



Direct and fast detection of *Alexandrium minutum* algae by using high frequency microbalance



Célia Sousa^{a,b,c}, Chantal Compère^d, Catherine Dreanno^d, Marie-Pierre Crassous^d, Fabienne Gas^e,
Beatrice Baus^e, Hubert Perrot^{a,b,*}

^a Sorbonne Universités, UPMC Univ Paris 06, UMR 8235, Laboratoire Interfaces et Systèmes Electrochimiques, F-75252 Paris, France

^b CNRS, UMR 8235, LISE, F-75005 Paris, France

^c Universidade Federal do Rio de Janeiro, Instituto de Química, Av. Athos da Silveira Ramos, 149, CT, Bl. A, sala 408, Ilha da Cidade Universitária, CEP: 21.941-909 Rio de Janeiro, RJ, Brazil

^d IFREMER, Centre de Brest, Laboratoire Détection, Capteurs et Mesures, F-29280 Plouzané, France

^e CEA Marcoule, Direction des Sciences du Vivant, IBEB/SBTN/LDCAE, F-30200 Bagnols sur Cèze, France

ARTICLE INFO

Article history:

Received 19 February 2014

Received in revised form 14 May 2014

Accepted 26 May 2014

Available online 11 June 2014

Keywords:

Alexandrium minutum

Quartz microbalance

Microalgae

Immunosensor

ABSTRACT

In this paper, a simple detection of a toxic algae, *Alexandrium minutum*, was developed using highly sensitive quartz crystal microbalance. In terms of performance, compared with other conventional analytical tools, the main interest of our immunosensor is based on a fast and direct detection of these living cells. This system requires the use of one monoclonal antibody directed against the surface antigen of *A. minutum*. We demonstrate that the whole living and motile algae are caught and detected. The high specificity of the biosensor is also demonstrated by testing several other dinoflagellate species. The frequency shift is correlated to the *A. minutum* cell concentration. This simple system is potentially promising for environmental monitoring purposes.

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1. Introduction

Although the red tides, called harmful algal blooms (HABs), are a natural occurring phenomenon, in the last decades, they have become a serious environmental and economic problem due to their adverse impact on the ecosystem and on human health (Hallegraeff, 1993; Van Dolah, 2000). Their frequency, intensity and geographic ranges have seriously increased since the 1970's. The increasing occurrence of algal blooms reflects the alteration in ecological systems as a consequence of human activities. HABs are often attributed to eutrophication, ballast water introduction and climate change (Hallegraeff, 2010). They are, nowadays, considered as one of the major problems that coastal countries must face, and it seems that this will be even greater in the near future (Masó and Garcès, 2006). In Europe, the genus *Alexandrium* is often involved in paralytic shellfish poisoning (PSP) events. The current method for the accurate identification of this dinoflagellate is based on the microscopic examination of its specific morphological features. This procedure requires considerable broad taxonomic expertise; it is time consuming and costly. As a result, the rapid high throughput of

samples is beyond the detection capabilities of most monitoring programs and does not effectively prevent or mitigate the adverse effects of these toxic events. Consequently, improved, reliable and cost effective monitoring methods are urgently needed. The application of biosensor technology is ideal for marine environmental monitoring and detection as the sensors are portable and provide selective and rapid responses in real time in the field (Kröger and Law, 2005; Zielinski et al., 2009).

Various types of sensors are currently being developed to detect microorganisms in biological and environmental samples. They are based on different transducers such as optical, electrochemical, gravimetric systems and several sensitive layers like immunomolecules or nucleic acid molecules. Quartz crystal microbalance (QCM) technology is particularly interesting for environmental monitoring and diagnosis due to its rapid analysis, satisfactory sensitivity, simple low cost instrumentation (Hao et al., 2009) and has become widely used for pathogen detection (Olsen et al., 2003; Buchatip et al., 2010; Hong et al., 2009; Shen et al., 2011; Guntupalli et al., 2012). The immunosensors are advantageous since they can be based on the direct detection of the whole cell through antibody selective layers immobilized on the transducer surface. The key feature of this approach is to obtain suitable antibodies particularly in terms of affinity and specificity. Nakanishi et al. (1996) were pioneers in using a conventional QCM and different methods of antibody immobilization to detect *Chatonella marina*.

