

Ratiometric Antifouling Electrochemical Biosensors Based on Multifunctional Peptides and MXene Loaded with Au Nanoparticles and Methylene Blue

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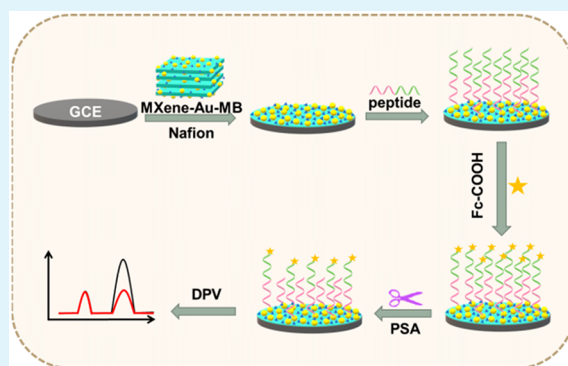
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ABSTRACT: A universal strategy for the construction of ratiometric antifouling electrochemical biosensors was developed based on multifunctional peptides and 2D nanomaterial MXene loaded with gold nanoparticles (AuNPs) and methylene blue (MB). The nanocomposite of MXene loaded with AuNPs and MB (MXene-Au-MB) exhibited excellent conductivity, where the AuNPs were able to capture biomolecules containing sulfhydryl terminus, and the MB molecules were used to generate electrochemical signal. The MXene-Au-MB was fixed on the electrode surface by Nafion, and the anchored peptide captured the electrochemical signal probe carboxyl-modified ferrocene (Fc) to construct an electrochemical biosensor. The multifunctional peptide endowed the sensing surface not only the assaying function but also the capability to resist nonspecific adsorption from complex samples. In the biosensing system, with the increase in the target concentration, the electrochemical signal of MB remained constant, whereas the electrochemical signal of Fc gradually decreased, and the ratiometric detection strategy greatly improved the accuracy of the biosensor. In the presence of a model target prostate-specific antigen (PSA), the recognizing sequence was recognized and cleaved, and the ratiometric signal of Fc and MB indicated the concentration of PSA accurately and sensitively, with a detection range from 5 pg/mL to 10 ng/mL and a limit of detection of 0.83 pg/mL. Electrochemical biosensors based on the MXene-Au-MB and multifunctional peptides possessed high selectivity, accuracy, and sensitivity even in real complex biological samples because of the excellent antifouling ability of the peptide. More importantly, the assaying of other targets can be easily realized with a similar biosensing strategy by changing the recognition sequence of the multifunctional peptide, and the detection of thrombin (TB) has also been achieved in this work.

KEYWORDS: electrochemical biosensor, MXene, antifouling, ratiometric detection, multifunctional peptide



1. INTRODUCTION

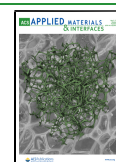
Since they were published for the first time in 2004,¹ two-dimensional (2D) nanomaterials have been widely studied and applied in many fields, such as electrocatalysis, optics, electrochemistry, etc., because of their excellent properties.^{2–4} So far, a great quantity of 2D nanomaterials, including boron nano chips (BNSs), black phosphorus (BP), graphite carbon nitride (g-C₃N₄), transition metal oxides (TMOs), and graphene have been discovered.^{5–10} Two-dimensional nanomaterials have broad application prospects in electrochemical biosensors because of their large specific surface area and excellent physical and chemical properties.¹¹ Recently, the new 2D nanomaterial MXene possesses a unique combination of excellent electrical conductivity, easily tunable structure, rich functional groups, hydrophilicity, and large interlayer spacing, making it an attractive research object. In the past few years, MXene has achieved encouraging results in many fields, including energy storage devices,^{12–15} photoelectrocataly-

sis,^{16,17} flexible wearable sensors,¹⁸ electronic devices,¹⁹ and photothermal conversion.²⁰ It should be noted that the unique accordion-like structure, strong reduction ability, and compatibility with other nanomaterials make it easy for MXene to form composites with other materials. It has been proved that MXene has a strong reducing ability and can reduce metal nanoparticles without reducing agents and stabilizers. For example, AuNPs-MXene nanocomposite and MXene/Ag composite have been obtained through the strong reduction ability of MXene and have excellent electrical conductivity and

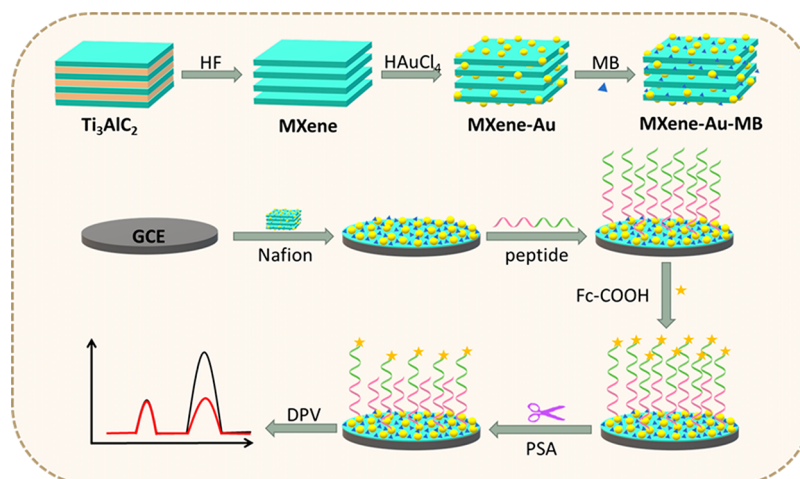
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Scheme 1. Schematic Illustration of Ratiometric Antifouling Electrochemical Biosensor Working Mechanism



