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ACS Appl. Mater. Interfaces, **Just Accepted Manuscript** • DOI: 10.1021/acsami.7b11385 • Publication Date (Web): 26 Sep 2017

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**Label-Free Electrochemical Sensing Platform for MicroRNA-21
Detection Using Thionine and Gold Nanoparticles Co-Functionalized
MoS₂ Nanosheet**

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ABSTRACT

Herein, we demonstrated a label-free and simple electrochemical sensing platform to detect microRNA-21 (miR-21) with highly sensitivity by using MoS₂ nanosheet functionalized with thionine and gold nanoparticles (MoS₂-Thi-AuNPs). Interestingly, thionine (Thi) was used as reducing agent to successfully synthesize MoS₂-Thi-AuNPs nanohybrid and as a signaling molecule to monitor DNA-RNA hybridization, which provided an ideal platform for label-free miR-21 detection. Upon hybridization with miR-21, the formation of DNA-RNA duplex on the electrode would greatly hinder the electron transfer, which caused the electrochemical signal decrease of Thi. After experimental condition optimization, the signal change of peak currents of Thi have a linear relation with the logarithm of miR-21 concentration ranging from 1.0 pM to 10.0 nM and the limit of detection (LOD) was 0.26 pM. Moreover, such biosensor could detect miR-21 in biological samples like human serum with satisfactory results.

KEYWORDS: molybdenum disulfide (MoS₂), gold nanoparticles (AuNPs), thionine (Thi), electrochemical biosensor, microRNA

INTRODUCTION

MicroRNA (miRNA) is a class of non-coding single-stranded RNA molecule with a length of 18-25 nucleotides, which has important regulation effect in diverse biological processes in organisms. Recent studies have demonstrated that miRNA participates in the cell growth, tissue differentiation and other disease state process.^{1, 2} Abnormal expression of miRNA may relate to the existence of cancer disease and even small expression level changes of miRNA could influence to the expression of genes significantly. For example, circulating miRNAs exist in the peripheral blood, which have a close relationship with specific cancer and have been considered as reliable biomarkers for cancer therapy.^{3, 4} In many cancerous tissues like breast cancer, lung adenocarcinomas and hepatocellular carcinoma, the expression level of miRNA-21 (miR-21) is significantly increased.⁵⁻⁷ Therefore, the determination of concentration of miRNA in biological samples has profound importance in pathological and diagnosis studies. However, the simple, precise and low-cost quantification of miRNA is always challenging. The most common methods for miRNA quantification are real-time fluorescent quantitative polymerase chain reaction (RT-PCR),⁸ northern blotting⁹ and microarrays,¹⁰ which are found to be time-consuming and less sensitive.

In recent years, electrochemical sensor is emerging as a powerful tool for biosensing and has attracted great interest on account of its high sensitivity, specificity and inexpensive equipment.^{11, 12} By integrating with unique features and functions of nanomaterials, various electrochemical sensors for miRNAs detection are developed to enhance the electron transferring, improve capture efficient and amplify the sensing signals, such as noble nanoparticles, graphene oxide, DNA nanostructures and nanodiamonds.¹³⁻¹⁷ For example, Rahaie developed an ultrasensitive electrochemical sensor to detect miR-155 by using graphene oxide and gold nanorod with an intercalating label Oracet Blue as signal indicator.¹⁸ Ma developed a sensitive electrochemical biosensor for miRNA detection with horseradish peroxidase (HRP) functionalized graphene quantum dot as sensing element.¹⁹ Xie employed carbon nanotubes, DNA nanostructures and nanodiamonds to immobilize DNA probes and

hybridization chain reaction (HCR) functionalized DNAzyme nanowires as signal indicator.¹⁷ However, most of the works adopted two or more steps to capture target miRNA and introduced signal indicator, which could complicate the detection procedure during the process. Therefore, label-free biosensors have attracted wide attention because it simplified the assembling processes and provided an effective way for target sensing. Some researchers employed electrochemical impedance spectroscopy (EIS) measurements^{21,22} to directly monitor the surface of the electrode. However, EIS results were always easily disturbed by surface contamination and adsorption. Therefore, it is significant to develop a simple, efficient and label-free platform to detect miRNA by using electrochemical indicator-functionalized nanomaterials with high sensitivity and selectivity.