* Corresponding author at: University Paris 06, LISE, F-75252 Paris 05, France. Tel.: +33 144277216; fax: +33 144274074.

E-mail address: hubert.perrot@upmc.fr (H. Perrot).

Electrical sensing combined with carbon nanotube structures was recently set up to detect *Aureococcus anophagefferens* (Ishikawa et al., 2009). Nucleic sensors are currently developed since their specificity appears to be excellent (Nayak et al., 2009). However, these systems do not allow the direct detection of whole targeted microorganisms. Preliminary steps of preparation such as cell lysis and targeted DNA amplification are needed before performing the detection step. Nevertheless, sophisticated sensors have already been designed thanks to conventional DNA immobilizations, through electrochemical detection (Diercks et al., 2008), optical detection (Lagier et al., 2007) or gravimetric detection (Gamby et al., 2008; Lazerges et al., 2006). An electrochemical genosensor for the detection of *Alexandrium minutum* was developed by Diercks et al. (2008). This system uses oligonucleotide probes hybridizing rRNA genes in the ribosomes of target cells. It depends on the rRNA content of the cells, but the rRNA level varies according to the cell growth cycle leading to error estimations in cell concentrations.

A sophisticated Environmental Sample Processor (ESP) based on nucleic hybridization was successfully designed for detecting plankton and microorganisms in the field in real time (Scholin et al., 2001; Scholin, 2010). This autonomous device is state of the art technology and nowadays it is too costly for routine environmental monitoring in any national program. In France, recurrent blooms of *A. minutum* mainly occurred in Brittany (f.i Bay of Morlaix, Penzé estuary, Bay of Brest, Rance estuary). These yearly blooms developed in the period June–July and often reached cell concentrations $>1 \times 10^5$ cells L^{-1} . For sanitary monitoring purposes, when the alert threshold of 10 000 cell L^{-1} is exceeded, biotoxin analysis is then performed in shellfishes. Therefore, the sensitivity of the biosensor or monitoring system needed to meet this alert threshold. It could be achieved by the immunosensor itself or by combining a phytoplankton concentration step prior the detection analysis.

In this paper, we explore for the first time, the possibility of using QCM as a simple, rapid and label free method to detect and quantify directly *A. minutum* algae after a simple and single step of concentration. The monoclonal antibody is prepared against these microalgae and is immobilized on the gold electrodes of quartz resonators. Fast and in situ measurements are obtained using high frequency microbalance based on a stable and thermo-regulated oscillator. The limits of detection, the dose responses and the selectivity of this QCM sensor are carefully evaluated.

2. Materials and methods

2.1. Quartz microbalance apparatus

9 MHz AT-cut planar quartz crystals (14 mm diameter) were purchased from RAKON Company (RAKON FRANCE, Mougins, France). Two identical gold electrodes were deposited on both sides of the crystal with a standard keyhole shape and they were connected to a BNC connector. The crystal was mounted between two O-Ring seals inserted

in a Plexiglas cell whose chamber volume was about 50 μL . Only one side of the quartz resonator was in contact with the solution. To produce a flow system, a P-1 peristaltic pump (Pharmacia) was connected to the chamber; a constant flux (70 $\mu L \text{ min}^{-1}$) of solutions was maintained during the experiments. A homemade oscillator was designed to drive the quartz crystal at 27 MHz (3rd overtone). A frequency counter (Philips PM 6685) was coupled to the QCM to monitor the microbalance frequency, f_m (Fig. 1a). Under the gravimetric regime, a linear relationship was obtained between the mass change, Δm , due to the different interactions on the resonator surface and the frequency change, Δf_m . At a 27 MHz resonant frequency, a theoretical value of 0.36 ng Hz^{-1} was deduced and an experimental value of 0.35 ng Hz^{-1} was used according to Bizet et al. (2000) and to Rodriguez-Pardo et al. (2004).

To validate the gravimetric regime of such a device, electroacoustic measurements were performed on this system under the same experimental conditions. The experimental set up was based on a standard network analyzer (HP 4194A) and was computer-controlled through a lab made software written in HP-VEE® language. Electrical admittance measurements were automatically carried out with a 10 mV perturbation signal at 201 frequencies around the resonance frequency. They were monitored in real time during the different reactions and the equivalent circuit could be estimated (Garcia-Jareno et al., 2000). The series resonant frequencies, f_s , and the motional resistance, R , were extracted during the experiments every 1.5 min.

