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Efficient strand displacement amplification via stepwise
movement of bipedal DNA walker on electrode surface
for ultrasensitive detection of antibiotic

View Article Online
DOI: 10.1039/D0AN00139B

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21

22 **Abstract**

23 DNA walkers, one of the artificial molecular machines which are constructed via
24 smart synthetic DNA, have attracted rapidly growing attention of researchers in
25 biosensing fields. In this work, we design an Exonuclease III (Exo III)-aided
26 target-aptamer binding recycling (ETBR) activated bipedal DNA machine for highly
27 sensitive electrochemical detection of antibiotic. To the best of our knowledge, this is
28 the first time that the bipedal DNA machine has been applied in electrochemical
29 sensing for antibiotics. On the one hand, the bipedal DNA walker exceeds the
30 conventional single swing arm DNA walker in terms of walking efficiency and
31 stability. On the other hand, ETBR strategy, along with the efficient strand
32 displacement amplification via stepwise movement of bipedal DNA walker
33 significantly promotes the signal amplification efficiency. Under optimal conditions,
34 this bipedal DNA machine possesses a detection limit to 7.1 fM within a linear
35 detection range from 10 fM to 100 pM. Moreover, this electrochemical biosensor is
36 expected to detect wide variety of analytes by using corresponding target recognition
37 probes. Thus, our proposed strategy offers a highly efficient, stable and practical
38 platform for small molecule analysis.

39 **Keywords:** Bipedal DNA walker; strand displacement amplification; electrochemical
40 biosensor; antibiotic

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1. Introduction

Antibiotics, as an effective medicine, are widely applied in inhibiting the growth and reproduction of bacteria.¹ Nevertheless, the misuse of antibiotics will inevitably lead to antibiotic residue and negative effects, such as nephrotoxicity, allergic reactions and antibiotic resistance.²⁻⁴ Therefore, developing a rapid and efficient strategy for the detection of antibiotics in the water and foodstuffs is highly advisable.

Nowadays, scholars have exploited multiple instrumental methods to quantify residual antibiotics, such as high performance liquid chromatography (HPLC),⁵ surface-enhanced Raman scattering (SERS),^{6, 7} surface plasmon resonance (SPR).⁸⁻¹⁰ However, these instrumental methods have inherent disadvantages such as poor specificity, expensive equipment and complicated operation. Thus, it is of great research significance to develop a fast and efficient method for quantitative analysis of antibiotics.

In this context, a variety of colorimetric,¹¹⁻¹³ fluorescent,¹⁴⁻¹⁶ electrochemical biosensors¹⁷⁻¹⁹ have been applied for the detection of antibiotics and other biomolecules. To improve the sensitivity, many signal amplification strategies including the hybridization chain reaction (HCR)²⁰⁻²², rolling circle amplification (RCA)^{23, 24}, catalyzed hairpin assembly (CHA)^{25, 26} and DNA walking machine²⁷⁻²⁹ have been developed. Among them, DNA walking machines, one of the artificial molecular machines, as the ingenious dynamic interaction and capabilities for signal amplification have been widely integrated into electrochemical biosensing fields. This

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63 is not only because of the advantages of high sensitivity, low cost, and flexible
64 operation of electrochemical strategies, but also the electrodes provide a stable and
65 suitable interface for DNA machines. For example, Chen et al proposed a one-step
66 electrochemical DNA walker for proteins identification based on target-induced
67 proximity recognition.³⁰ Cai's group designed an electrochemical platform through
68 using DNA walker as signal amplification method for the detection of breast cancer
69 cell³¹. However, above DNA machine employed the single swing arm as assembly
70 operation which the walking efficiency will be restricted. To address this problem,
71 some bipedal DNA walker methods were constructed to improve the walking
72 efficiency³². For instance, Yang's group designed a bipedal DNA walker via
73 simultaneous binding of two DNA linked-affinity ligands for sensing thrombin.³³
74 Although bipedal DNA machine can improve walking efficiency to a certain degree,
75 the conventional strategy of a target generated only one DNA walker limited the total
76 number of DNA walkers and their walking speed.