electrocatalytic activity.^{21,22} In addition, as an exceptional charge transport accelerator and an excellent immobilization matrix, MXene has considerably high surfaces to volume ratio, exceptional conductivity, and adsorption capacity, which would be quite suitable for electrochemical biosensing, but there is not much research in this area.

As one of the most commonly used detection methods in scientific research, electrochemical biosensors have the advantages of good selectivity, high sensitivity, and simple operation. Nevertheless, it is still a challenge to specifically and accurately detect targets in complex biological samples because of nonspecific adsorption of biomolecules to the sensing interface. So it is urgent to build an antifouling interface to improve the selectivity and lifetime of electrochemical sensing. Currently, the most commonly used antifouling materials are polyethylene glycol (PEG)^{23–25} and zwitterionic materials.^{26,27} However, PEG is susceptible to oxidative damage when being used for a long time, whereas most zwitterionic materials cannot be synthesized easily. In recent years, Luo's group has been designing and synthesizing series of zwitterionic peptides as antifouling materials because of their excellent biocompatibility, a convenient synthesis process, and design that allows them to be highly hydrophilic and electrically neutral as required.^{28–33}

The internal standard method has attracted more and more attention to effectively avoid the potential influence of internal and external factors.^{34,35} Our group has constructed a dual-signal ECL cytosensor with high sensitivity and reliability,³⁶ in which luminol was used as the internal standard molecule, and CdTe quantum dot (QD) was used as the signal molecule to be labeled on the cancer cell aptamer to indicate the number of cancer cells. The ratiometric dual signal spectrum effectively eliminates the interference in the system to get more accurate detection results, even in a complex environment. Previous work has also proved that the built-in correction function in the ratio strategy can minimize the deviation between multiple electrodes and effectively overcome the potential impact of system or background noise.^{33,34,37–39}

In the present work, gold nanoparticles (AuNPs) grew on the surface of MXene in situ without additional reductant and stabilizer. Meanwhile, methylene blue (MB), as an electrochemical internal molecule, was also decorated to the MXene-Au to form a novel nanocomposite material (MXene-Au-MB). Subsequently, the nanocomposite material was fixed to the

surface of the electrode with the help of Nafion. After modifying with a designed multiple-functional peptide, which has an anchor sequence (CPPPP), an antifouling sequence (DRDRDRD), and a recognizing sequence (HSSKLQK for PSA, GSRPVLG for TB),³³ an antifouling sensing surface with excellent conductivity and large specific surface area was obtained. Aspartic acid (D) and lysine (K) are both hydrophilic amino acids, and D is negatively charged and K is positively charged. So the hydrophilic and electrical neutral antifouling sequence has an antifouling property. By changing the recognizing sequence of the peptide, we can detect prostate-specific antigen (PSA) and thrombin (TB), respectively, indicating a universal ratiometric antifouling electrochemical biosensor was constructed successfully.

EXPERIMENTAL SECTION

Synthesis of $\text{Ti}_3\text{C}_2\text{Tx-MXene}$. MXene is made by etching Al in Ti_3AlC_2 with HF at room temperature.⁴⁰ Briefly, 2 mg of Ti_3AlC_2 powder and 8 mL of 40% HF solution were mixed and stirred for 48 h at room temperature. After the etched reaction, the material was washed repeatedly by deionized water and centrifuged until the supernatant reached a pH ≥ 6 . The precipitate was then dried in vacuum at 60 °C for 12 h to obtain MXene for future use.

Synthesis of MXene-Au. MXene-Au nanocomposite was synthesized according to refs 21 and 22. First, 1 mg of MXene was added to 5 mL of deionized water and sonicated for 20 min to make it evenly dispersed in water. The HAuCl_4 (1 mL, 20 mM) solution was then added dropwise under stirring. After 5 min of reaction, the supernatant and precipitate were separated by centrifuging and washing three times. Finally, the precipitate was dried in a vacuum at 60 °C for 12 h to obtain MXene-Au for subsequent use.

Synthesis of MXene-Au-MB. MXene-Au composite was added into a 10 μM MB solution and treated by ultrasonicated for 30 min, the MB molecules were adsorbed on MXene-Au to obtain the MXene-Au-MB nanocomposite through electrostatic adsorption. The excess MB was removed by centrifugation and washing three times.