2.2. Chemical reagents

Phosphate buffered saline solution (PBS; 137 mM NaCl, 2.7 mM KCl, 10 mM $\text{Na}_2\text{HPO}_4 \cdot 2\text{H}_2\text{O}$, 2 mM KH_2PO_4 , pH 7.4), H_2SO_4 95%, H_2O_2 30%, HCl, NaCl, and bovine serum albumin (BSA) were purchased from Sigma Aldrich. All the other chemicals used in this study were analytical grade.

2.3. Immobilization of the monoclonal antibody against *A. minutum* cells

The AMI6 monoclonal antibody against *A. minutum* was selected because of its high specificity for this study. The production, specificity and characteristics have been previously reported (Gas et al., 2009, 2010). This IgG was purified on a protein G HiTrap® affinity column (GE Healthcare Life Sciences) prior to immobilization and diluted in PBS at the final concentration of $100 \mu\text{g mL}^{-1}$.

The gold-coated QCM electrodes were cleaned by soaking in a Piranha solution (30% H_2O_2 and concentrated H_2SO_4 , 1:3 v/v) for 20 min, rinsed thoroughly with distilled water and then, dried under argon prior to assembling them in the QCM cell. The cleaned electrodes were flushed with 0.1 M HCl, distilled water and then PBS. The antibody solution flowed through the QCM cell for 90 min at ambient temperature. The excess and unbound antibodies were removed by rinsing the chamber with PBS for 30 min. Then, a PBS solution containing 1% BSA solution was injected for 10 min to block any non-specific interactions that might occur between the gold substrate and antigens (Brewer et al., 2005). After

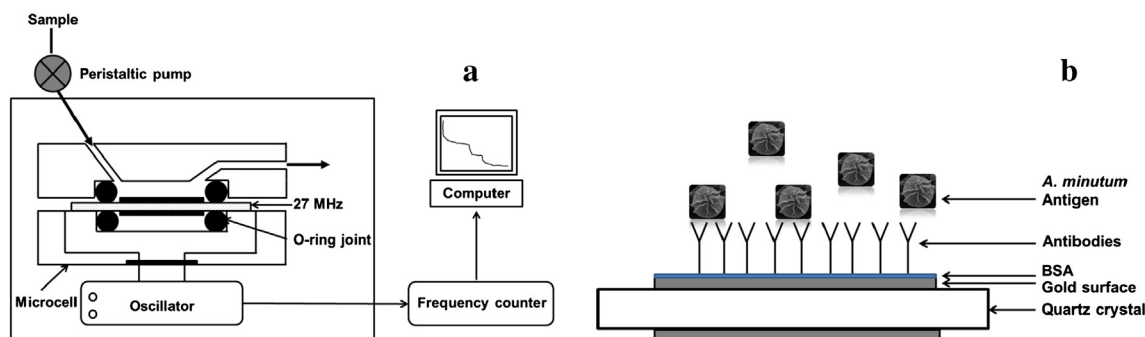


Fig. 1. Schematic illustration of the experimental QCM detection system (a) and functionalized QCM sensor by monoclonal AMI6 antibodies (b).

the antibody immobilization step, PBS, free of BSA flowed through the QCM cell until the stabilization of the base line (Fig. 1b).

2.4. Dinoflagellate strains and cultures

All the microalgae strains, *A. minutum*, *Heterocapsa triquetra*, *Karenia mikimotoi* and *Alexandrium catenella* came from IFREMER's collection. They are indicated by using letters A (*A. minutum*), H (*H. triquetra*), K (*K. mikimotoi*), AC (*A. catenella*), respectively. The phytoplankton strains were cultivated using natural seawater enriched f/2 medium (Guillard, 1975). This methodology is commonly used for phytoplankton culture (Andersen, 2005) and is justified in a recent study about the use of quantitative LFIA test using the same monoclonal antibody (paper in preparation). Tests were performed with several stages of a phytoplankton culture and several *A. minutum* strains coming from different genetic populations and geographic areas. We did not observe any significant difference or bias to quantify the *A. minutum* cells in these various samples using the calibration curves set up with *A. minutum* coming from our laboratory cultures.