77 In response, we report an efficient strand displacement amplification method by
78 stepwise movement of bipedal DNA walker on the electrode for ultrasensitive
79 detection of antibiotic. To the best of our knowledge, this is the first time that the
80 bipedal DNA machine has been utilized for electrochemical sensing for antibiotics.
81 This proposed strategy features with several aspects. Firstly, ETBR can output
82 countless DW in comparison to the previous way of a target generates only one DNA
83 walker. Secondly, bipedal DNA walker can not only improve amplification

efficiency, but also obviously enhance stability and walking efficiency by simultaneous combining two DNA capture probes. Additionally, by replacing antibiotic aptamer by the corresponding DNA substrates for target recognition, this bipedal DNA machine can readily enable the detection of various targets. Therefore, this electrochemical biosensor provides a stable, universal and efficient method for ultrasensitive antibiotics detection and other molecules with trace amounts in bioanalysis.

2. Experimental

2.1. Reagents and materials

All the antibiotics used in the experiment and 6-mercapto-1-hexanol (MCH) were bought from Aladdin Chemistry Co. Ltd (Shanghai, China). All oligonucleotides were HPLC-purified and synthesized by Sangon Biotechnology Co. Ltd (Shanghai, China). The oligonucleotide sequences were listed in **Table 1**. All other chemical reagents were purchased from Sinopharm Chemical Reagent Co. Ltd (Beijing, China). All solutions were prepared using ultrapure water with an electric resistance of >18.25 M Ω cm obtained through a Millipore Milli-Q water purification system (Billerica, MA, USA). Phosphate Buffered Saline (PBS, 10 mM PB, 0.14 M NaCl, pH 7.4) was used during the experiments. Before the experiment, the S1, S2 and CP were mixed to form S1-S2-CP complex at 95 °C for 5 min and cooled to indoor temperature.

Table 1 Oligonucleotides sequences used in this study.

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DOI: 10.1039/D0AN00139B

Oligonucleotide name	Sequence (5' to 3') description
Hairpin probe 1(HP1)	TAGCCTTTTTTT <u>TGGGGG</u> TTGAGGCTAAGCCGAC
Hairpin probe 2 (HP2)	<i>TTTTTTCATACCTTTATCAGTACCTCTCTATTACACTTTTT</i> <i>TCATACCT</i> ACTGATAAAGGTATGAAAAAAGGCTA
S1	GTCATTCAATCAAGGTGTGCATCTTTT
S2	CGATCGTTTTTTTTTTT
Capture probe (CP)	GGTATGAAAAAACGATCGGATGCACACCTTGATTG AATGACATATAT-SH
Signal probe (SP)	MB-GTCATTCAATCAAGGTGTGCTCCGATCGTTTTTTT CATTTT

The underlined of HP1 is the kanamycin aptamer. The italic of HP2 is the sequence of BW.

2.2 Apparatus and measurements

Cyclic voltammetry (CV), differential pulse voltammetry (DPV) and electrochemical impedance spectroscopy (EIS) were measured with a CHI660D electrochemical workstation bought from Shanghai Huachen Instrument Co. Ltd. We used a classical three-electrode system which consisted of an Ag/AgCl electrode as the reference electrode, platinum wire as the auxiliary electrode and a gold electrode with a diameter of 2 mm as the working electrode. CV and EIS were performed in 5

113 mM $[\text{Fe}(\text{CN})_6]^{3-/4-}$ containing 0.25 M KCl. DPV was performed in PBS (10 mM, pH
114 7.4) in the voltage range from 0 to -0.5 V.

115 2.3 Gold electrode treatment and modification

116 The working electrodes were polished carefully with 0.3 and 0.05 mm alumina
117 suspension. Next, the gold electrodes were immersed in a freshly prepared piranha
118 solution ($\text{H}_2\text{O}_2/\text{H}_2\text{SO}_4$, the volume ratio was 1:3) for 2 h, and then rinsed using
119 ultrapure water thoroughly. Immediately, the gold electrodes were cleaned with
120 ultrasonication in ethanol solution. Then, in the process of electrode modification, 3
121 μL pure of 10 μM the mixture solution S1-S2-CP was modified on the gold electrode
122 via the Au-S bond for 2 h at 37°C. Then, the electrodes were rinsed using PBS to
123 wipe off the unabsorbed S1-S2-CP. Thereafter, the gold electrodes were immersed
124 into a 2 mM MCH solution for 1 h to block the unoccupied sites of the electrode and
125 eliminate nonspecific adsorption. Finally, the working electrodes were completely
126 rinsed with PBS and stored at 4°C.