Construction of the Electrochemical Sensing Surface. A glassy carbon electrode (GCE) was polished and ultrasonically cleaned before use. The MXene-Au-MB solution containing 5% Nafion was fixed on the surface of the GCE electrode and naturally dried at room temperature. The modified electrode was incubated in a 0.2 mg/mL peptide solution (containing 10 mM TCEP) at 4 °C for 12 h and then washed three times with PBS to remove unreacted peptides to obtain the peptide/MXene-Au-MB/GCE sensing interface. Ferrocenedicarboxylic acid (Fc) was dispersed in PBS containing EDC (0.4 M) and NHS (0.1 M) and incubated for 1 h to activate the carboxyl group on the Fc. Subsequently, the peptide/MXene-Au-MB/

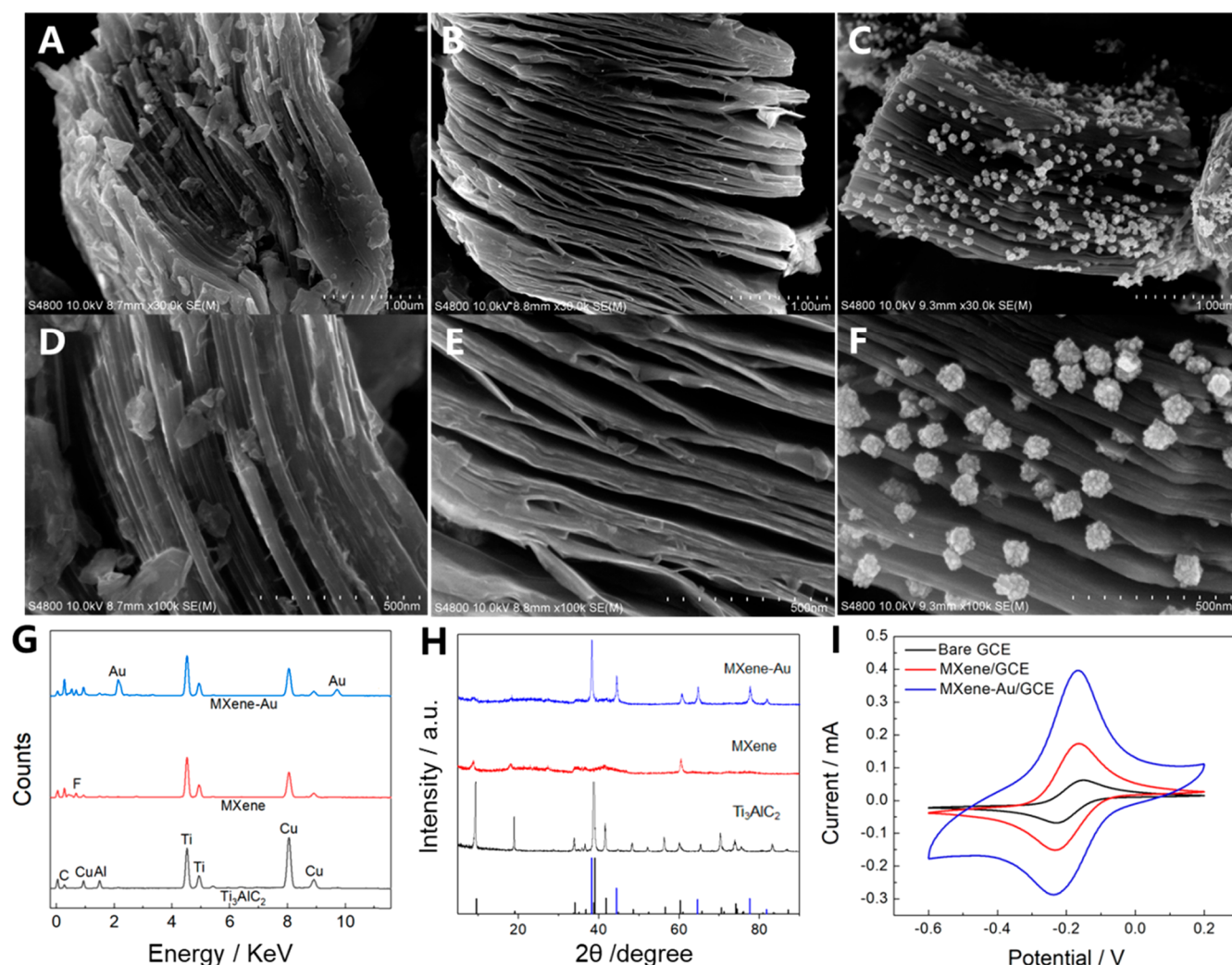


Figure 1. SEM images of (A, D) Ti_3AlC_2 , (B, E) MXene, and (C, F) MXene-Au composite at different magnifications. (G) EDS spectra and (H) XRD patterns of Ti_3AlC_2 , MXene, and MXene-Au composite. (I) CVs at bare GCE, MXene/GCE, and MXene-Au/GCE in 5 mM $[\text{Ru}(\text{NH}_3)_6]^{3+}$ containing 0.1 M KCl by scanning the potential from -0.6 to 0.2 V at a scan rate of 0.1 V/s.