The cultures were grown at 18 ± 1 °C under cool-white fluorescence light, at a photon flux of $150 \mu\text{E}/\text{m}^2/\text{s}$ with 14:10 LD photoperiod. At the end of the exponential growth phase, the cells were harvested by centrifugation (5000 g, 10 min) and kept frozen at -20 °C for later. Frozen and living cells were both used in the experiments. The microalgae medium culture used in the QCM experiment was also called working solution (WS).

The algae culture were also concentrated by using a convenient and quick procedure. The cell suspensions were filtered on cellulose filter papers with particle retention levels down to $2.5 \mu\text{m}$ (Whatman) and then, the recovered algae were resuspended with the WS. In order to determine the cell concentration before each QCM experiment, 1 ml of algal suspension was sampled and fixed with a drop of Lugol's iodine solution and the cells were counted under a light microscope using a hemocytometer (Karlson et al., 2010).

2.5. Quartz crystal microbalance measurements

The same experimental procedure was followed for all the QCM measurements. The working solution, WS, was injected through the detection cell until stabilization before the injection of the same solution containing the selected algae at a given concentration. After 1 h of circulation with the algae solution, the WS and the WS diluted at 50% with distilled water flowed through the QCM cell for 20 min and 10 min respectively. It should be noted that the immunosensors obtained were single used and for each detection experiment a new quartz resonator was prepared.

At the end of each experiment, the quartz crystal resonator was carefully removed from the cell, dipped in distilled water, dried, maintained in a glass container and a formaldehyde drop was deposited on the gold surface before scanning electron microscopy (SEM) observations. This fixation step allowed good preservation of the immobilized algae. Additionally, a direct estimation of the algal coverage rate was estimated for each picture through a Bruker soft (Area Fraction Report).

2.6. Scanning electronic microscopy analysis

The scanning electron analysis was carried out according to the protocol of Gas et al. (2009). The quartz crystal resonators were observed using a Quanta 200 instrument (FEI Company, Hillsboro, Oregon, USA).

3. Results and discussion

3.1. Non-specific interactions

Firstly, the response of the microbalance sensor without the presence of antibodies on the gold surface was examined by using living

cells and frozen cells in the WS (data not shown). The gold electrode surface was previously covered by a thin BSA layer directly adsorbed. In each case, the relative microbalance frequency change, Δf_m , was negligible ($5 \text{ Hz} < \Delta f_m$) demonstrating that there was no interaction between the BSA coated surface and the microalgae solution even when highly concentrated cell suspensions were injected ($[A. \text{minutum}] > 1 \times 10^7 \text{ cells L}^{-1}$). Thus, the BSA was used to saturate free sites on the gold electrode and then, unspecific interaction was reduced to a minimum ratio in this way. These results were also confirmed by SEM, no cell was observed on the QCM sensor (data not shown).

3.2. Direct detection of *A. minutum* algae

Fig. 2 shows the direct detection of *A. minutum* cells with the QCM when the AMI6 antibody was adsorbed on its surface. When the monoclonal antibody solution flowed through the microcell, a microbalance frequency change of -500 Hz was observed. This steep decrease in frequency is due to the immobilization of these antibodies by direct adsorption onto the surface. The other change in frequency between the blocking step by BSA and the washing step by WS is related to the different compositions of these solutions. It is well known that the QCM device is sensitive to a change in viscosity/density and conductivity of the solution (Arnaud, 2008).

When the living algae were injected through the chamber after the saturation step, a fast and steep microbalance frequency change, Δf_m , of -540 Hz was observed reflecting the interactions between the modified gold surface and the living cells suspended in the WS. After 30 min, the microbalance frequency reached a plateau. When WS was injected during this last phase, the microbalance frequency remained constant demonstrating that the interactions between the antibodies against the algae were very strong i.e. cells stayed on the surface since no mass change was detected. The experiments have been repeated and reasonable standard deviations were found for this technique in the range of a few tenths of percent.

