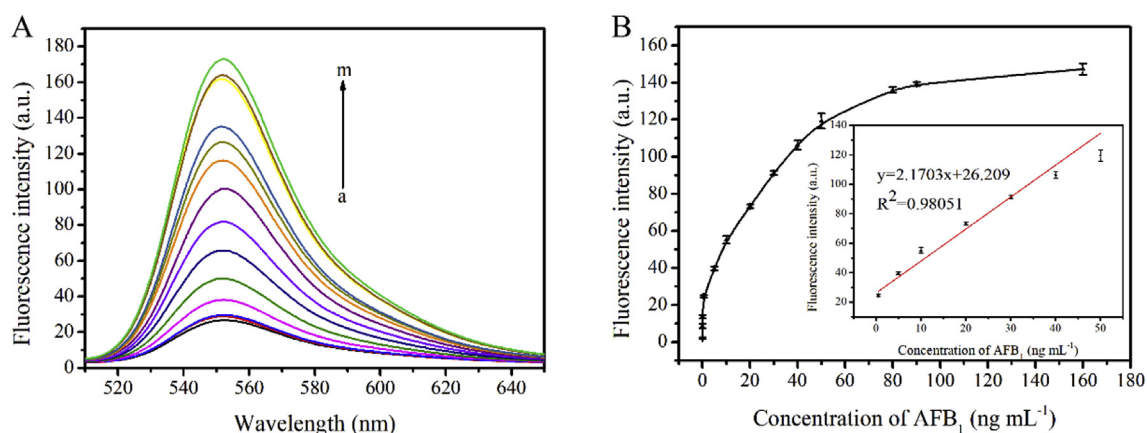


Fig. 6. (A) Fluorescence spectra of the proposed aptasensor with different concentrations (from a to m) of AFB₁: 0, 0.0005, 0.005, 0.05, 0.5, 5, 10, 20, 30, 40, 50, 80, 90, 160 ng mL⁻¹; (B) Linear relationship between the fluorescence intensity and AFB₁ concentration. Inset: a calibration curve for determination of AFB₁. Every data point was the mean of three measurements.

Table 2

Comparison of the method proposed in this study with other methods previously for AFB₁.

Methods		LOD (ng mL ⁻¹)	Reference
Antibody based methods	Immunochromatography assay	2.5	[55]
	Clean-up tandem immunoassay column	5	[56]
	Enzyme immunoassay	2	[57]
Aptamer based methods	A simple aptamer-based fluorescent assay	1.6	[50]
	A Rapid Label-Free Fluorescent Aptasensor	0.1	[51]
	An aptamer-based dipstick assay	0.1	[34]
	electrochemical aptasensor assay	0.25	[52]
	Aptasensor assay	0.13	This work

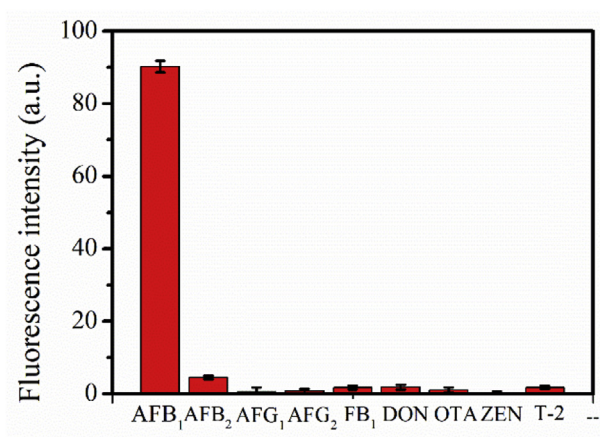


Fig. 7. Histogram of selectivity in the presence of 20 ng mL⁻¹ mycotoxins: AFB₁, AFB₂, AFG₁, AFG₂, FB₁, DON, OTA, ZEN, T-2. Each data point was the mean of three measurements with error bars.

oil and corn samples obtained from the local supermarket. Before the assay, the two samples were detected by HPLC to verify the absence of AFB₁. As shown in Table 3, the average recoveries of the spiked samples varied from 89.70%–94.70%.

Seven samples obtained from local supermarket were also applied to detect the concentration of AFB₁ by the aptasensor and HPLC [58] respectively (Table 4). No significant difference was found in the two methods ($P > 0.05$).

4. Conclusion

In summary, we develop a new aptasensor for AFB₁ detection which includes aptamers as a specific recognition element, mesoporous silica nanoparticles as a carrier and Rh6G as a fluorescent indicator. Under the optimal experimental conditions, the fluorescence intensity could be linearly increased when the concentrations of AFB₁ ranging from 0.5–50 ng mL⁻¹ (LOD = 0.13 ng mL⁻¹). This method will be promoted on the sensitivity and specificity for higher affinity and specificity aptamers sequence of AFB₁ obtained successfully in the future. Overall, this method could be applied for non-professional operators in a very simple mode. Besides, it

Table 3

Recoveries of AFB₁ in corn oil and corn samples for applicability of aptasensor ($n = 3$).

Recoveries sample	Found in sample (ng g ⁻¹)	Added AFB ₁ (ng g ⁻¹)	Found AFB ₁ (ng g ⁻¹)	% Recovery	% RSD
Corn oil	0	1	0.9	90.30	13.93
	0	3	2.84	94.70	7.02
	0	5	4.62	92.40	7.26
Corn	0	1	0.93	93.00	8.24
	0	3	2.69	89.70	7.75
	0	5	4.69	93.80	5.25

Table 4
AFB₁ contamination levels in samples.

Samples	AFB ₁ contaminated levels (ng g ⁻¹)	
	Aptasensor	HPLC-FLD
Peanut oil	17.02	16.68
Pepper	0.09	0.08
Corn oil	ND	ND
Corn	ND	ND
Oatmeal	ND	ND
Rice	ND	ND
Peanut	ND	ND

Note: P > 0.05.

apparently is a low-cost method compared with the general quantitative detection method such as HPLC with immunoaffinity column. With the advantages of novel “gate” design, simple pre-treatment, low cost, short time-consuming, high sensitivity and strong specificity, the developed aptasensor can be considered as a promising detection tool of AFB₁ in grain and oil products safety field.

Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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