GCE electrode was incubated in activated Fc solution for 2 h to construct the Fc/peptide/MXene-Au-MB/GCE sensing interface via amide bonds. Finally, the electrode was washed with PBS buffer three times.

Electrochemical Measurements. The Fc/peptide/MXene-Au-MB/GCE sensing interface was incubated in PSA solution for 40 min. After washing with PBS, the biosensor response data were recorded using DPV, the scanning potential is from -0.5 to 0.5 V, the amplitude is 50 mV, the pulse period is 0.5 s, and the potential increment is 4.0 mV.

RESULTS AND DISCUSSION

Response Mechanism of the Ratiometric Antifouling Electrochemical Biosensor. As shown in Scheme 1, after the Al layer in Ti_3AlC_2 was etched by HF to obtain MXene, AuNPs were synthesized on MXene material to obtain MXene-Au by an in situ synthesis method. MXene is a stable and powerful loading platform for adsorbing organic dyes,^{41,42} and the strong electrostatic adsorption (Figure S1) between MXene-Au and MB makes MB firmly loaded into MXene-Au to obtain MXene-Au-MB nanocomposite. After fixing the MXene-Au-MB to the GCE with Nafion, the specific surface area and the conductivity of the electrode is greatly improved. At the same time, MB can be introduced into the electrode

surface as an internal standard molecule. In addition, AuNPs can provide peptide binding sites so that the peptides can be immobilized on the electrode surface through Au-S bonds. The other end of the peptide has an amino group, so the Fc can be captured on the modified electrode surface through the amide bond. In the presence of targets, the free part of the peptide can be recognized and cleaved. Fc molecules fall off from the electrode surface, resulting in a decrease in the DPV signal of Fc, whereas the signal of MB remains unchanged. The ratio change of Fc and MB signals can be used to determine the PSA concentration. More importantly, one part of the peptide is designed to have antifouling performance, so the biosensor can detect PSA in human serum and other complex medium with a quite good selectivity. It is worth noting that by changing the recognition sequence in the peptide, the same detection theory can be applied to the highly selective and sensitive detection of other targets.

Characterization of Ti_3AlC_2 , MXene, MXene-Au, and MXene-Au-MB. The morphology and elements of MXene, MXene-Au, and MXene-Au-MB are recorded by SEM, EDS, XRD, and fluorescence, and obvious changes can be observed from Figure 1. After the Al element is corroded, Ti_3AlC_2 with a densely stacked layered structure (Figure 1A, D) becomes