Molybdenum disulfide (MoS₂) is a new two-dimensional lamellar nanomaterials with graphene-like lamellar structure, novel physical photoelectric and chemical properties. In recent years, MoS₂ has become an ideal base material due to its excellent performance in easy modification and super-large specific surface area, which could react with abundant nanomaterials (like noble metals, metal oxides) and organic molecules to form novel nanocomposite. Various biosensor platforms based on MoS₂ were built to achieve highly sensitive detection for biological or chemical molecules. For instance, Zhang's group reported the first example of the target-induced displacement assay for DNA and small molecules detection in solution based on the fluorescence quenching capacity of MoS₂ nanosheet.²² Lin and Li's group developed the electrochemical biosensor for the direct detection of H₂O₂ by taking the advantage of the inherent electrocatalytic activity of thin-layer MoS₂.²³ Yang and Jiao's group reported an electrochemical assay for direct detection of DNA based on the adsorption between single-strand/double-strand DNA and thin-layer MoS₂ nanosheets. An ultrahigh sensitivity could be achieved by taking the advantage of large surface area of MoS₂.²⁴ To further expand the properties of MoS₂, our group have developed a series of electrochemical biosensors based on metal-decorated MoS₂ for the detection of various biomolecules, like protein, dopamine, adenosine triphosphate and thrombin with high sensitivity and selectivity.²⁵⁻²⁹

Herein, we demonstrated a simple and rapid hybridization assay for label-free miR-21 detection based on our rationally designed MoS₂ nanosheet modified with thionine and gold nanoparticles (MoS₂-Thi-AuNPs). DNA worked as recognition probes and thionine (Thi) was employed as an electrochemical indicator for subsequent electrochemical detection and reducing agent for the reduction of HAuCl₄. The MoS₂-Thi-AuNPs nanocomposite were synthesized by microwave-assisted hydrothermal method²⁹ and then assembled onto the glassy carbon electrode (GCE) by adsorption (Figure 1). After that, capture DNA was assembled on the nanosheet to form recognition layer. Due to the electron transfer between Thi and the electrode, the current peaks of Thi could be obviously observed. After the hybridization with miR-21, duplex DNA-RNA structure was formed, which would greatly hinder the electron transfer on the electrode and caused the signal decrease of Thi. The highly sensitive detection of miR-21 with high specificity was achieved by measuring the decrease of Thi electrochemical signal. The detection limit was calculated to be 0.26 pM (S/N = 3). More importantly, the binding of Au-S was robust and the target hybridization-induced signal change allowed the detection of specific target in the real samples.