After the detection of *A. minutum* cells by the functionalized QCM sensor, the surface of the gold electrode was observed by SEM. In Fig. 3a and b, images of the alga cells were captured by the QCM sensor. In Fig. 3a, a large surface was explored, indicating the number of cells present on the whole surface. The coverage rate could be deduced through SEM analysis by taking the mean number of algae observed on the surface. A value of 10–15% was directly determined here through the Bruker software estimation. The *A. minutum* shape can be considered as a hemispheric shell and a diameter of 10–15 μm can be roughly estimated (Fig. 3b). According to the morphological observations

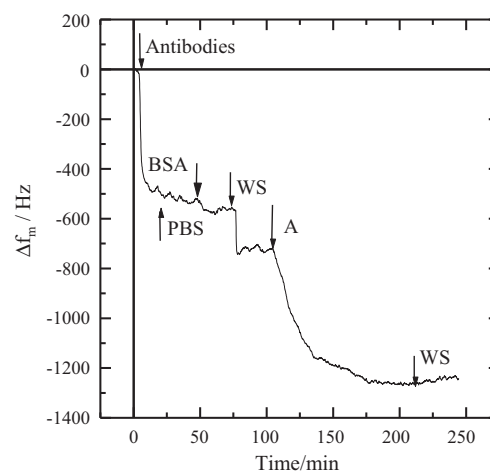


Fig. 2. Direct and fast detection of living cells ($5.6 \times 10^6 \text{ cells mL}^{-1}$) through QCM measurements after modification of the gold electrode with adsorbed antibodies against *A. minutum*.

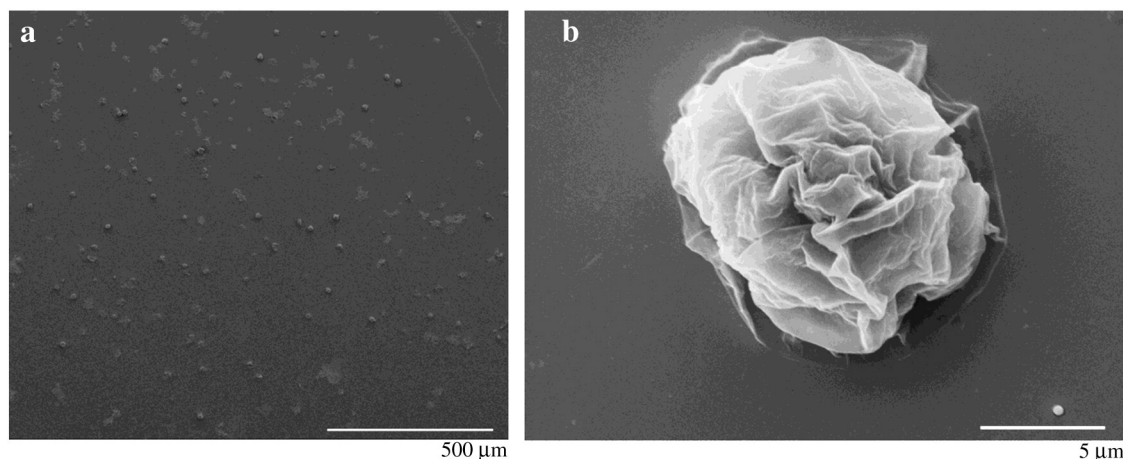


Fig. 3. Scanning electron pictures of the gold electrode after the capture of *A. minutum* (living cells) by antibodies adsorbed on gold interaction at different scales (a, scale bar: 500 µm, b, scale bar: 5 µm).

reported on the same microalga species by several authors (Probert, 1999; Labry et al., 2008), the same order of magnitude in parameters was also obtained. Then, an estimated mass per cell can be calculated by taking into account a cell density of 1200 kg m^{-3} and a volume of $8 \times 10^{-15} \text{ m}^3$ if an hemispheric shell is used with a 15 µm diameter. It should be noted that this volume is in agreement with the observations of Lim (Lim and Ogata, 2005). Therefore, a weight value of $9.6 \times 10^{-12} \text{ kg}$ per cell is obtained. In Fig. 2, the microbalance changes, Δf_m , is estimated at -540 Hz which leads, in terms of mass change, to 194 ng by considering the gravimetric regime validated. Then, the theoretical number of algae is around 20, which corresponds to a surface coverage rate of 0.05%. However, a value of 10–15% was obtained by SEM analysis. This difference can be explained by electroacoustic measurements. In Fig. 4a, the relative series resonant frequency, Δf_s , is plotted against time during the detection step. Regarding Fig. 2, using the standard microbalance approach, the microbalance frequency changes, Δf_m , present a similar shift compared with Δf_s and the time constant is in the same order of magnitude. The two systems of detection, conventional microbalances or electroacoustic admittances, give the same response. The validity of the gravimetric regime can be controlled through the calculation of the ratio, $\frac{\Delta R}{\Delta f_s}$, which appears as a pertinent variable for this purpose (Lucklum et al., 1999, 2000). A transient with large values was observed several minutes during the cell