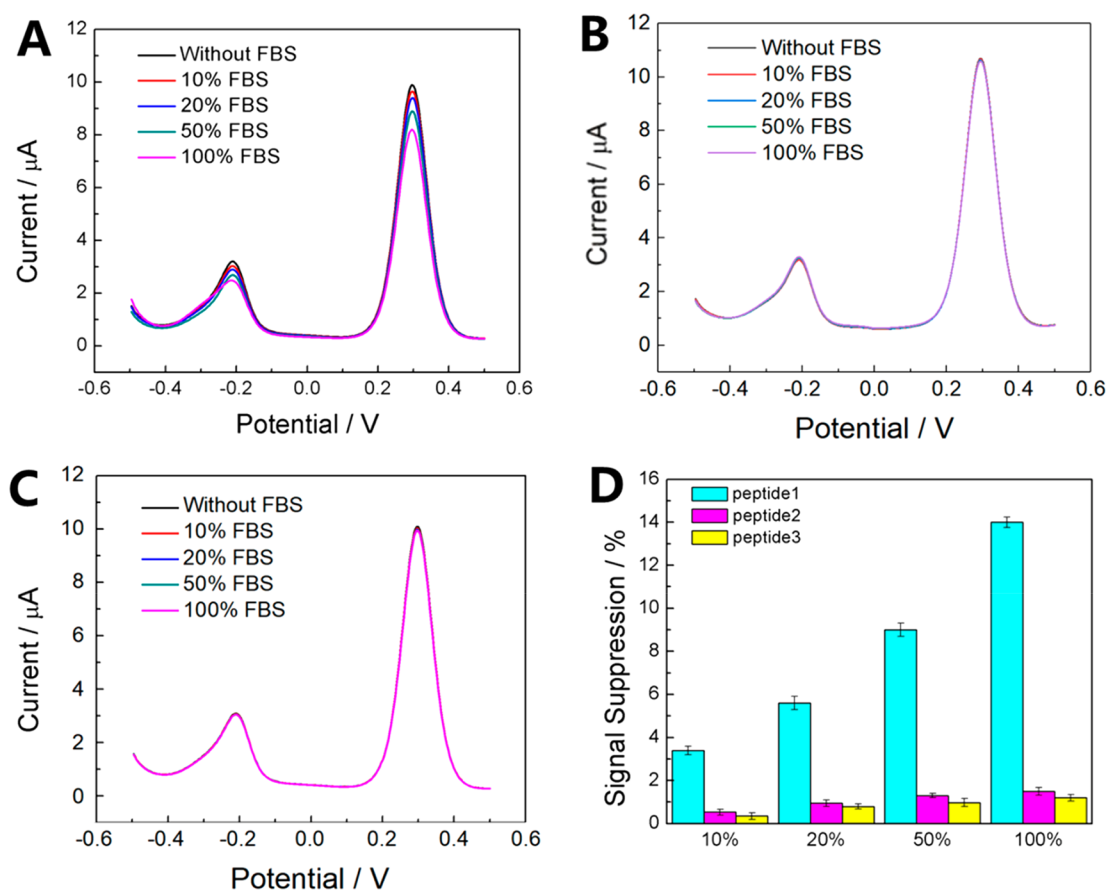


Figure 2. DPV responses in PBS (0.1 M, pH 7.4) of biosensor with (A) peptide1 (general peptide), (B) peptide2 (antifouling peptide) and (C) peptide3 (cleaved antifouling peptide) before and after incubation in FBS samples with different concentrations. (D) Antifouling performance analysis histogram of biosensor with peptide1 (blue), peptide2 (pink), and peptide3 (yellow) based on the signal change of Fc.

MXene with an accordion-like multilayer structure (Figure 1B, E). After in situ synthesis of AuNPs, a large number of AuNPs (Figure 1C, F) with a size of about 50 nm can be seen on the surface of MXene material. The composition of Ti_3AlC_2 before and after HF treatment and the synthesis of MXene-Au are illustrated by EDS. As shown in Figure 1G, Al element that exists in Ti_3AlC_2 (black line) decreases significantly and the F element appears (red line) after HF etching. When AuNPs are in situ synthesized to the surface of MXene, the Au element appears (blue line), demonstrating the successful synthesis of MXene-Au. Meanwhile, the phase transition was recorded in XRD mode as shown in Figure 1H. Interestingly, after etching, the peak intensity of the formed MXene becomes extremely weak, indicating that the crystallinity and configuration deformation are significantly reduced compared to the standard comparison card of Ti_3AlC_2 . In addition, the (002) peak moves to the lower angle and becomes wider, which is consistent with the results in the literature.⁴³ In addition, the MXene-Au-MB nanocomposite was characterized by fluorescence spectroscopy, as shown in Figure S2. MXene-Au is nonfluorescent, whereas 10 μM MB detects a fluorescent signal at the emission wavelength of 675 nm. Different amounts of MXene-Au were added to 1 mL of 10 μM MB solution to fully react and then centrifuged to obtain the supernatant, and the fluorescence intensity of the supernatant was detected. The fluorescence intensity of the supernatant directly reflects the adsorption capacity of MXene-Au to MB. With the increase in the amount of MXene-Au added, the fluorescence intensity of

the supernatant gradually decreased, indicating that MB was effectively loaded in the MXene-Au nanomaterials through strong electrostatic interaction, indicating that MXene-Au-MB was successfully prepared.

As shown in Figure 1I, the electrochemical behavior of MXene-Au is characterized by CV. Obviously, modification of MXene increases the redox peak current and the introduction of AuNPs further elevates the redox peak current. This is because both MXene and AuNPs can effectively increase the conductivity of the electrode. The specific surface areas of bare GCE, MXene/GCE, and MXene-Au/GCE are measured using an electrochemical method and shown in Figure S3. The cyclic voltammetry curves of different modified electrode surfaces at different scanning speeds in 5 mM $[\text{Ru}(\text{NH}_3)_6]^{3+}$ (pH 7.4, 25 $^\circ\text{C}$) were studied. According to the Randles-Sevcik equation, the electrochemical active area is calculated: $I_p = 2.69 \times 10^5 A D^{1/2} \nu^{1/2} c$. Here, I_p is the peak current (in A), A is the electrochemical active area (in cm^2), D is the diffusion coefficient of 5 mM $[\text{Ru}(\text{NH}_3)_6]^{3+}$ ($6.0 \times 10^{-6} \text{ cm}^2/\text{s}$), c is the concentration of $[\text{Ru}(\text{NH}_3)_6]^{3+}$ ($5 \times 10^{-3} \text{ M}$), and ν is the scan rate (in V/s). Through calculation, $A_{(\text{bare GCE})}$ is 0.06163 cm^2 , $A_{(\text{MXene/GCE})}$ is 0.21066 cm^2 , and $A_{(\text{MXene-Au/GCE})}$ is 0.42277 cm^2 , the electrochemically active area has a significant increase. The excellent conductivity and large specific surface area are both beneficial to enhance the sensing performance and sensitivity.