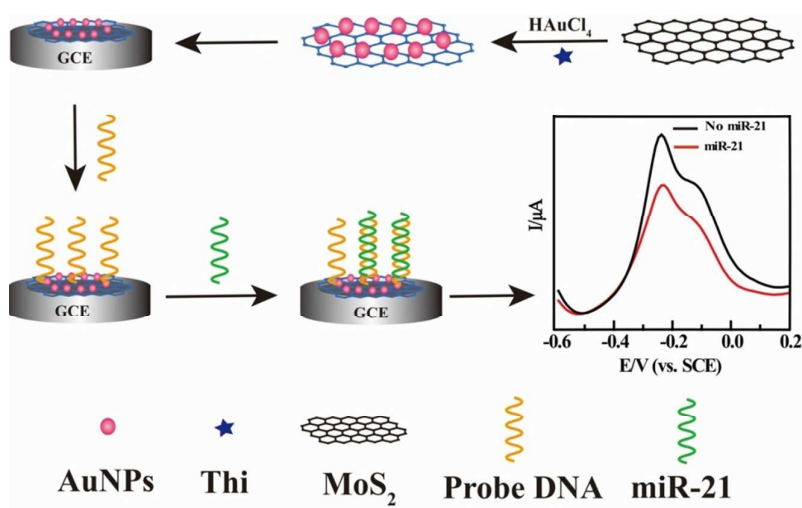


Figure 1. Schematic fabrication of electrochemical platform for miR-21 detection based on MoS₂ nanosheet modified with thionine and gold nanoparticles (MoS₂-Thi-AuNPs).

MATERIALS AND METHODS

Chemicals and Apparatus.

The details information was provided in Supporting Information.

Preparation of MoS₂ nanosheet and MoS₂-Thi-AuNPs nanocomposite.

MoS₂ nanosheet and MoS₂-Thi-AuNPs nanocomposite were synthesized according to our previous works.^{29,30} The details were demonstrated in Supporting Information.

Fabrication of the MoS₂-based electrochemical biosensor.

A glassy carbon electrode (GCE) (3 mm in diameter) was treated with alumina powder, water, ethanol in succession. After being dried under nitrogen, MoS₂-Thi-AuNPs nanocomposite was modified on the electrode, which was defined as MoS₂-Thi-AuNPs/GCE. After that, 1 μM probe DNA was dropped onto the electrode's surface and incubated at 25 °C for 16 h. The unbounded probe DNA was removed from the modified electrode by rinsing 3 times with 10 mM Tris-HCl (pH 7.4). Then, 1 mM MCH was incubated for 60 min in the reaction buffer at 37 °C to block the active surface of electrode. Finally, the as-prepared electrode based on MoS₂-Thi-AuNPs was used to detect target miR-21 with various concentrations under optimal condition.

The detection of miR-21.

The miR-21 solution with different concentrations was firstly diluted by 10 mM Tris-HCl with 140 mM NaCl and 5 mM MgCl₂ (pH=7.4). Then, 5 μL miRNA solution was added to the surface of MoS₂-Thi-AuNPs modified electrode. After 50 minutes hybridization, unreacted miR-21 was removed by using 10 mM Tris-HCl (pH 7.4) for 3 times. The electrochemical detection was performed in 10 mM Tris-HCl (pH 7.4). For SWV measurement, the detection time was 8 s and the scanning rate was 0.1 V/s and the range was -0.6 V to 0.2 V.

RESULTS AND DISCUSSION

Characterization of MoS₂ nanosheet and MoS₂-Thi-AuNPs nanocomposite.

From the transmission electron microscopy (TEM) image (Figure 2A), a typical wrinkled and thin-layered nanostructure was observed, indicating that few layers

MoS₂ nanosheet was exfoliated successfully. After being reacted with HAuCl₄ and Thi, 40 nm well-dispersed clover-shape AuNPs were formed on the surface of MoS₂ nanosheet (Figure 2B), suggesting that MoS₂-Thi-AuNPs nanocomposite had been successfully synthesized. The size statistical analysis indicated that AuNPs' diameter was normal distributed with ~40 nm in average (Figure S1). Furthermore, we employed UV-vis spectrum to record the process of reaction and the result was shown in Figure S2A. Two board peaks located ranging from 240 to 290 nm and from 530 to 650 nm of MoS₂-Thi-AuNPs were observed in the spectra, which were corresponding to the peaks of MoS₂ nanosheet, Thi and gold nanoparticles. Meanwhile, the color solution of MoS₂-Thi-AuNPs nanocomposite presented to be light grayish green, which was different from that of MoS₂ nanosheet (light brown) and Thi (blue) solution (Figure S2B). The scanning electron microscope (SEM) image and energy dispersive spectrometer (EDS) results confirmed that the Mo, N, S and Au elements were existed in the nanocomposite, proving that AuNPs had successfully supported on the MoS₂ nanosheet (Figure S3). Furthermore, the data of X-Ray Diffraction (XRD, Figure S4) and X-ray photoelectron spectroscopy (XPS, Figure S5) also proved the existence of Thi and AuNPs on MoS₂ nanosheet's surface. All data showed that the MoS₂-Thi-AuNPs nanocomposite was successfully prepared.