127 2.4. Kanamycin detection

128 The details of the experiment were as follows: Sample solution of 10 μL
129 kanamycin (the final concentrations ranging from 0 to 100 pM), 3 μL pure of 10 nM
130 (final concentration) HP1, 3 μL pure of 500 nM (final concentration) HP2, 2 μL
131 10×NEBuffer and 2 μL Exo III were mixed at 37°C for 2 h to perform the Exo
132 III-assisted target-aptamer binding recycle. Subsequently, the 20 μL aforementioned
133 solution together with 3 μL pure of 10 μM SP were added to the surface of Au

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134 electrode and incubated in a humid atmosphere at 37 °C for 2 h. Next, the gold
135 electrode was completely rinsed with 1×PBS for three times. Finally, the DPV signal
136 of the biosensor was collected in 15 mL of PBS (10 mM, pH 7.4).

137 **3. Results and discussion**

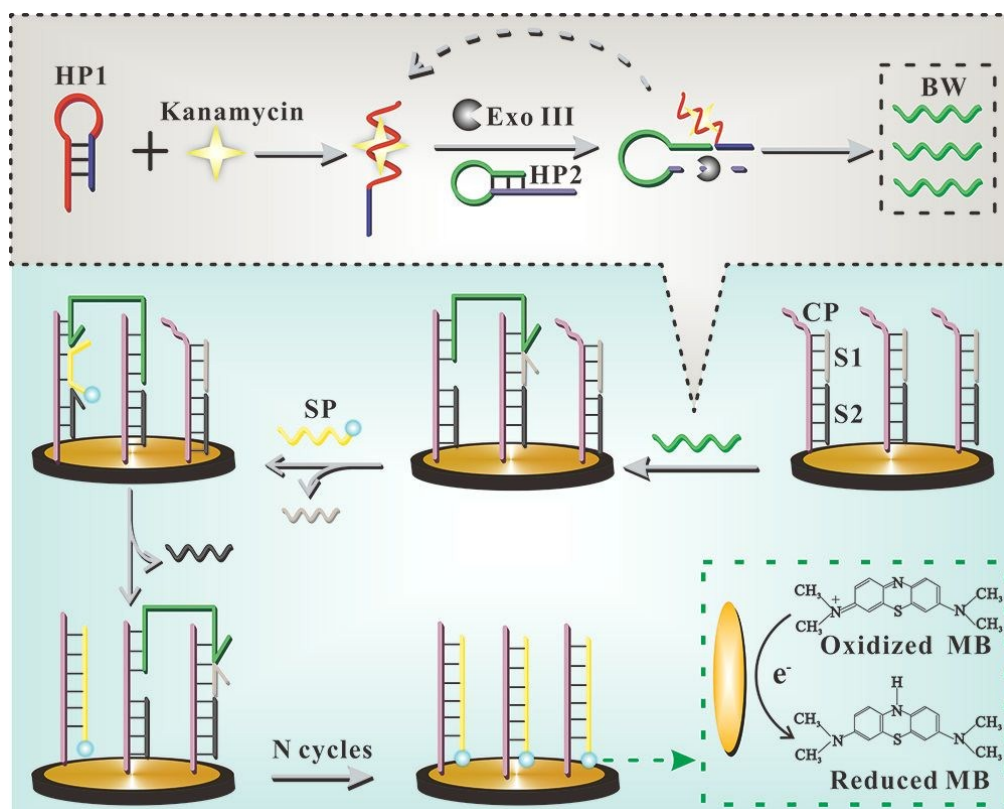
138 *3.1 Principle of the proposed electrochemical biosensor strategy*