Characterization of the Developed Biosensing Interfaces. The construction of the sensing interface and the

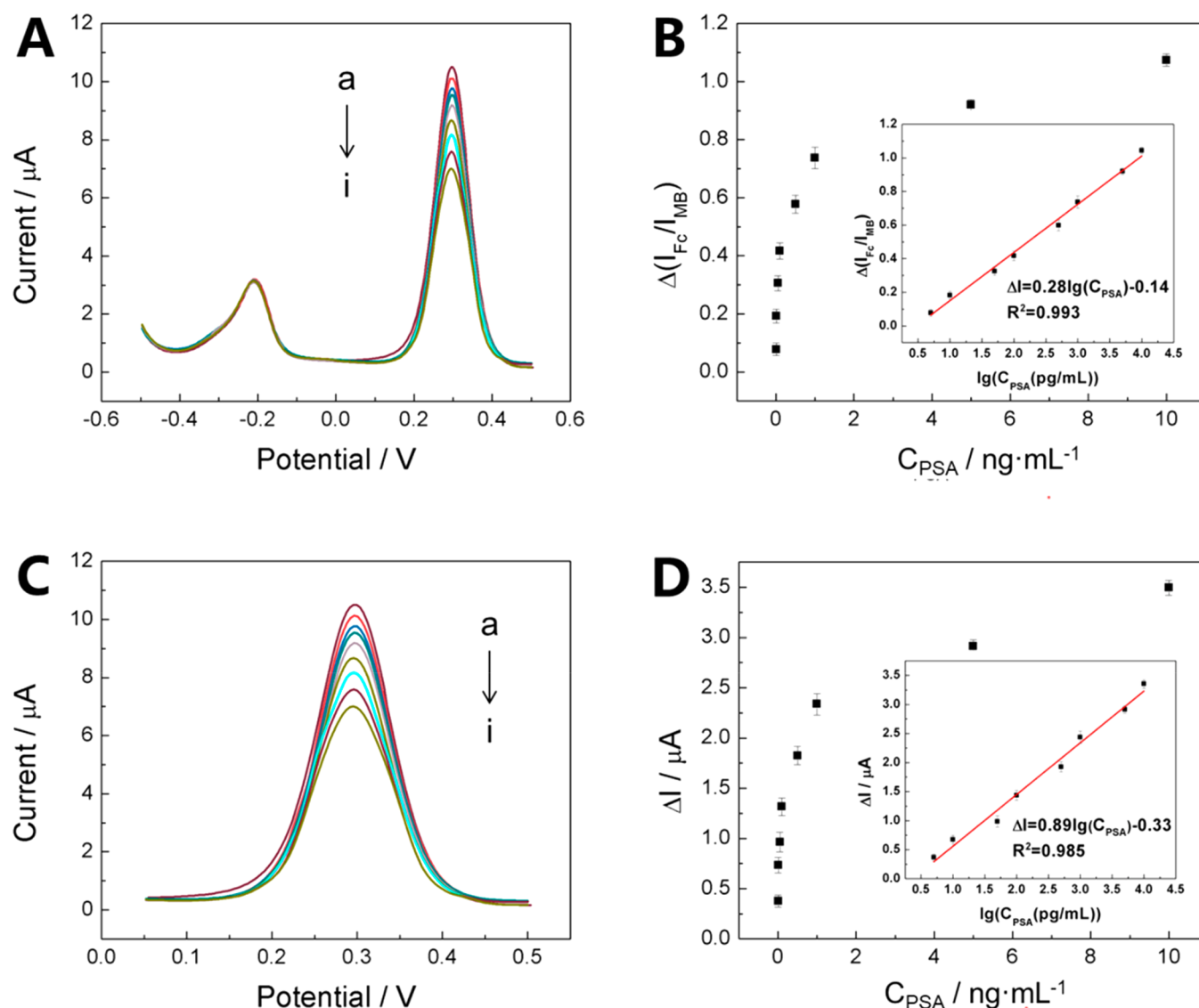


Figure 3. (A) In the presence of an internal standard, the DPV response of the sensing interface under different PSA concentrations: (a) 0 pg/mL, (b) 5 pg/mL, (c) 10 pg/mL, (d) 50 pg/mL, (e) 0.1 ng/mL, (f) 0.5 ng/mL, (g) 1 ng/mL, (h) 5 ng/mL, and (i) 10 ng/mL. (B) PSA calibration curve with internal standard. (C) Sensing interface responds to the DPV signal of PSA without the internal standard. (D) PSA calibration curve without internal standard.

feasibility of the sensor were evaluated by ESI and DPV as shown in Figure S4. The diameter of the semicircle in the impedance spectrum is equal to the electron transfer resistance (R_{et}), which is used to indicate the electron transfer ability of the redox probe at the electrode interface. It can be observed from Figure S4A that the diameter of the semicircle is significantly reduced after the electrode surface is modified with MXene compared to bare GCE. After modifying MXene-Au-MB, the diameter of the semicircle is further reduced. This indicates that MXene and AuNPs can effectively accelerate electron transfer between the redox probe and the electrodes. After modifying the peptide, the impedance increased significantly because of the nonconductivity of the peptide. The conductivity of Fc tagged on the peptide can almost be ignored compared to the nonconductivity of the peptide. From Figure S4B, we can see that the DPV curves of bare GCE (black line) and MXene-Au/GCE (yellow line) have no oxidation peak in PBS solution. After the MXene-Au-MB nanocomposite is fixed on the electrode surface, the oxidation

peak of MB can be detected at about -0.2 V (red line). The peptide is then anchored on the electrode surface, Fc is captured, and the oxidation peak of Fc appeared at the position of $+0.3$ V (blue line), which proves the successful construction of the sensing interface. After incubating the modified electrode in PSA solution for a period of time, it can be found that the DPV signal of Fc was significantly reduced (green line) indicating the releasing of Fc induces by the recognizing and cutting of the peptide sequence by PSA. From the results in Figure S4B, we can see that the DPV signal of MB remains almost unchanged. All of the results prove that the electrochemical biosensors are constructed as expected.