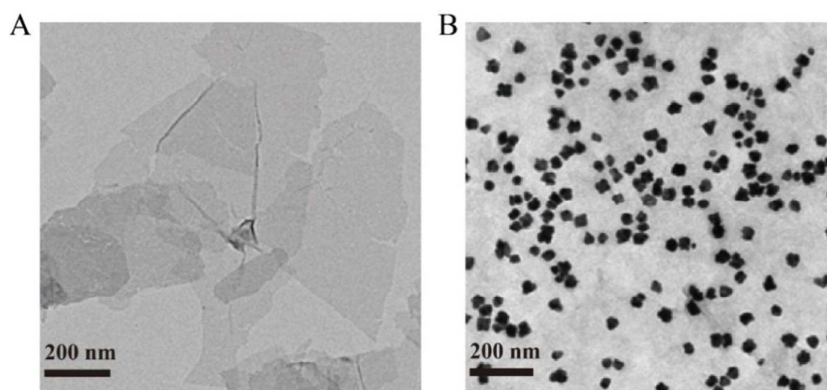


Figure 2. Transmission electron microscopy (TEM) imaging of (A) as-prepared MoS₂ nanosheets and (B) synthetic MoS₂-Thi-AuNPs nanocomposite.

Electrochemical characterization of MoS₂-Thi-AuNPs/GCE biosensor.

The modification steps of the MoS₂-based biosensor was characterized by electrochemical impedance spectroscopy (EIS) technique in 5 mM [Fe(CN)₆]^{3-/4-} and

0.1 M KCl (Figure 3A). The electron-transfer resistance (R_{et}) of each modified electrodes was estimated from the scale of the semicircle. The result depicted that the R_{et} of bare GCE was only 78 Ω (curve a). If MoS₂-Thi-AuNPs was successfully immobilized onto the surface of GCE (curve b), the R_{et} was increased significantly to 192 Ω . After being functionalized with probe DNA, the R_{et} of the electrode was increased to 305 Ω (curve c) because DNA was negative charge. Then, the R_{et} was further increased to about 343 Ω (curve d) after MCH blocking the remaining active sites of the electrode's surface. A larger R_{et} (592 Ω) was then obtained after miR-21 be recognized by probe DNA (curve e), suggesting that the double-strand structure of DNA-RNA duplex hindered the electron transfer. The EIS results showed that the MoS₂-based platform was successfully constructed and the designed detection strategy was worked.

Thi is a commonly used electrochemical indicator, which is employed as signal indicator to test the detection feasibility in this work. Figure 3B demonstrated the square wave voltammetry (SWV) curves of different modified electrodes in 10 mM Tris-HCl (pH 7.4). For bare GCE, no obvious peak was observed (curve a in Figure 3B). After being modified with MoS₂-Thi-AuNPs nanocomposite, a distinct anodic peak current of Thi at -0.23 V was observed (curve b in Figure 3B), indicating that Thi in the nanocomposite still had good electrochemical activity. After being functionalized by probe DNA, the current value of Thi was greatly decreased (curve c in Figure 3B) because the assembled DNA hindered the electron transfer on the electrode. Then the current was continue to decline after MCH blocking (curve d in Figure 3B) and miR-21 hybridization (curve e in Figure 3B), respectively. This phenomenon was corresponding with the results of EIS test, which further proved that the designed sensing platform had a good performance in target recognition and signal indicating.