139 The detection principle of bipedal DNA machine is depicted in **Scheme 1**.
140 Kanamycin is used as a model analyte. This bipedal DNA machine contains a DNA
141 modified electrochemical sensing interface, namely S1-S2-CP, hairpin probe 1 (HP1)
142 containing the kanamycin aptamer sequence, hairpin probe 2 (HP2), a signal probe
143 (SP) labeled with methylene blue (MB) and Exo III. To achieve high sensitivity and
144 specificity for antibiotics quantification, Exo III-aided target-aptamer binding
145 recycling (ETBR) is used as the first stage of signal amplification. When the target is
146 present, target specifically binds the HP1 and change its conformational, the
147 target-HP1 complex anneal with the protruding tail of HP2 forming DNA duplex with
148 a blunt 3'-terminus. Immediately, Exo III recognizes the duplex and catalyzes the
149 stepwise removal of mononucleotides from the blunt 3' end of HP2 in the 3' to 5'
150 direction, which induces the generation of bipedal walker (BW) and the release of
151 target-HP1. Simultaneously, free target-HP1 continues to hybridize with another HP2,
152 initiating a new cycle. After ETBR process, a mass of BW can be produced in the
153 solution. Subsequently, the resulting BW is dropped on electrode surfaces for efficient
154 strand displacement amplification and electrochemical sensing. The bipedal DNA

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machine is activated with the combination of BW and CP, while releases S1.

Ultimately, SP interacts with CP and leads to the release of S2, accompanying in the complete hybridization of CP and SP concurrent restoration of initially BW. Meanwhile, the free BW moves autonomously upon combining with surplus CP, accordingly, numerous SP gathered on electrode surfaces and a dramatically strong DPV signal is obtained due to the reduction of MB. In summary, the electrochemical biosensor provides a stable, universal and efficient strategy for kanamycin identification and related small molecule analysis.



Scheme 1. Schematic illustration of the biosensor for kanamycin detection.

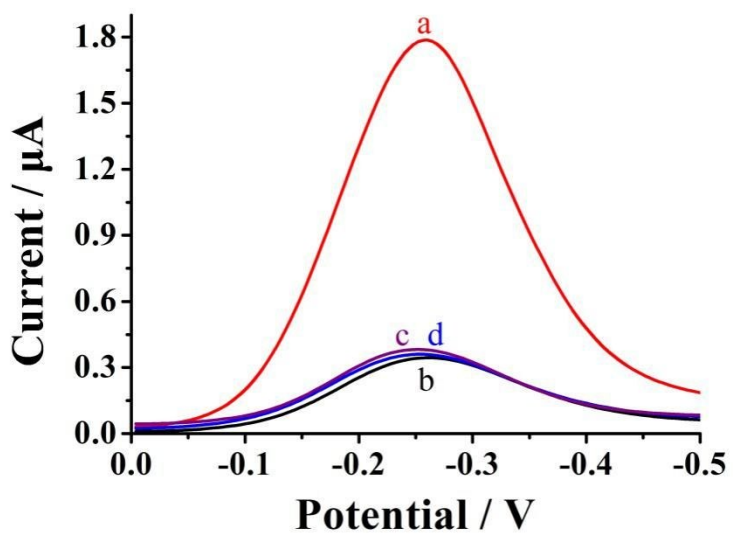
3.2 Feasibility analysis of bipedal DNA machine

To confirm the feasibility of our bipedal DNA machine, a series of control

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167 experiments were investigated, and these results were displayed in **Fig.1**. For the
168 positive sample, a significantly strong DPV signal was obtained (curve a), indicating
169 that numerous SP combined to the CP on the electrode and accelerated the electron
170 transfer between the electrode and MB. In contrast, it was found a negligible current
171 signal for the blank sample (curve b). This result proved the activation of DNA
172 machine relied on the specific kanamycin-aptamer interaction. When there is no Exo
173 III in the reaction solution, we observed a slightly weaker electrochemical response
174 (curve c), implying Exo III was essential for the operation of bipedal DNA walker. In
175 addition, when kanamycin was replaced by terramycin, a low DPV response similar to
176 the blank sample was collected (curve d), which proved the high specificity of this
177 electrochemical biosensor for kanamycin detection.