The static water contact angle is used to evaluate and confirm the hydrophilicity and wettability of the electrode surface. From Figure S5 and Table S1, we can see that the MXene-Au-MB has good hydrophilicity because of its large number of hydroxyl groups. Compared to bare GCE, the water contact angle of MXene-Au-MB/GCE is reduced from 57.29° to 28.98° . When the peptide and Fc are immobilized on the

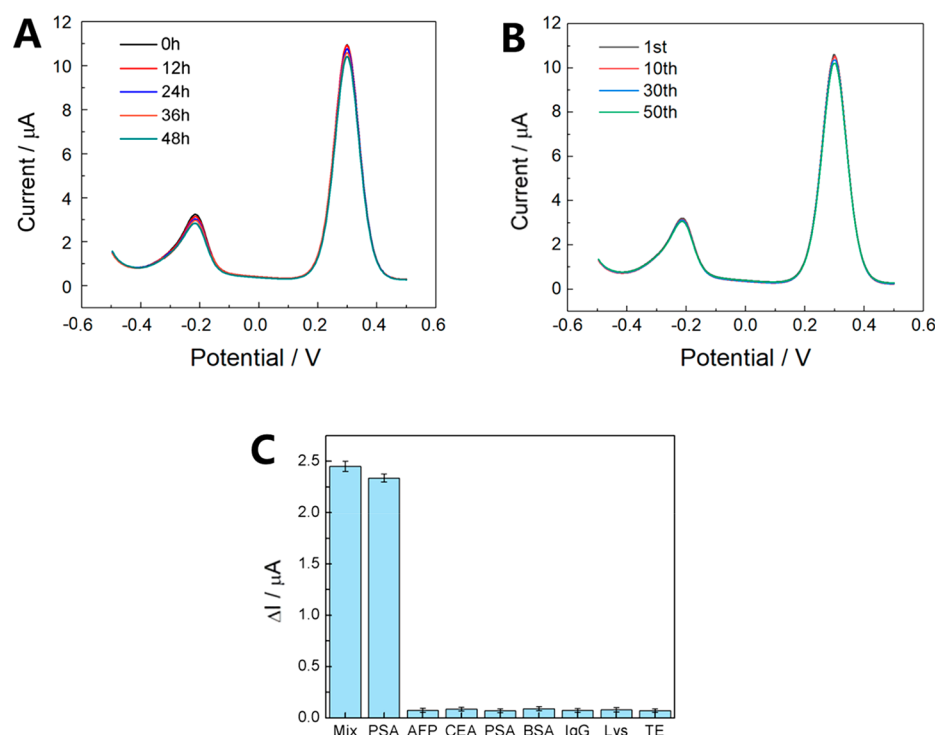


Figure 4. (A) DPV responses of biosensor after immersing in PBS (0.1 M, pH 7.4) for 0, 12, 24, 36, and 48 h. (B) DPV responses of the biosensor after 1, 10, 30, and 50 consecutive scans in PBS (0.1 M, pH 7.4). (C) Selectivity of the biosensor for PSA.

electrode surface, the water contact angle is further reduced to 21.07° because of the excellent hydrophilicity of the peptide (in fact, peptide1, 2, 3 are hydrophilic, with almost no difference in contact angle). Excellent hydrophilic interface is one of the important factors for the antifouling of the sensor.

In addition to excellent hydrophilicity, electrical neutrality is another important factor for antifouling biosensors. The zeta potential of the peptide in Figure S6 showed that the designed peptide shows good electrical neutrality before (peptide2) and after (peptide3) cutting.

Generic Antifouling Characteristics. To evaluate the antifouling ability of the sensing surface, we used three kinds of peptides to modify the electrode. One is the general peptide (peptide1), one is the antifouling peptide (peptide2), and one is the cleaved antifouling peptide (peptide3). Different electrode surfaces were incubated in fetal bovine serum (FBS) for 30 min, and the nonspecific adsorption capacity of different interfaces was compared by DPV method as shown in Figure 2A–C. By analyzing the changes of the Fc signal, it can be seen from Figure 2D that after incubation in 100% FBS, the DPV signal of biosensor with general peptide modification decreased by 14.06%, because peptide1 is positively charged (Figure S6A) and contains hydrophobic amino acids (L, H), it may adsorb proteins. And the DPV signal of biosensor with antifouling peptide modification decreased by 1.55% and the DPV signal of biosensor with cleaved antifouling peptide modification also only decreased by 1.2%. These results are the perfect evidence of the antifouling performance of the biosensor due to the good resistance to nonspecific adsorption of the antifouling peptide.