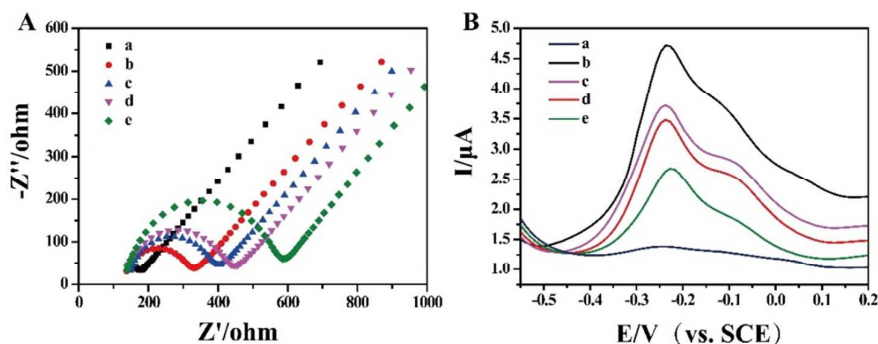


Figure 3. (A) Nyquist plots and (B) SWV curves of (a) bare GCE, (b) MoS_2 -Thi-AuNPs/GCE, (c) DNA/ MoS_2 -Thi-AuNPs/GCE, (d) MCH/DNA/ MoS_2 -Thi-AuNPs/GCE, and (e) miR-21/MCH/DNA/ MoS_2 -Thi-AuNPs/GCE in 5 mM $\text{K}_3\text{Fe}(\text{CN})_6/\text{K}_4\text{Fe}(\text{CN})_6$ (1:1) solution containing 0.1 M KCl and 10 mM Tris-HCl (pH 7.4), respectively. The concentration of probe DNA and miR-21 was 100 nM and 10 nM, respectively.

Condition optimization for detection.

To obtain the optimal reaction condition, the concentration of the probe DNA, the incubation time of probe DNA and the hybridization time between probe DNA and miR-21 were tested because these factors would directly affect the hybridization efficiency of the biosensor. The peak current decrement of Thi was defined as $\Delta I = I_{\text{no miR-21}} - I_{\text{miR-21}}$, while $I_{\text{no miR-21}}$ was the peak current at -0.23 V without miR-21 and $I_{\text{miR-21}}$ was the peak current with target miR-21. The probe concentration from 0.1 nM to 10 μM was tested and 1 μM was optimized as the loading concentration of probe DNA (Figure S6). Obviously, a maximum difference of peak current was observed when the incubation time was 16 h and did not change obviously after 16 h (Figure 4A). Therefore, we chose 16 h as the optimal incubation time for the detection of miR-21. Similarly, 50 min was selected as the hybridization time between probe DNA and miR-21. (Figure 4B).

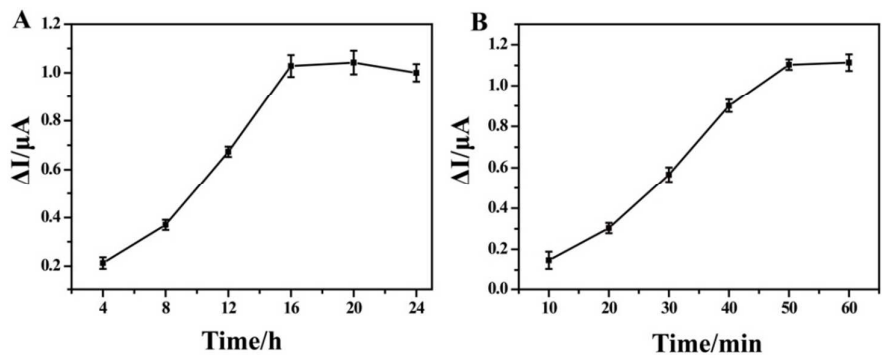


Figure 4. (A) The optimization for the incubation time of probe DNA from 4 to 24 h (4 h, 8 h, 12 h, 16 h, 20 h and 24 h) and (B) The optimization for the hybridization time between probe DNA and miR-21 from 10 to 60 min (10 min, 20 min, 30 min, 40 min, 50 min and 60 min).