interaction before reaching a stationary value around $+8 \Omega \text{ Hz}^{-1}$ shown in Fig. 4b. In this present case, it means that the gravimetric regime of the microbalance is questionable. This idea is already underlined in other examples where cells are detected by QCM (Pomorska et al., 2010). When this type of biomolecules is detected, the gravimetric regime is not well respected. In this way, the number of captured cells estimated through QCM measurements can be inaccurate. This may explain the difference in surface coverage estimated through the QCM approach and the SEM analysis. Moreover, it strengthens the necessity to build a calibration curve for any biomolecules to be quantified in order to obtain the correct concentration.

3.3. Calibration curves for *A. minutum* detection

A calibration curve is obtained by measuring the microbalance frequency shift, Δf_m , at the same time of interaction at different algae concentrations (Fig. 5). A good linear fitting was obtained in our concentration range with a correlation coefficient, R , of -0.99 . A direct estimation of an unknown concentration of *A. minutum* cells in environmental samples can be then deduced. The detection limit seems to be around $1 \times 10^6 \text{ cells mL}^{-1}$ under our experimental conditions. These detection ranges are in agreement with previous studies using QCM

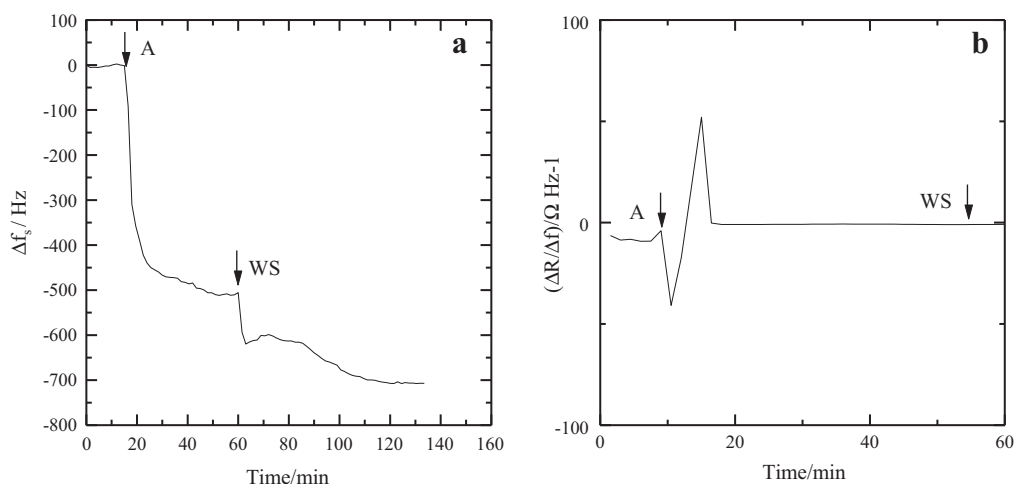


Fig. 4. Electroacoustic measurements during algae detection: a) Changes in the resonant frequency series, Δf_s , and b) Changes in the variable, $\frac{\Delta R}{\Delta f_s}$, of the quartz resonator during the algae detection, living cells ($5.6 \times 10^6 \text{ cells mL}^{-1}$).

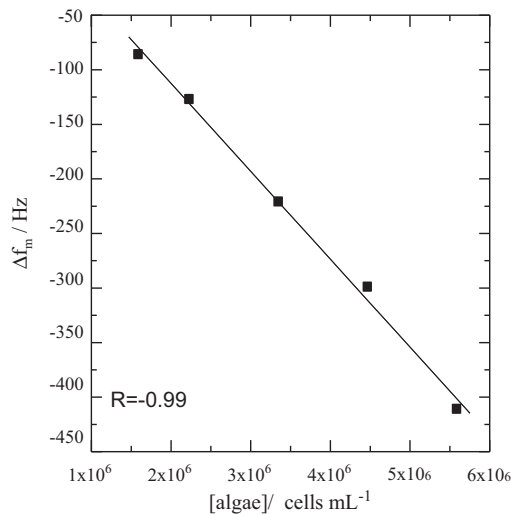


Fig. 5. Calibration curve obtained with different cell concentrations of *A. minutum*.

immunosensor detecting other microorganisms (Hong et al., 2009) and a marine microalga species, *C. marina* (Nakanishi et al., 1996).