178
179 **Fig.1.** DPV signal of the electrochemical DNA machine collected by analyzing 100
180 pM kanamycin (a). Curves b to d were for the control experiments which were
181 performed with blank sample (b), 100 pM kanamycin without Exo III (c), kanamycin

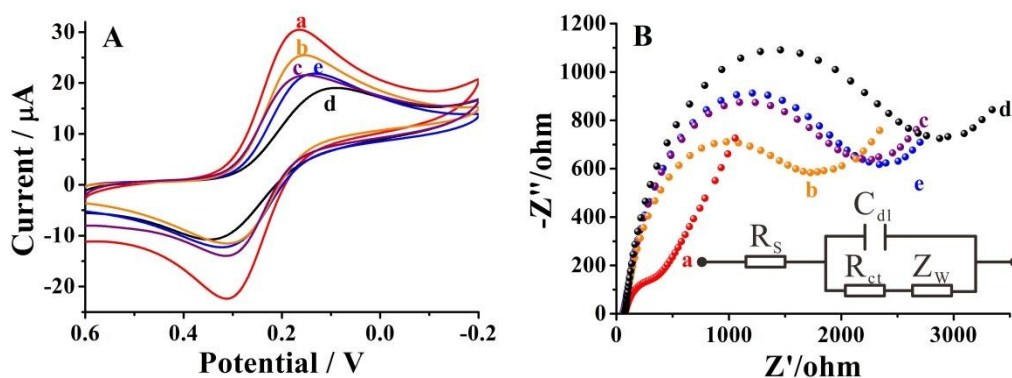
was replaced by 100 pM non-target terramycin (d).

3.3 Electrochemical characterization of the modified electrodes

Fabrication of the bipedal DNA machine was further verified by cyclic voltammetry (CV), electrochemical impedance spectroscopy (EIS), respectively. As depicted in Figure 2A, a pair of well-defined current peaks was obtained for bare Au electrode (curve a). Since negatively charged oligonucleotides could repel $[\text{Fe}(\text{CN})_6]^{3-/4-}$, the peaks weakened after the immobilization of S1-S2-CP (curve b). After being blocked with MCH, there was a decreased redox current (curve c), which was attributed that the nonconductive property of MCH was able to impede the electron transfer on electrode surfaces. After the incubation of the modified electrode with kanamycin, the redox current showed noticeable decrease (curve d), suggesting the bipedal DNA machine was activated. Further, after the incubation of the aforementioned electrode with SP, the peak current had a significant increase (curve e), proving SP facilitated the release of BW. All of the above data indicated the fabrication process of the electrochemical sensing interface. Similarly, EIS results were also recorded and summarized the fabrication procedure of the DNA machine in Figure 2B. As shown in Figure 2B, the bare Au electrode showed a greatly small electron transfer resistance (Ret) (curve a). After the immobilization of S1-S2-CP on electrode surface, the Ret displayed a substantial increase, manifesting the negatively charged phosphate backbone excluded the electron transfer on electrode surfaces (curve b). Since the non-specific sites were blocked by MCH, the Ret continued to

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203 increase (curve c). Additionally, with the incubation of kanamycin, the Ret exhibited
204 an obvious increase (curve d), certifying the bipedal DNA walker had been
205 assembled. Next, after the incubation of the modified electrode with SP, there was a
206 certainly decrease for Ret due to SP combined CP to induce the free of BW. The EIS
207 changes were agreement with the CV responses, which forceful described the
208 preparation process of the electrochemical DNA machine interface.