Analytical performance of MoS₂-based biosensor.

Under the optimal conditions, SWV detection was adopted to investigate the detection performance of the MoS₂-based biosensor (Figure 5). Once the addition of miR-21, the current of Thi at -0.23 V was decreased (Figure 5A) and the dynamic range for miR-21 detection was checked. The peak currents variation of the biosensor kept a linear relation with the logarithm of the concentration of miR-21 from 1.0 pM-10 nM (Figure 5B). The obtained equation was $\Delta I (\mu A) = 0.204 \log_{10} c_{miR-21} (pM) + 0.279$ ($R^2=0.996$) and the LOD was calculated to be 0.26 pM ($S/N=3$). The detection performance was better than or comparable to some previous works,³¹⁻³⁶ which was listed in Table S2. The high sensitivity may attribute to the large surface area of MoS₂-Thi-AuNPs nanocomposites,²⁹ which allows the loading of large amounts of probe DNA for target miR-21 capturing and Thi for electrochemical sensing. Moreover, the AuNPs supported on MoS₂ was not only convenient for the immobilization of probe DNA through Au-S bond, but also increase the electronic conductivity of electrode,^{37, 38} which is beneficial to increase the sensitivity of biosensor. The experimental data suggested that the MoS₂-Thi-AuNPs nanocomposite might be a potential candidate to construct electrochemical platform for target molecules detection.

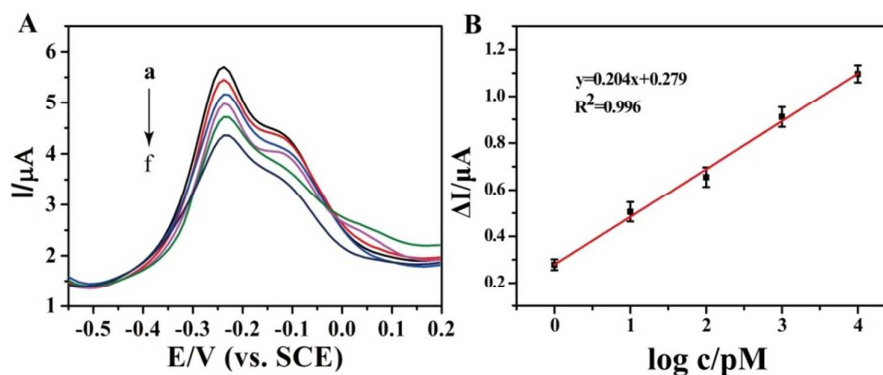


Figure 5. (A) SWV curves of the biosensor responded to various miR-21 concentrations (a-f: 0, 1.0, 10, 100, 1000 and 10000 pM). (B) Regression curves for miR-21 detection.

Selectivity test.

For the selectivity test, single-base terminal mismatched miRNA (SM-T), single-base middle mismatched miRNA (SM-M) and non-complementary miRNA (NC) were analyzed under the same condition. The SWV curves was recorded in Figure 6A. The calculated ΔI of the MoS₂-based biosensor for target miRNA detection was 1.9 times, 2.6 times and 20.0 times than that for SM-T, SM-M, NC, respectively (Figure 6B). It should be noted that the as-prepared MoS₂-based biosensor could effectively discriminate single-base terminal mismatched from single-base middle mismatch miRNA sequence. The result suggests that the mismatch site on terminal or in middle of the miRNA strand have a strong impact on the electrochemical signals. The reason may be that the DNA-RNA hybridization occurs to a lower extent for middle-mismatched sequence than that for terminal-mismatched sequence.^{39, 40} All data indicated that the biosensor possessed excellent selectivity and discriminated ability.