Electrochemical Detection of PSA and TB with Constructed Biosensor. It can be observed from Figure 3A that the Fc DPV signal is significantly reduced with increasing PSA concentration, whereas the MB signal changes

slightly. This result is consistent with expectation of the internal standard method. The biosensor has a wider detection range and a lower detection line (Table S3). As shown in Figure 3A, B, a correlation coefficient of 0.993 is shown in the detection range of 5 pg/mL to 10 ng/mL. The corresponding regression equation is $\Delta I = 0.28 \lg(C_{\text{PSA}}) - 0.14$ and the corresponding detection limit is calculated as 0.83 pg/mL. When there is no internal standard signal molecule, the measured correlation coefficient is 0.985, and the corresponding detection limit is 1.52 pg/mL, which is worse than the biosensor with internal standard signal molecule (Figure 3C, D). From the results, we can see that the measurement accuracy can be improved by the internal standard method. More importantly, by changing the recognition sequence of the peptide, the biosensor also achieves a highly selective and accurate detection of TB. The detection range is 0.1 pM to 10 nM, the detection line is 16.1 fM, and the linear correlation coefficient is 0.995 (Figure S7A, B), which are better than other detection methods (Table S4). In addition, when there is no internal standard, the detection limit of the biosensor for TB is 42.5 fM, and the linear correlation coefficient is 0.981, which is worse than the result when there is an internal standard (Figure S7C, D).

Selectivity and Stability of the Biosensor. The stability of the biosensor is tested using the DPV method. After immersing in PBS buffer for 48 h, the signal still remained 95% of the initial value (Figure 4A). As shown in Figure 4B, after DPV scans 50 times continuously, the DPV signal almost remains unchanged. Bovine serum albumin (BSA), alpha fetoprotein (AFP), thrombin (TB), carcinoembryonic antigen (CEA), and immunoglobulin G (IgG) are used as possible interferences to evaluate the selectivity of the electrochemical biosensor. Under the same conditions, 10.0 ng/mL BSA, AFP, TB, CEA, IgG, Lys, and TE, 1.0 ng/mL PSA and all mixtures

were incubated with the biosensor, respectively. As shown in Figure 4C, the DPV signal (ΔI) increases greatly only when PSA exists ($\Delta I = I_0 - I_1$, I_0 is the current value of Fc before reaction and I_1 is the current value of Fc after reaction). With the same results, the excellent selectivity of the biosensor for TB detection is shown in Figure S8. All of the above proves that the biosensor has excellent stability, internal reproducibility, and selectivity.

Clinical Serum Sample Analysis. The biosensor was used to analyze and detect PSA clinical serum samples provided by the Affiliated Hospital of Heze Medical College. The accuracy of the biosensor for PSA detection in actual samples was compared with or without internal standards. It can be seen from Table 1 that the relative error of the internal standard

Table 1. Analytical Results of PSA in Clinical Serum Samples

sample	ECL method (ng/mL)	this method (ng/mL)	relative error (%)	RSD (%)	method without internal standard (ng/mL)	relative error (%)
1	1.57	1.52	−3.2	3.9	1.43	−8.9
2	5.31	5.18	−2.4	2.1	4.96	−6.6
3	8.23	8.56	4.0	2.4	7.51	−8.7

method (<4.0%) is far less than that without the internal standard method (<8.9%), which indicates that the internal standard method can improve the measurement accuracy. The results are consistent with the data provided by the conventional ECL method in the Affiliated Hospital of Heze Medical College, which shows that its clinical application is feasible. By adding the standard recovery method, the biosensor is also suitable for detecting TB in human serum samples with relative error less than 3.5% (Table S2).

CONCLUSIONS

A novel nanocomposite of MXene loaded with Au nanoparticles and methylene blue was successfully prepared and used for the construction of ratiometric antifouling electrochemical biosensors together with multifunctional peptides. The MXene-Au-MB possesses excellent electrochemical activity, and the Au nanoparticles can be used for biomolecules immobilization, whereas the MB can be used as an internal standard to generate reference electrochemical signal, which greatly improves the reproducibility of the sensor and the accuracy of detection. In addition, the designed multifunctional peptides exhibit excellent antifouling capability, which can allow the biosensor to detect targets in complex biological samples with low fouling. More importantly, the biosensing system has been proved to be a universal antifouling detection strategy by changing the recognition sequence of the peptides. The MXene-Au-MB combined with various functional peptides can be used for the construction of different ratiometric antifouling electrochemical biosensors with high sensitivity and accuracy even in complex biological media.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsami.1c04933>.

Materials and instruments, zeta potential, fluorescence spectrum, characterization of specific surface area, ESI

and DPV characterization, static water contact angle, and electrochemical detection of TB (PDF)

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Notes

The authors declare no competing financial interest.

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