209
210 **Fig.2.** (A) CVs and (B) EIS described for the bare Au electrode (a), CP-S1-S2/Au (b),
211 MCH/CP-S1-S2/Au (c), kanamycin/MCH/CP-S1-S2/Au (d), kanamycin/SP/MCH/
212 CP-S1-S2/Au (e). CV and EIS measurements were performed in containing 0.25 mM
213 KCl and 5.0 mM $\text{K}_3[\text{Fe}(\text{CN})_6]$

214 *3.4 Optimization of the experimental conditions*

215 To achieve optimal performance for kanamycin detection by electrochemical
216 biosensor, the experimental conditions, including homogeneous reaction time, the
217 concentration of Exo III and the molar ratio of HP2 and CP were researched. As
218 shown in Fig. 3A, it was observed that the DPV peak intensities for the positive
219 sample enhanced continuously with increasing of the reaction time until a plateau was

reached at 2 h (curve a). In contrast, for blank sample, there was almost no visible change in current intensity (curve b). Thus, 2 h was chosen as the optimized homogeneous reaction time in the following experiments. In addition, the concentration of Exo III was essential for the current responses of the bipedal DNA machine. As depicted in the Fig. 3B, we researched the relationship between the concentration of Exo III and the $(I_1 - I_0)/I_0$ values. I_1 and I_0 represent the DPV signal in the presence and absence of 100 pM kanamycin, respectively. In Fig. 3B, the $(I_1 - I_0)/I_0$ increased with an increase in the concentration of Exo III and became saturated after 10 U; so, we selected 10 U as the optimal concentration for Exo III. In the Fig. 3C, the current peak intensities gradually increased with the augment of the molar ratios and reached a maximum intensity when the ratio of HP2 and CP was 1:20. Once beyond the ratio of 1:20, the current intensities reduced upon an increase of the ratio. For the blank sample, the current peak intensities displayed no obvious increase with increasing of the molar ratios. So, 1:20 was used as the subsequent experiments.

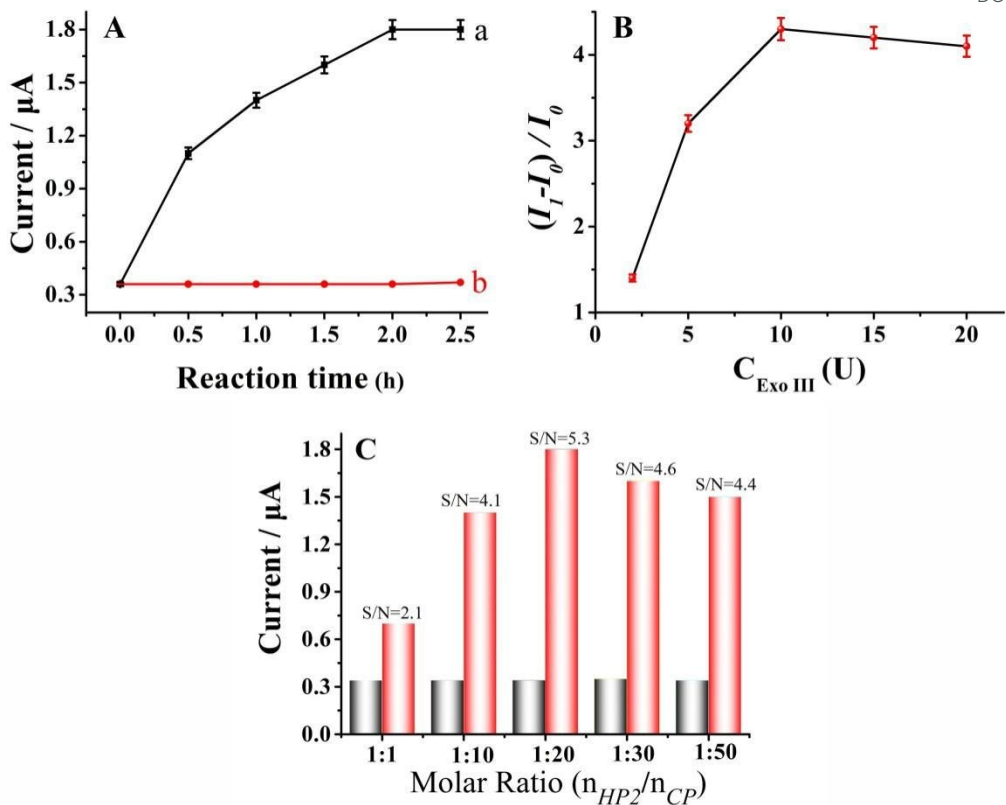


Fig.3. (A) Effect of the homogeneous reaction time on the DPV signal intensity of the biosensor in the presence (curve a) and absence (curve b) of 100 pM kanamycin; (B) Effect of the concentration Exo III to the DPV peak intensities of the bipedal DNA machine; (C) Effect of the molar ratio of HP2 and CP to current peak of the bipedal DNA machine in the presence (the red bar) and absence (the black bar) of 100 pM kanamycin. Error bars were standard deviations across three repetitive experiments.