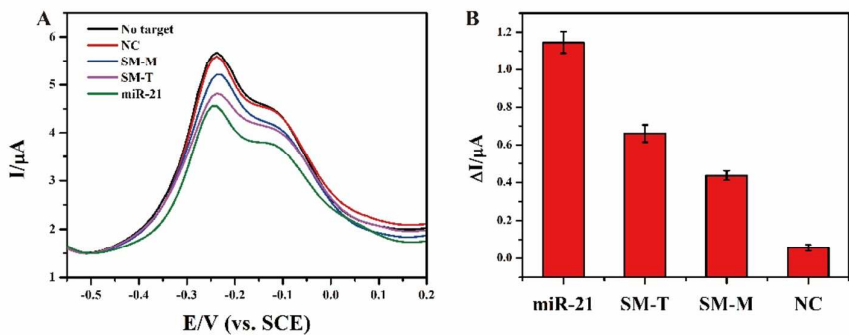


Figure 6. (A) Typical SWV curves of the MoS₂-based biosensor for the same concentration (10 nM) of target miR-21, single-base terminal mismatched miRNA (SM-T), single-base middle mismatched miRNA (SM-M) and non-complementary miRNA (NC) detection. (B) ΔI responses for miR-21 detection.

Real samples analysis.

To test the target miR-21 detection in biological samples, method of standard addition was adopted to analyze miR-21 in human serum. Various amounts of miR-21 (1 pM, 10 pM and 1000 pM) were spiked into 1% human serum. Each sample was then analyzed with the biosensors. As shown in Table 1, the relative standard deviation (RSD) of this biosensor was less than 7% and the recovery was more than 96%, indicating that the MoS₂-based biosensor had a potential application in real biological samples.

Table 1. Analysis data of the MoS₂-based sensor for miR-21 determination in human serum.

Sample	Added (pM)	Found (pM)	RSD (%)	Recovery (%)
1	1	1.03, 0.94, 1.07	6.57	101.0
2	10	9.26, 9.86, 9.74	3.30	96.2
3	1000	987.79, 1026.32, 976.56	2.61	99.6

CONCLUSIONS

In conclusion, we successfully fabricated a simple and efficient electrochemical platform for label-free detection of miR-21 based on a MoS₂-Thi-AuNPs nanocomposite, in which Thi was used as reducing agent for the reduction of HAuCl₄

and electrochemical signal indicator for miR-21 detection. It does not require the labeling of DNA or RNA strands and the whole detection time was within 1 h. Under optimal conditions, the biosensor had a wide linear range from 1.0 pM to 10 nM) and a low detection limit (0.26 pM). Moreover, the biosensor could effectively distinguish single-base mismatched miRNA from target miR-21. As expectedly, such MoS₂-based platform could determine miR-21 in real sample with satisfactory results, suggesting the MoS₂-based biosensor offered potential application in cancer-related biomolecules analysis.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge via the Internet at <http://pubs.acs.org>.

The synthesis process of MoS₂ nanosheet and MoS₂-Thi-AuNPs nanocomposite, UV-vis absorption spectra and photo images of MoS₂ nanosheet, Thi and MoS₂-Thi-AuNPs nanocomposite, and the comparison of different nanomaterials-based electrochemical sensors for miRNA detection were listed in Supporting Information.

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Notes

The authors declare no competing financial interest.

ACKNOWLEDGEMENTS

This work was financially supported by the Key Research Program of National Nanotechnology and Science (2017YFA0205302) the National Natural Science Foundation of China (21475064, 21605087, 21373260 and 21305070), the Program for Changjiang Scholars and Innovative Research Team in University (IRT_15R37),

the Priority Academic Program Development of Jiangsu Higher Education Institutions (PAPD, YX03002), the China Postdoctoral Science Foundation funded project (BX201700123), the Scientific Research Foundation of Nanjing University of Posts and Telecommunications (NY215058), and the Natural Science Fund for Colleges and Universities in Jiangsu Province (16KJB150032).

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