3.4. Detection of *K. mikimotoi*, *A. catenella*, and *H. triquetra*

The specificity of a biosensor is essential for developing a safe detection tool for environmental monitoring. The specificity was tested with the main toxic species monitored on our coasts. In order to study the specificity of our system, the affinity of the functionalized QCM immunosensor was evaluated by using different dinoflagellates, showing the same shape or frequently found in the same ecological niche. Samples of *K. mikimotoi*, *A. catenella*, *A. minutum* and *H. triquetra* were injected into the QCM immunosensor (Fig. 6a and b). As a result, a significant Δf_m was only observed in the presence of *A. minutum*. For the other alga species, a steady microbalance frequency signal was observed indicating that non-specific interactions did not occur. The AMI6 antibody demonstrates a very high specificity for *A. minutum* since it does not cross-react with *K. mikimotoi*, *A. catenella* and *H. triquetra* cells. Each graph represents one experiment done on the same modified resonator in a continuous way after successive additions of the different algae. In our case, a single use is proposed as this tool can be used as a first indicator.

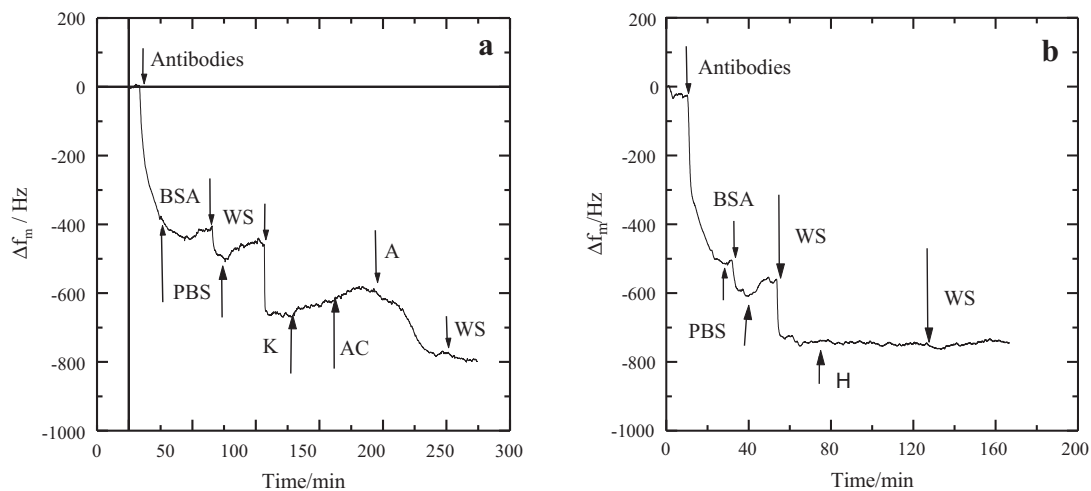


Fig. 6. Specificity of the QCM immunosensor. a) Detection of *K. mikimotoi* (2.8×10^6 cells mL⁻¹), *A. catenella* (6.8×10^5 mL⁻¹) and *A. minutum* (1.5×10^6 cells mL⁻¹) and b) detection of *H. triquetra* (2.5×10^6 cells mL⁻¹) with antibodies adsorbed on the gold electrode of the microbalance.

4. Conclusion

The feasibility of a direct and fast detection of toxic algae *A. minutum* is demonstrated in this work by using a 27 MHz microbalance incorporated into a microcell. In terms of gravimetric interpretation, our acoustic device can not accurately estimate the mass of algae immobilized after the detection steps. This result is due to the shape, size, mechanical properties and the way of interaction of these specific biomolecules. The gravimetric regime is only retained when uniform and purely elastic films are deposited on the QCM surface. This result strengthens the need to set up correlation curves for any new microorganisms. In terms of selective identification, stability, cost and simple use, this biosensor shows very attractive potentialities for environmental monitoring. However, even if this system is suitable for some bloom samples, the relative lack of sensibility of this biosensor could easily be overcome by a preliminary concentration of the sea water sample.

Acknowledgments

This work is part of the HAB-SEACHIP project, supported by a grant from the ANR PRECODD program.

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