3.5 Analytical performance of the electrochemical biosensor for detecting kanamycin

Under the above-mentioned optimized experimental conditions, the analytical performance of the electrochemical biosensor was investigated towards the detection of kanamycin at various concentrations. As seen from Fig. 4A, the DPV intensity

increased with increasing kanamycin concentration from 0 to 100 pM. The plot of the DPV peak intensities versus the logarithm value of kanamycin concentrations showed an excellent linear dependence in the range from 10 fM to 100 pM (inset in Fig. 4B), and the calibration equation was calculated to be $I (\mu\text{A}) = 0.11 + 0.33 \lg(C_{\text{kanamycin}} / \text{fM})$ with the correlation coefficient of 0.994. The limit of detection (LOD) was calculated to be 7.1 fM in terms of the rule of 3 times standard deviation of the blank divided by the slope of the calibration line. Furthermore, we compared this bipedal DNA machine with previous reported methods of quantitative assay of kanamycin and the results were shown in **Table 2**. It was noticed that the sensitivity and detection range of the biosensor preceded the majority of the existing methods for kanamycin which is due to the ETBR and bipedal DNA walker excellent signal amplification efficiency.

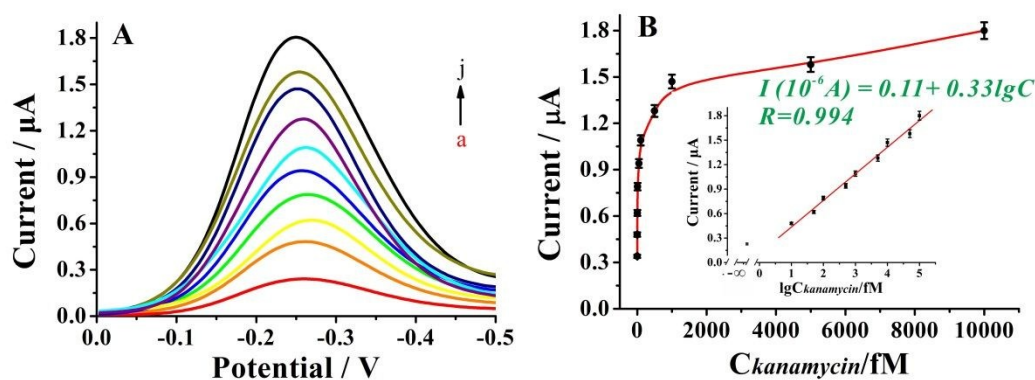


Fig.4. (A) DPV responses of our bipedal DNA walker toward kanamycin with different concentrations (from curve a-j: 0 pM, 10 fM, 50 fM, 100 fM, 500 fM, 1 pM, 5 pM, 10 pM, 50 pM, 100 pM respectively); (B) Variance of DPV intensity as a function of kanamycin with various concentrations. Inset: the calibration curve of

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262 DPV peak current versus the logarithm of kanamycin in range from 0 to 100 pM.

263 Error bars showed the standard deviations across three repetitive experiments.

264 **Table 2.** Comparison of different assay strategies for kanamycin determination.

Detection method	Detection range	Detection limit	Reference
ELISA	1.7 pM-3.4 pM	7.7 pM	34
HPLC	10 nM-150 nM	25 nM	35
SERS	1 nM-100 nM	0.75 nM	36
SPR	100 nM-10 μM	43 nM	10
Colorimetry	50 pM-500 pM	14.7 pM	13
Fluorescence	24.75 nM-137.15 nM	7.5 nM	37
Luminescence	0.2 μM-150 μM	14.3 nM	38
Electrochemistry	1 nM to 10 mM	1 nM	39
Electrochemistry	1 pM to 10 nM	0.5 pM	40
Electrochemistry	10 fM-100 pM	7.1 fM	This work

265 *3.6 Detection Specificity of the bipedal DNA machine*

266 To evaluate the specificity of our strategy for the kanamycin assay, the
267 electrochemical biosensor was challenged with target kanamycin and other non-target
268 antibiotics containing ampicillin, streptomycin, oxytetracycline, tobramycin and

terracycline. The results were shown in **Fig.5**. The DNA machine only exhibited a strong DPV response in the presence of kanamycin, while the current response to several non-target antibiotics caused extremely weak current intensities. These results distinctly proved the excellent specificity of our sensing platform for the kanamycin detection.

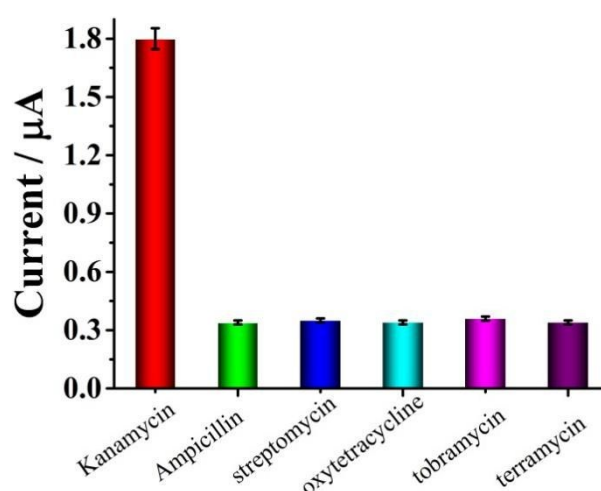


Fig.5. DPV responses acquired of the DNA machine for different antibiotics detection. The concentration of kanamycin was 100 pM, other non-target antibiotics were 1 nM. Error bars were standard deviations across three repetitive experiments.

3.7 Real sample analysis

To test the applicability of our biosensor, the quantitative assay of kanamycin in drinking water was performed. Various concentrations of kanamycin were added into the drinking water samples and measured without pretreatment except for dilution with the reaction buffer. As shown in **Table 3**, these results obtained by our strategy were approximate to spiked amount, the recovery of the proposed strategy was in the

range of 95.5% to 105.7%. These data clearly revealed the bipedal DNA machine could hold a great promise for real sample analysis with a desirable reliability and accuracy.

Table 3 Application of the DNA machine to detect kanamycin in drinking water.

Spiked amount (fM)	Measured amount (fM)	Recovery (%) $\pm$ SD, n=5
1.00×10^1	$(1.032 \pm 0.012) \times 10^1$	103.2 $\pm$ 1.2
1.00×10^2	$(1.042 \pm 0.015) \times 10^2$	104.2 $\pm$ 1.5
1.00×10^3	$(1.010 \pm 0.042) \times 10^3$	101.0 $\pm$ 4.2
1.00×10^4	$(1.013 \pm 0.041) \times 10^4$	101.3 $\pm$ 4.1
1.00×10^5	$(0.969 \pm 0.014) \times 10^5$	96.9 $\pm$ 1.4

4. Conclusion

In summary, we develop an Exonuclease III-aided target-aptamer binding recycling (ETBR) activated bipedal DNA machine for ultrasensitive electrochemical detection of antibiotic. As far as we know, this is the first time that bipedal DNA machine has been integrated into electrochemical sensing for antibiotics. Benefiting from ETBR and efficient strand displacement amplification provided by bipedal DNA walker, our electrochemical biosensor achieves improved sensitivity with a detection limit as low as 7.1 fM. Furthermore, the bipedal DNA machine surpasses conventional single swing arm DNA machine in terms of walking efficiency and

stability. Therefore, this proposed electrochemical platform indeed offers an ingenious and practical method for sensing of antibiotics residues and related food safety analysis.

Acknowledgements

This work was supported by National Natural Science Foundation of China (31471644), the Primary Research & Development Plan of Shandong Province (2017GSF220009), program for Taishan Scholar of Shandong provinces (ts201712048).

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